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# Rational development of a highly selective ratiometric fluorescent probe for hydrogen polysulfides



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#### 1. Introduction

Reactive sulfur species (RSS) are a family of sulfur-containing molecules widely existing in biological systems. Many RSS, such as thiols, hydrogen sulfide (H<sub>2</sub>S) and sulfur dioxide (SO<sub>2</sub>), play crucial roles in many physiological processes [1,2] and have attracted increasing attention in the field of biomedical research or clinical diagnosis. Very recently a new research hotspot in RSS which focuses on the chemical biology of hydrogen polysulfides (H<sub>2</sub>S<sub>n</sub>,  $n \ge 2$ ) has emerged [3–6]. It was found that  $H_2S_n$  exhibit higher potency than H<sub>2</sub>S in activating ion channels, transcription factors and tumor suppressors [3-7]. Thus,  $H_2S_n$  might be the real signal molecules in those above-mentioned physiological processes. The formation of endogenous H<sub>2</sub>S<sub>n</sub> may be from various pathways in biological systems: (1) the oxidation of endogenous H<sub>2</sub>S by reactive oxygen species (ROS) like ClO<sup>-</sup> [8]. (2) subsequent conversion from persufides (e.g. CysSSH, GSSH, GSSSH, etc.) which are produced through cystathionine- $\beta$ -synthetase (CBS) and cystathionine  $\gamma$ lyase (CES) mediated cysteine trans-sulfuration [9,10]. Though some advances about the biological effects of  $H_2S_n$  have been

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#### ABSTRACT

A novel ratiometric fluorescent probe, ACC-Cl, was developed by virtue of a unique  $H_2S_n$ -mediated coumarindithiolone formation under mild condition. The probe ACC-Cl exhibited excellent sensitivity and selectivity toward  $H_2S_n$  over other RSS such as biothiols and  $H_2S$ . More importantly, titration of  $H_2S_n$  to the solution of ACC-Cl under the existence of 1 mM GSH, the obvious ratiometric fluorescence response still can be observed. Significantly, preliminary biological experiments indicated its potential for sensing  $H_2S_n$  in living cells.

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obtained, the underlying physiological properties of  $H_2S_n$  remain largely unknown. One bottleneck might be the lack of effective methods for accurately detecting  $H_2S_n$  in biological systems.

Given its advantages of excellent sensitivity and high spatiotemporal resolution, fluorescent probes now have been recognized as the effective molecular tools that can help to detect or visualize biologically active molecules in vivo [11]. Pioneered by Xian's group, the first fluorescent probe for H2Sn was developed by taking advantage of  $H_2S_n$ -mediated benzodithiolone formation (Scheme 1) [12]. From then on, this strategy was used to construct more specific fluorescent probes for H<sub>2</sub>S<sub>n</sub> [13-24]. While these probes can selectively distinguish H<sub>2</sub>S<sub>n</sub> from other RSS, most of them respond to H<sub>2</sub>S<sub>n</sub> in a fluorescence "turn-on" manner. It's known that "turn-on" fluorescent probes usually suffer from data distortion in the quantitative measurements caused by environment changes [25,26]. By contrast, ratiometric fluorescence probes allow the measurement of the fluorescence intensity at two different wavelengths, which can solve the problems arising from environmental factors and can also increase the dynamic range of fluorescence measurement [27–30]. Unfortunately, ratiometric fluorescent probes for H<sub>2</sub>S<sub>n</sub> are rare up to now [18,19,31].

In this work, we have developed a novel ratiometric fluorescent probe for  $H_2S_n$  detection by taking advantage of a unique  $H_2S_n$ -mediated coumarindithiolone formation under mild condition. The

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#### (A) General strategy (Suitable for developing switch-on fluorescence probe)



(B) This work (Potentially suitable for developing ratiometric fluorescence probe)



Scheme 1. Proposed Strategy for Ratiometric H<sub>2</sub>S<sub>n</sub> Probes.

probe exhibited excellent sensitivity and selectivity toward  $H_2S_n$  over other RSS such as biothiols and  $H_2S$ . Preliminary biological experiments indicated its potential for sensing  $H_2S_n$  in living cells.

#### 2. Materials and methods

#### 2.1. Materials and instruments

Unless otherwise noted, all reagents were obtained from commercial suppliers and used without further purification. Solvents were purified by standard methods prior to use. Twice-distilled water was used throughout all experiments. NMR spectra were recorded on a BRUKER 300 spectrometer with TMS as an internal standard. All accurate mass spectrometric experiments were performed on a Xevo G2 QTof MS (Waters, USA). UV–vis absorption spectra were recorded on a TU-1901 (Puxi, P.R. China) spectrophotometer. Fluorescence spectra were recorded at room temperature using a HITACHI F-4600 fluorescence spectrophotometer with both the excitation and emission slit widths set at 5.0 nm. Cell imaging was performed with a Zeiss LSM 710 laser scanning confocal microscope. TLC analysis was performed on silica gel plates and column chromatography was conducted using silica gel (mesh 200–300), both of which were obtained from Qingdao Ocean Chemicals, China.

#### 2.2. General procedure for spectral measurements

A stock solution of ACC-Cl was prepared at 1 mM in  $CH_3CN$ . Solutions of various testing species were prepared from  $H_2S_n$ , glutamate (Glu), proline (Pro), serine (Ser), glutathione (GSH), cysteine (Cys), L-Cystine, DL-dithiothreitol (DTT), Na<sub>2</sub>S, Na<sub>2</sub>SO<sub>3</sub>, NaSO<sub>4</sub>, NaS<sub>2</sub>O<sub>3</sub>, NaClO,  $H_2O_2$ , CaCl<sub>2</sub>, MgCl<sub>2</sub> and ZnCl<sub>2</sub>, in twice-distilled water. Homocysteine (Hcy) was prepared in PBS buffer (10 mM, pH = 7.4). A typical test solution (10.0 mL) was prepared by placing 0.05 mL of ACC-Cl (1 mM), 2.95 mL of CH<sub>3</sub>CN, and an appropriate aliquot of each analyte stock solution into an appropriate amount of PBS buffer. The resulting solution was shaken well and kept at room temperature (25 °C) for 120 min before recording its fluorescence spectra.

#### 2.3. Cell culture and fluorescence imaging

RAW264.7 cells were seeded in a 6-well plate in Dulbecco's modified Eagle's medium (DMEM) supplemented with 10% fetal bovine serum and 1% penicillin. The cells were incubated under

an atmosphere of 5%  $CO_2$  and 95% air at 37 °C for 24 h. Cell imaging was performed with a Zeiss LSM 710 laser scanning confocal microscope. Before each experiment, cells were washed with PBS buffered solution 3 times. Excitation wavelength: 405 nm. Emissions were collected at 440–460 nm for blue channel and 520–560 nm for green channel.

#### 2.4. Cell viability assay

RAW264.7Cells were grown in CMEM medium supplemented with 10% FBS (Fetal Bovine Serum) and 1% antibiotics at 37 °C under an atmosphere of 5% CO<sub>2</sub>. Immediately before the experiment, the cells well placed in a 96 well plate, followed by addition of increasing concentrations of ACC-Cl. The final concentrations of the probe were kept from 0 to 12  $\mu$ M. The cells were then incubated at 37 °C in an atmosphere of 5% CO<sub>2</sub> and 95% air for 24 h, followed by MTT assays (n=5). Untreated assay with CMEM (n=5) was also conducted under the same conditions.

#### 2.5. Determination of the fluorescence quantum yield

Fluorescence quantum yields for 4, 7, and ACC-Cl were determined by using fluorescein ( $\Phi_f = 0.79$  in ethanol) [32] as a fluorescence standard. The quantum yield was calculated using the following formula (1):

$$\Phi_{F(x)} = \Phi_{F(s)} (A_S F_X / A_X F_S) (n_X / n_S)^2$$
(1)

Where  $\Phi_F$  is the fluorescence quantum yield, A is the absorbance at the excitation wavelength, F is the area under the corrected emission curve, and n is the refractive index of the solvents used. Subscripts <sub>S</sub> and <sub>X</sub> refer to the standard and the unknown, respectively.

#### 2.6. Synthesis of compound 1

Compound **1** was synthesized according the literature method [33]. The spectroscopic characteristics were in good agreement with those found in the literature. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  10.32 (s, 1H), 7.87 (d, *J* = 9.3 Hz, 1H), 6.74 (dd, *J* = 9.3, 2.5 Hz, 1H), 6.48 (d, *J* = 2.5 Hz, 1H), 3.51 (q, *J* = 7.2 Hz, 4H), 1.29 (d, *J* = 7.1 Hz, 6H).

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