



# Investigation of power battery thermal management by using mini-channel cold plate



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## ABSTRACT

In order to guarantee the safety and extend the cycle life of Li-ion power batteries within electric vehicles, a mini-channel cold plate-based battery thermal management system is designed to cool a rectangular Li-ion battery. A three-dimensional thermal model of the cooling system was established and the effects of number of channels, flow direction, inlet mass flow rate and ambient temperature on temperature rise and distribution of the battery during the discharge process were investigated. The results suggest that the maximum temperature of the battery decreases with increases in the number of channels and inlet mass flow rate. The effect of flow direction on cooling performance was smaller after mass flow rate increased. The cooling performance improved with the increase of inlet mass flow rate but the increasing trend became smaller, and the mass flow rate as  $5 \times 10^{-4} \text{ kg s}^{-1}$  was optimal. The simulation results will be useful for the design of mini-channel cold plate-based battery thermal management system.

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## 1. Introduction

The power density of a conventional Li-ion power battery is 4–5 times higher than a typical lead-acid battery. As a result, Li-ion power batteries are currently receiving significant attention from the scientific community [1,2]. However, Li-ion batteries generate a large amount of heat due to electrochemical reactions during discharging process [3,4], which may overheat the battery or lead to non-uniform temperature distribution. As a result, the safety of the battery can be compromised and its cycle life may be reduced [5,6]. Hence, battery thermal management (BTM) is necessary to control the battery temperature within an acceptable range and maintain a uniform temperature distribution during operation [7–9]. Generally, the thermal management techniques that are employed for BTM include: forced convection with air and liquids, as well as the use of solid–liquid phase change materials (PCM) [10–14].

Because of the low thermal conductivity of air [15], extremely high velocities of air are required to sufficiently cool Li-ion batteries using active cooling techniques [10,11]. Compared with air, higher rates of cooling can be achieved in BTM systems that employ liquids due to their high thermal conductivities [12]. PCM are also excellent candidates for BTM due to the large amount of heat they can absorb during a solid–liquid phase transition

[13,14]. Karimi and Li [16] obtained the temperature distribution within a battery module when air, silicon oil and PCM are used as heat transfer mediums that are embedded within cooling channels and located on the side of the battery module. Using numerical simulations, it was shown that the cooling performance of silicon oil was between air and PCM, but the characteristic of low thermal conductivity before phase change impedes heat spreading and thus limits the application of PCM for BTM [17,18]. BTM system based on liquid cooling can be subdivided into conventional liquid cooling system, cooling system with cold plates and cooling system with heat pipes. In conventional liquid cooling system, heat is taken away by channel arranged between battery module or by a jacket employed around the battery module. In our previous work [19], we established a two-dimensional single-phase convective heat transfer model, and obtained the maximum temperature of the battery using air and water as cooling medium, respectively. The results showed that when the cooling medium was changed to water from air, the battery maximum temperature fell from 55.82 °C to 49.59 °C.

Based on the large amount of liquid latent heat of vaporization, the use of heat pipes for BTM allows for a significant reduction in battery operating temperature [9,20–22]. Greco et al. [23] proposed a simplified thermal network based on the heat transfer principle associated with the use of heat pipes, with which a one-dimensional computational model was developed. The quantitative results of the one-dimensional computational model were compared with an analytical solution and a three-dimensional

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