Author's Accepted Manuscript

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PII:S0956-5663(17)30780-7DOI:https://doi.org/10.1016/j.bios.2017.11.051Reference:BIOS10127

To appear in: Biosensors and Bioelectronic

Received date: 19 September 2017Revised date: 9 November 2017Accepted date: 16 November 2017

Cite this article as: Dongyang Wang, Bintong Huang, Jie Liu, Xia Guo, Guzailinur Abudukeyoumu, Yang Zhang, Bang-Ce Ye and Yingchun Li, A novel electrochemical sensor based on Cu@Ni/MWCNTs nanocomposite for simultaneous determination of guanine and adenine, *Biosensors and Bioelectronic*, https://doi.org/10.1016/j.bios.2017.11.051

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A novel electrochemical sensor based on Cu@Ni/MWCNTs nanocomposite for simultaneous determination of guanine and adenine

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Abstract: A novel electrochemical sensing platform based on combination of multi-walled carbon nanotubes and copper-nickel hybrid nanoparticles (Cu@Ni/MWCNTs) was developed for simultaneous detection of guanine (G) and adenine (A). The Ni/MWCNTs and Cu@Ni/MWCNTs nanocomposites were characterized by transmission electron microscopy (TEM) and energy-dispersive X-ray spectroscopy (EDS). The electrochemical behaviors of G and A on the modified electrode were explored by cyclic voltammetry (CV) and differential pulse voltammetry (DPV) in phosphate buffer with pH 3.0. Under the optimal conditions, electrical signals were linear over the concentration ranges from 5.0 to 180 μ M and 8.0 to 150 μ M for simultaneous determination G and A with the detection limit as low as 0.35 μ M and 0.56 μ M (S/N=3), respectively. Furthermore, linear concentration ranges in individual determination are 1.0~180 μ M and 2.0~150 μ M with detection limits of 0.17 μ M and 0.33 μ M (S/N=3) for G and A, respectively. The sensor was successfully used to quantify G and A in real samples. The Cu@Ni/MWCNTs composite presented here can serve as a promising candidate for developing electrochemical sensor devices and plays an important role in widespread fields.

Keywords: *Multi-walled carbon nanotubes; Copper nickel nanoparticles; Guanine; Adenine; Electrocatalytic oxidation; Synergistic effect*

1. Introduction

Guanine (G) and adenine (A) are important purine bases in deoxyribonucleic acid (DNA) and ribonucleic acid (RNA). Purine bases are crucial biomolecules which play various roles in cell proliferation (Burnstock 2002), coronary blood flow regulation (Burnstock 1997), cardiovascular system (Ralevic et al. 1991), etc. Abnormal changes in purine content and metabolism may affect immune function and cause various diseases including cancer, epilepsy (Cekan and Sigurdsson 2009; Pietrzyk et al. 2010), lupus erythematous, renal calculi and anemia (Fox 1981; Nyhan 2005). Hence, accurately monitoring G and A in biosamples is of great meaning for assessment of health status and early warning of several diseases. Analytical approaches that have been reported for detecting G and A include liquid chromatography (Gill and Indyk 2007), calorimetry (Heisler et al. 2002), capillary zone electrophoresis (Yeh and Jiang 2002) and mass spectrometry (Huang and Chang 2007). However, most of them suffer from disadvantages such as consumption of a large amount of organic reagents, expensive equipment and time-consuming procedures. In comparison, electrochemical sensor for the determination (Hui et al. 2015; Lynge et al. 2011; Sun et al. 2008; Yari and Derki 2016; Yari and Saidikhah 2016) of G and A can be very attractive since it allows developing a sensitive, fast, simple and cost-effective assay, especially suitable for complex and low concentration biological samples without complicated pre-treatment.

To developing a sensor, sensitivity and selectivity are always key issues for practical application (Yin et al. 2015). To improve the sensitivity, various modifiers like carbon nanotubes (CNTs), graphene, precious and transition metals, semiconductors and artificial polymers (Dan et al. 2015; Jiang and Zhang 2010; Wang et al. 2006; Wang et al. 2009) have been utilized to decorate electrodes. Recently, studies of bimetallic nanoparticles as catalysts have

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