



Study of ozone generation in an atmospheric dielectric barrier discharge reactor



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ABSTRACT

Ozone (O₃) generation in a dielectric barrier discharge (DBD) reactor driven by a pulsed power supply was investigated at atmospheric pressure and room temperature. An O₃ generation efficiency model is established in which discharge power, O₂ concentration, gas flow rate, and volume of the discharge space are included. Constants in the O₃ generation efficiency model were obtained by fitting the model with experiment results. O₃ concentration can be simply calculated from the energy density and initial O₂ concentration. Comparison on O₃ concentrations from calculation with references is given.

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Introduction

Ozone (O₃) is a useful chemical and widely used in many fields, such as advanced oxidation processes (AOPs), chemical–biological processes (CBP), and semiconductor industry [1,2]. O₃ also is a powerful chemical in food and medical treatments [3–7]. Generally, O₃ is generated by applying high-voltage to a dielectric barrier discharge (DBD) reactor of a discharge space in which an oxygen (O₂)-containing gas is present. It has been understood that a number of tiny breakdown channels occur in the discharge space; those channels are suggested as microdischarges having a time order of microseconds, where O₃ is generated [8–18]. O₃ generation reactions in microdischarges begin with the dissociation of O₂ molecules to oxygen atoms (O) by the impact of O₂ with energized electrons in an electric field. O atoms then combine with O₂ to yield O₃. The energy efficiency of O₃ generation is strongly related with the production efficiency of O atoms in the microdischarges.

The energy efficiency of O₃ generation (ξ) using an AC power supply can be obtained from the approximation given by Eliasson and Kogelschatz [8], as defined by

$$\xi = \frac{2\rho_D}{ev_d E/n}. \quad (1)$$

Wei et al. [19] developed a numerical model which describes the influence of both electrical and discharge configuration parameters on ozone concentration in pulsed positive dielectric barrier discharge. Factors of pulse repetition frequency, difference of peak pulsed voltage and corona inception voltage, gap length, relative permittivity, gas pressure, gas flow rate, and pulse duration were taken into account. O₃ concentration is given in a form in which 9 parameters are required.

Related with the dissociation of O₂ to O by the impact of O₂ with electrons as shown in Eq. (2), the constant k_1 of the dissociation process is sensitive to the amplitude of the electric field. The value of k_1 in 1/(cm³ s) can be calculated using Eq. (3), when a microwave power supply is applied [20].



$$k_1 = 2 \times 10^{-\left(7.8 + \frac{14.7}{\sigma}\right)} + 10^{-\left(7.4 + \frac{17.1}{\sigma}\right)}, \quad (3)$$

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Nomenclature	
Symbols	
d	alumina spacer thickness in mm
E	energy injection over one pulse discharge duration in J/Hz
E_0	electric field amplitude in V/cm
ED	energy density in J/m ³
f	pulse frequency in Hz
F	total gas flow rate at the inlet of the DBD reactor in m ³ /s
ΔF	difference on total gas flow rates between the inlet and outlet of the DBD reactor in m ³ /s
E/n	reduced electric field in Td
I_i	discharge current in A at discharge time t_i
I_{i+1}	discharge current in A at discharge time t_{i+1}
k	ozone generation rate constant in mol ^{0.5} m ^{1.5} /J
k_1	kinetic constant rate of O ₂ dissociation in 1/(cm ³ s)
N_m	concentration of neutrals in 1/cm ³
P	energy injection density in W/m ³
P_{in}	energy injection power in W
P_{INC}	inception energy injection power for O ₃ generation in W
r_{O_2}	O ₂ consumption rate in mol/J
r_{O_3}	O ₃ generation efficiency in mol/J
T_e	temperature of electrons in eV
t_i	discharge time in s
t_{i+1}	discharge time in s
V_i	discharge voltage in V at discharge time t_i
V_{i+1}	discharge voltage in V at discharge time t_{i+1}
V_R	total discharge space volume in m ³
x	O ₂ conversion in percentage
α	constant
β	constant
η	O ₃ generation efficiency in g/kWh
θ	energy related factor in eV
v_d	electron drift velocity in cm/s
ν_e	collision frequency of electrons with neutrals in Hz
ξ	number of oxygen atoms produced per eV in 1/eV
ρ_D	total O ₂ dissociation rate coefficient in cm ³ /s
ω	frequency of the microwave field in Hz
[O ₂] ₀	initial O ₂ concentration in inlet gases of the DBD reactor in mol/m ³
[O ₂]	O ₂ concentration in mol/m ³
[O ₃]	O ₃ concentration in g/m ³
[O ₃ ']	O ₃ concentration in mol/m ³
*	excited state
Abbreviations	
CT	current transformer
DBD	dielectric barrier discharge
e	electron
HV	high-voltage
OSC	oscilloscope

$$\theta = \frac{E_0 \cdot \nu_e}{(\omega^2 + \nu_e^2)^{\frac{1}{2}} N_m} \times 10^{16}. \quad (4)$$

k_1 was also suggested as a constant of $2 \times 10^{-9} 1/(\text{cm}^3 \text{ s})$ [21]. k_1 can be obtained by Lee et al. using Eq. (5) in a DBD reactor [22]. k_1 is given in a more complicated form in which parameters, such as the energy branch to electron, excitation rate, impact power, discharge area, channel height of gas flow path, and average electron density, are required [23]; such parameters are difficult to be obtained. However, the influence of energy on ozone generation is additionally required if k_1 is given in a constant form.

$$k_1 = 4.2 \times 10^{-9} \exp\left(-\frac{5.6}{T_e}\right). \quad (5)$$

Recently, ozone generation using DBD reactors is still an active study [24–28]. The aim of this work is to find an O₃ generation efficiency model in which factors or parameters are simply obtained from experiments. At first, the O₃ generation in a DBD reactor driven by a pulse power supply was experimentally investigated. Secondly, an O₃ generation efficiency model was established using factors of discharge power (energy injection), O₂ concentration, gas flow rate, and volume of the discharge space those are easily given or measured. Finally, constants in the O₃ generation efficiency model were obtained after fitting the model with experiment results.

Experimental setup

Fig. 1 shows the experimental setup for the investigation of O₃ generation in a DBD reactor. A pulsed high-voltage (HV) from a power supply (DP-12K5-SCR, PECC, Japan) was applied to the DBD

reactor. The discharge voltage and current waveforms were measured using a voltage probe (V-P, P6015A, bandwidth DC–75 MHz, Tektronix, USA) and a current transformer (CT, TCP0030, bandwidth DC–120 MHz, Tektronix, USA), respectively. The signals from the voltage probe and current transformer were digitized and recorded using a digital phosphor oscilloscope (OSC, DPO 3034, bandwidth 300 MHz, Tektronix, USA).

The DBD reactor consists of two alumina plates (purity 96%, $50 \times 50 \times 1 \text{ mm}^3$, Kyocera, Japan) and two metal electrodes (aluminum tapes, $31 \times 31 \text{ mm}^2$), those plates and electrodes were sandwiched closely. The distance between two alumina plates was adjusted using two alumina spacers of different thicknesses ($d = 0.29, 0.85, 1.48, 2.03, \text{ or } 2.36 \text{ mm}$). Microdischarges occur in a discharge space between two alumina plates when high-voltage is applied to two metal electrodes. A gas mixture of nitrogen (N₂, purity 99.9999%) and O₂ (purity 99.9999%) was supplied to the inlet of the DBD reactor using two mass flow controllers (MFC) at a fixed

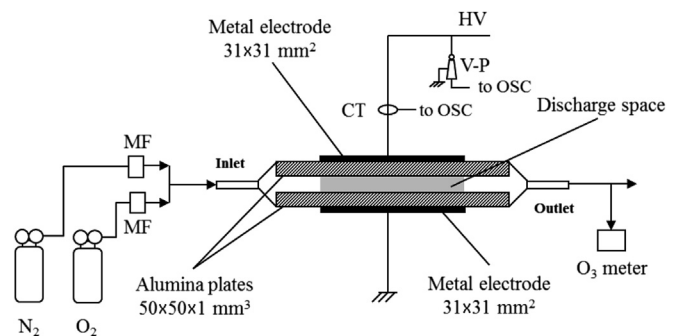


Fig. 1. Schematic diagram of O₃ generation in a DBD reactor.

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