



Mushrooms dehydration in a hybrid-solar dryer, using a phase change material



Alejandro Reyes*, Andrea Mahn, Francisco Vásquez

Department of Chemical Engineering, University of Santiago of Chile, Av. L.B. O'Higgins 3363, Santiago, Chile

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ABSTRACT

Mushrooms were dehydrated in a hybrid solar dryer provided with a solar panel of a total exposed surface of 10 m², electric resistances and paraffin wax as a phase change material. Mushrooms were cut in 8 mm or 12 mm slices. At the outlet of the drying chamber the air was recycled (70% or 80%) and the air temperature was adjusted to 60 °C. At the outlet of the solar panel the air temperature rose up to 30 °C above the ambient temperature, depending on solar radiation level.

The effective diffusivity, estimated by the Simplified Constant Diffusivity Model, considering or not shrinkage, fluctuated between $2.5 \cdot 10^{-10}$ m²/s and $8.4 \cdot 10^{-10}$ m²/s with R^2 higher than 0.99, agreeing with values reported in literature. The empirical Page's model resulted in a better adjustment, with R^2 above 0.998.

In all runs the dehydrated mushrooms showed a notorious darkening and shrinkage. Rehydration assays at 30 °C showed that in less than 30 min rehydrated mushrooms reached a moisture content of 1.91 (dry basis). Rehydrated mushrooms had a higher hardness compared with fresh mushrooms. The Simplified Constant Diffusivity Model and the Peleg's model adjusted to the rehydration data with RMSE values below 0.080.

Thermal efficiency fluctuated between 22% and 62%, while the efficiency of the accumulator panel varied between 10% and 21%. The accumulator allowed reducing the electric energy input.

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1. Introduction

The energy necessary for drying usually comes from fossil fuels, whose prices are continuously rising and their negative environmental impact due to CO₂ emissions is increasingly challenged. The use of renewable non conventional energy has allowed reducing the use of fossil fuels [1–4]. Additionally, the solar energy is a promising energy source, despite its daily and seasonal fluctuations that represent a severe drawback. This makes it necessary to use additional energy sources (biomass, hydrocarbons or electric energy) and/or solar energy accumulators with phase change materials (PCM), that allow the operation of solar dryers during the low or null irradiation periods [7,8].

PCM are materials that accumulate energy when changing from solid to liquid state (melting heat), and turns it over when changing from liquid to solid state (solidification heat). Phase change heat exhibits a high heat density and a minimum temperature variation during fusion and solidification periods. PCM are classified in organic and inorganic. Organic PCM have the advantage of

keeping their properties independently of how many times they melt or solidify [9,10].

One of the most popular organic PCM is paraffin wax, as it is chemically stable, does not degrade after repeated melt/solidify cycles and has a thermal energy storage capacities about 200 J/g. PCM can be produced in various chemical formulations and therefore with different melting temperatures. Some drawbacks of paraffin wax are its relatively low thermal conductivity and a considerable volume increase during melting.

The drying process shows two stages: (i) convective heat transfer towards the surface of the wet substrate followed by the conductive transfer of energy inside the substrate; and (ii) mass transfer from the core of the substrate towards the surface, followed by the withdrawal of water from the surface. The drying air absorbs the water only if its relative humidity is below saturation. Temperature control is relevant especially for thermo-labile foodstuffs, since the quality attributes can be impaired [5,6].

Solar drying can be classified in (a) direct and (b) indirect. Although the first one may be efficient and cheaper, it has disadvantages such as contamination with dirt, insects and microorganisms. In order to avoid these drawbacks, the foodstuff is located in a drying chamber, and receives energy indirectly from the drying

* Corresponding author. Tel.: +56 02 27181819.

E-mail address: alejandro.reyes@usach.cl (A. Reyes).

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