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# Gradual-cooling solidification approach to alleviate macrosegregation in large steel ingots



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#### ABSTRACT

Based on the mechanisms of macrosegregation evolution in large steel ingots, an approach named as gradualcooling solidification (GCS) was proposed to alleviate the macrosegregation of large steel ingots. A three-phase mixed columnar dendritic-equiaxed solidification model was employed to investigate the approach. The solidification model considered the dendritic structure of equiaxed grains, nucleation and growth of equiaxed crystals, growth of columnar trunks, thermal-solutal buoyancy, sedimentation of equiaxed crystals, and columnar-to-equiaxed transition (CET). After the verification of model accuracy by a reported experiment results of a 55-ton steel ingot, this model has been used to study the GCS approach. The simulation results showed that the GCS approach has significant potential to alleviate the macrosegregation of the large steel ingot; e.g., for a 55ton steel ingot, the variation range of segregation value decreased from 0.854 in the conventional casting case to 0.077 in the GCS case.

### 1. Introduction

Macrosegregation is the macroscopic heterogeneity of a solute species that occurs in most large scale ingot castings, this heterogeneity becoming seriously with the increasing of ingot size, as reported recently by Ge et al (2018). The formation mechanism has been extensively studied for several decades. Hultgren (1973) suggested that macrosegregation may occur when relative motion (flow) appears between the solid phase and the surrounding liquid. Relative motion has been identified as having the following major phenomena: thermal buoyancy, solutal buoyancy, sedimentation or floatation of free moving grains (Li et al., 2014a) or inclusion (Li et al., 2014c), solidification contraction-induced fluid flow or deformation, stirring, and others. Because of indirect observation of macrosegregation during it forming, modeling is a general method for macrosegregation investigation. The model considered these relative motion incorporating with mass, momentum, species, and heat transfer.

Several models were developed to simulate the macrosegregation formation since the first attempt by Fujii et al. (1979) on mushy zone in 1970s. After that, Wang and Beckermann (1996) systematically studied the effects of the grain sedimentation and thermal-solutal buoyancy on the final macrosegregation in the 1990s. In the 2000s, Combeau et al. (2009) presented a study of the morphology and motion of equiaxed grains on the final macrosegregation but their model omitted the columnar phase. Later, Wu and Ludwig (2006, 2007) provided a mixed three-phase solidification model with consideration of the interaction among the bulk liquid, the equiaxed grains, and the columnar phases. Recently, Ge et al. (2017) presented a four-phase dendritic model to study the interaction between macrosegregation and shrinkage defects during solidification.

However, most above studies focused on the improvement of the simulation accuracy, and only a few studies have attempted to find an effective way to alleviate macrosegregation. Li et al. (2014b) made a research of the multiple pouring process on the control of macro-segregation. Sang et al. (2010) added solid steel balls into melt during the pouring process to reduce the degree of macrosegregation. Recently, Ren et al. (2017) proposed a so called layer casting method which proved to be a promising method to increase the homogeneity of ingots, however, a lot of efforts should be implemented to modify its shortages, such as inclusion problem.

In the current study, a gradual-cooling solidification (GCS) aiming at alleviating macrosegregation was proposed, which was verified using a dendritic equiaxed-columnar mixed solidification model based on our previous researches, i.e., two-phase dendritic model from Ge et al.

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Nomenclature		$\Gamma_{\rm env}$	Envelope transfer rate (kg m <sup><math>-3</math></sup> s <sup><math>-1</math></sup> )
		$\rho_l, \rho_s, \rho_c$	Density (kg m $^{-3}$ )
Symbol a	lescription (unit)	$c_l, c_s, c_c$	Species concentration (wt.%)
		$c_{mix}$	Mix concentration (wt.%)
<i>c</i> <sub>0</sub>	Initial concentration (wt.%)	$c_p$	Specific heat $(J kg^{-1} K^{-1})$
c <sub>ref</sub>	Reference concentration (wt.%)	$d_s, d_{env}$	Diameter of solid and envelop (m)
$C_{ls}, C_{lc}$	Species exchange (kg m $^{-3}$ s $^{-1}$ )	G	Temperature gradient (K m <sup>-1</sup> )
$D_l$	Diffusion coefficient $(m^2 s^{-1})$	H	Heat transfer coefficient (W $m^{-2}K^{-1}$ )
$f_l, f_s, f_{env},$	$f_c$ Volume fraction (1)	$h_b h_s, h_c$	Enthalpy $(J kg^{-1})$
$\overrightarrow{g_l}, \overrightarrow{g_s}$	Reduced gravity (m s <sup><math>-2</math></sup> )	k	Solute partitioning coeff. (1)
$H^*$	Volume heat transfer coeff. ( $W m^{-3} K^{-1}$ )	$M_{ls}, M_{lc}$	Net mass transfer rate $(kg m^{-3} s^{-1})$
$\Delta h_f$	Latent heat (J kg <sup>-1</sup> )	$N_e$	Grain production rate by nucleation $(m^{-3}s^{-1})$
$k_l, k_s, k_c$	Thermal conductivity (W $m^{-1} K^{-1}$ )	Р	Pressure (N m $^{-2}$ )
т	Slope of the liquidius in phase diagram (K)	$S_s$	Surface area concentration of solid phase $(m^{-1})$
n	Grain number density $(m^{-3})$	Т	Temperature (K)
$Q_{ls}, Q_{lc}, Q_{lc}$	$Q_{cs}$ Energy transfer (J m <sup>-3</sup> s <sup>-1</sup> )	Т	Time (s)
Senv	Surface area concentration of envelope $(m^{-1})$	$U_{ls}, U_{lc}, U_{lc}$	$J_{cs}$ Momentum exchange rate (kg m <sup>-2</sup> s <sup>-2</sup> )
Ť	Cooling rate (K $s^{-1}$ )	$v_{Rs}$	Solid phase growth speed (m $s^{-1}$ )
$\overrightarrow{u_l}, \overrightarrow{u_s}$	Velocity (m s <sup><math>-1</math></sup> )	β	Solidification shrinkage (1)
$v_{Rc}$	Columnar growth speed in radius direction $(m s^{-1})$	$\lambda_1$	Columnar grain space (m)
$v_{tip}$	Dendrite tip velocity (m s <sup><math>-1</math></sup> )	$\mu_l$	Viscosity (kg m <sup><math>-1</math></sup> s <sup><math>-1</math></sup> )
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(2016) and mixed three-phase model from Li et al. (2014a). A schematic description of current process is presented in Fig. 1 including a conventional solidification process (CSP) and a gradual-cooling solidification (GCS) process. As shown in Fig. 1, the thermal insulation material (insulation bricks) averagely covered the iron mold after filling with the melt, and all the thermal insulation was averaged and divided into 16 parts along the height of the mold. The solidification started from the bottom part of the ingot when the first part of the thermal insulation material was removed. After a while, the second part of thermal insulation was removed, and the ingot continued solidifying due to the heat that was transferred out of the ingot. The thermal insulation material was removed piece by piece, and the melt gradually solidified from the bottom to the top. Therefore, the process was referred as the gradual-cooling solidification (GCS) process. The main difference between CSP and GCS was the location of the heat transfer. In the GCS case, heat transfer only occurred around the mold where the insulating material had been removed (Fig. 1(c)), while heat transfer occurred around the entire mold in the CSP case (Fig. 1(b)). In this paper, we just presented this process to exhibit its potential of macro-segregation alleviation based on simulation results.

## 2. Mathematic model

A dendritic equiaxed-columnar mixed solidification model was adopted in the current simulation to study the feasibility of GCS. Three individual phases were taken into account: liquid, dendritic equiaxed grains, and columnar grains. The model integrated the macroscopic phenomena of the mass, momentum, heat, and species transfer with the



Fig. 1. Schematic description of (a) filling the ingot, (b) the conventional solidification process, (c) the gradual-cooling solidification process.

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