

# Economic feasibility studies of high pressure PEM water electrolysis for distributed H<sub>2</sub> refueling stations

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## ABSTRACT

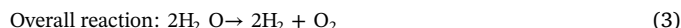
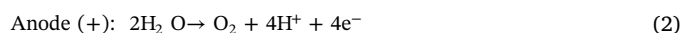
In this paper, we report economic feasibility studies focusing on profitability analysis of high pressure polymer electrolyte membrane (PEM) water electrolysis for distributed H<sub>2</sub> refueling stations in Korea. From capital and operating costs, a unit H<sub>2</sub> production cost of 6.24 \$ kgH<sub>2</sub><sup>-1</sup> was obtained for a H<sub>2</sub> capacity of 700 m<sup>3</sup> h<sup>-1</sup>, which is equivalent to handling about 300 fuel cell electric vehicles. Based on cost estimations, profitability analysis using cash flow diagrams was performed to assess the economic feasibility of high pressure PEM water electrolysis and various key economic indicators like net present value (NPV), discounted payback period (DPBP), and present value ratio were obtained for both different discount rates and capacity factors. In conclusion, employment of high pressure PEM water electrolysis was found to be economically profitable showing reasonably high NPVs (16–52 MM\$) and short DPBPs (4–7 years).

## 1. Introduction

Hydrogen (H<sub>2</sub>) widely used in various industry such as petrochemical, petroleum refining processes, fertilizer, fuel cell stacks, fuel cell electric vehicles (FCEVs), and chemicals [1–3] has taken the spotlight as an eco-friendly energy carrier that can possibly replace a current fossil fuel based energy production. H<sub>2</sub> has been produced so far through various methods such as steam methane reforming (SMR) [4–6], oil/naphtha reforming [7–9], coal gasification [10–12], and water electrolysis (WE) [13–15]. Most (about 96%) of H<sub>2</sub> production has been from fossil fuels and in particular SMR is the most widely used method for H<sub>2</sub> production accounting for about 48% in a world-wide H<sub>2</sub> production [16]. However, SMR also produces CO<sub>2</sub>, a greenhouse gas requiring the development of an eco-friendly H<sub>2</sub> production method to replace this SMR. Among many potential candidates, WE has gained much attention as a CO<sub>2</sub>-free H<sub>2</sub> production method. In addition to pure H<sub>2</sub> production, recent studies showed that power to gas (P2G) and power to liquid/fuel (P2L/P2F) technology are also feasible to convert H<sub>2</sub> produced from WE into useful products like methane, methanol, dimethyl ether (DME), etc. via surplus electricity from wind turbine or

solar cells [17–20].

With applied electricity, water is decomposed into H<sub>2</sub> and O<sub>2</sub> in WE as shown in Eqs. (1)–(3). The types of WE that are under active development are alkaline water electrolysis (AWE) [21–24], polymer electrolyte membrane (PEM) water electrolysis (PWE) [25–28], and solid oxide electrolyzer cell (SOEC) [29–32].



AWE is a water electrolysis method using an alkaline electrolyte (20–30% KOH, NaOH solution) and a separation membrane to split H<sub>2</sub> and O<sub>2</sub>. AWE has advantages of convenient construction and operation at high pressure, but it has relatively low current density and efficiency due to low operating temperature and requires high maintenance cost because of membrane corrosion from electrolyte solution [33–35]. In contrast, PWE has high current density and efficiency [36–38] and can be operated at a higher pressure than AWE enough to possibly eliminate the use of compressor to pressurize produced H<sub>2</sub>. SOEC under

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## Nomenclature

|       |   |
|-------|---|
| AWE   | Alkaline water electrolysis               |
| CEPCI | Chemical engineering plant cost index     |
| CFD   | Cash flow diagram                         |
| CRF   | Capital recovery factor                   |
| DDB   | Double declining balance                  |
| DME   | Dimethyl ether                            |
| DPBP  | Discounted payback period                 |
| FCEV  | Fuel cell electric vehicle                |
| FCI   | Fixed capital investment                  |
| MACRS | Modified accelerated cost recovery system |
| NPV   | Net present value                         |
| P2F   | Power to fuel                             |
| P2G   | Power to gas                              |

|        |   |
|--------|---|
| P2L    | Power to liquid                                 |
| PA     | Profitability analysis                          |
| PEM    | Polymer electrolyte membrane                    |
| PVR    | Present value ratio                             |
| PWE    | Polymer electrolyte membrane water electrolysis |
| SA     | Sensitivity analysis                            |
| SF     | Stream factor                                   |
| SL     | Straight line                                   |
| SMR    | Steam methane reforming                         |
| SOEC   | Solid oxide electrolyzer cell                   |
| TEA    | Techno-economic analysis                        |
| US DOE | United States Department of Energy              |
| WE     | Water electrolysis                              |
| WCI    | Working capital investment                      |

development can be operated at a high temperature due to the usage of thermally stable solid oxide and has high conversion efficiency and low electricity consumption [39–41].

Fig. 1 shows a schematic of a H<sub>2</sub> refueling station for FCEVs using high pressure PWE currently being developed in Korea. Compared to AWE, a compressor is not required for high pressure PWE thus leading to possible cost savings in capital and operating costs. The H<sub>2</sub> production capacity considered in this study is 700 m<sup>3</sup> h<sup>−1</sup> from PWE under capacity factors from 20 to 80% load that can handle about 300 FCEVs a day (estimated from 14 FCEVs a day with 30 m<sup>3</sup> h<sup>−1</sup>).

Techno-economic analysis (TEA) to evaluate both technical and economic analysis is essential to determine the feasibility of a process of interest [42,43] and various research groups reported TEA studies regarding PWE. Tremel et al. [25] performed TEA for the synthesis of methanol, diesel, DME, synthetic natural gas, and ammonia from hydrogen production through WE and concluded that methanol production is the best option based on its feasibility to implement. Kopp et al. [44] carried out technical and economic analysis of 6 MW PEM electrolysis and reported the calculated efficiency of the P2G plant based on three options, which are electricity purchased at the European power exchange, surplus electricity from a marketing company, and participating in the control reserve market. Ferrero et al. [45] performed techno-economic assessment of P2G using H<sub>2</sub> obtained from variable renewable electricity storage through alkaline, PEM, and SOEC WE and estimated H<sub>2</sub> costs in 2030 scenario of 2.0–2.3 € kg<sup>−1</sup> for mobility and 1.0–1.2 € kg<sup>−1</sup> for grid injection. In our previous studies [3,46], comprehensive economic analysis through a unit H<sub>2</sub> production cost, sensitivity analysis (SA), profitability analysis (PA) employing a discounted cumulative cash flow diagrams (CFDs), and uncertainty analysis using a Monte-Carlo simulation method targeting various H<sub>2</sub> production capacities for H<sub>2</sub> refueling stations in Korea was conducted. Unit H<sub>2</sub> production costs of 17.99, 16.54, and 20.18 \$ kgH<sub>2</sub><sup>−1</sup> were estimated for AWE, PWE, and SMR for a H<sub>2</sub> refueling station with a H<sub>2</sub> production capacity of 30 m<sup>3</sup> h<sup>−1</sup>. The main difference of a unit H<sub>2</sub> production cost from the work by Ferrero et al. [45] can be ascribed to a lower H<sub>2</sub>

production capacity. With these unit H<sub>2</sub> production costs, discounted cumulative CFDs were constructed to provide discounted payback periods (DPBPs, a period required to recover all fixed capital investment (FCI)) for different discount rates of 2, 6, 8, 10, and 14%. Furthermore, uncertainty analysis using a Monte-Carlo simulation method showed significantly changeable unit H<sub>2</sub> production costs and provided useful economic guidelines to properly take economic fluctuations into account.

Extending our previous TEA studies coupled with Korean government's recent plan to construct about 200 H<sub>2</sub> refueling stations by 2025, here we report the economic feasibility of high pressure PWE for H<sub>2</sub> refueling stations in Korea with a H<sub>2</sub> production capacity of 700 m<sup>3</sup> h<sup>−1</sup>, which is considered as distributed H<sub>2</sub> production by US DOE [47]. In particular, a unit H<sub>2</sub> production cost based on capital and operating costs, SA to determine the most influential economic factors, and discounted cumulative CFDs to obtain DPBP, net present value (NPV, net profit at the end of the process), and present value ratio (PVR, ratio of negative cash flows and positive cash flows) were analyzed to properly evaluate techno-economic feasibility of high pressure PWE under active development in Korea.

## 2. Method

### 2.1. Unit H<sub>2</sub> production cost

Economic analysis methods employed in this paper were based on the work of Turton et al. [48] and a unit H<sub>2</sub> production cost for 700 m<sup>3</sup> h<sup>−1</sup> was estimated from our previous data for 30, 100, and 300 m<sup>3</sup> h<sup>−1</sup> [3]. Total costs are the sum of capital costs and operating costs. Capital costs include H<sub>2</sub> production equipment, storage, compressor, pump, dispenser, construction, and supplement while operating costs consist of electricity, labor, maintenance, other operating cost, water, land rent, and natural gas. All economic data were adjusted to 2015 by using chemical engineering plant cost index (CEPCI) of 547.2 as of October 2015. To convert capital costs into annualized ones,

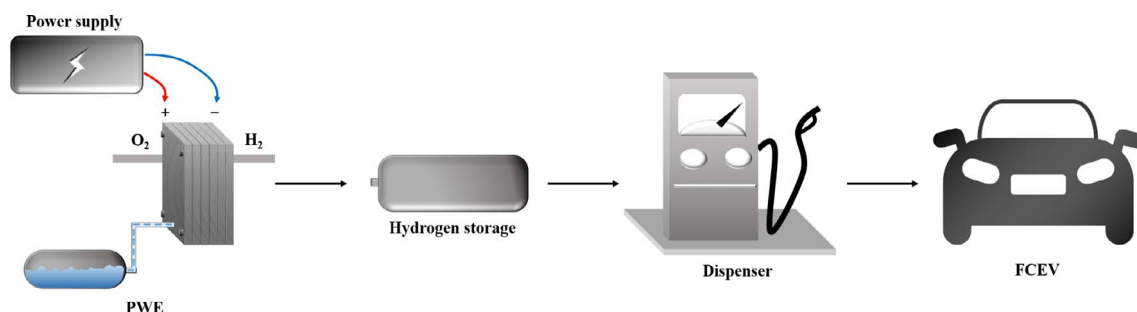


Fig. 1. Schematic of a H<sub>2</sub> fueling station using high pressure PEM water electrolysis (PWE).

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