



Review

Regeneration of carbonaceous adsorbents. Part II: Chemical, Microbiological and Vacuum Regeneration



Francisco Salvador*, Nicolas Martin-Sanchez, Ruth Sanchez-Hernandez, M. Jesus Sanchez-Montero, Carmen Izquierdo

Dpto Química Física, Facultad de Química, Universidad de Salamanca, Plaza de la Merced s/n, 37008 Salamanca, Spain

ARTICLE INFO

Article history:

Received 21 March 2014

Received in revised form 17 June 2014

Accepted 8 August 2014

Available online 18 August 2014

Keywords:

Review regeneration of activated carbon

Regeneration mechanism

Regenerating agents

Classification of regeneration methods

ABSTRACT

Due to their excellent adsorbent properties, porous carbonaceous materials are employed in separation applications, especially in the purification of polluted liquid and gaseous streams. The progressive accumulation of pollutants on the surface of these materials reduces their adsorption capacity until they are finally exhausted. The regeneration of these carbonaceous adsorbents is an economically and environmentally attractive option. Numerous regeneration methods have been developed, but until now, they had not been properly classified or deeply reviewed. In this work, all existing methods are described in-depth, and a double-criterion is proposed to allow clear and rigorous classification of current and future regeneration methods. This classification is based on the traditional division which distinguishes among Thermal, Chemical and Microbiological Regeneration, and a fourth major group, Vacuum Regeneration, is added. Thermal Regeneration has already been considered in a previous review. This second review considers both Chemical Regeneration (within which Regeneration with Liquid Water and NaOH, Electrochemical, Solvent, Supercritical and Oxidative Regeneration are identified) and Microbiological and Vacuum Regeneration.

© 2014 Elsevier Inc. All rights reserved.

Contents

1. Introduction	278
2. Important definitions	278
3. Chemical Regeneration	279
3.1. Regeneration with Liquid Water	279
3.2. NaOH regeneration	280
3.2.1. Regeneration pathways	280
3.2.2. Influence of experimental variables on NaOH regeneration	281
3.2.3. RE of NaOH regeneration	281
3.3. Solvent regeneration	281
3.3.1. Regeneration pathways	281
3.3.2. Influence of experimental variables on solvent regeneration	282
3.3.3. RE of Solvent Regeneration	282
3.4. Supercritical Regeneration	283
3.4.1. Influence of experimental variables on Supercritical Regeneration	283
3.4.2. Modeling	284
3.4.3. EE and RE of Supercritical Regeneration	284
3.5. Electrochemical Regeneration	284
3.5.1. Regeneration pathways	285

* Corresponding author. Tel.: +34 923 294478.

E-mail addresses: salvador@usal.es (F. Salvador), nicolas_martin@usal.es (N. Martin-Sanchez), ruthsh@usal.es (R. Sanchez-Hernandez), chusan@usal.es (M.J. Sanchez-Montero), misiego@usal.es (C. Izquierdo).

3.5.2.	Influence of experimental variables on Electrochemical Regeneration	285
3.5.2.1.	Variables related to the electrochemical cell	285
3.5.2.2.	Variables related to the electrical current	286
3.5.3.	RE of Electrochemical Regeneration	286
3.6.	Oxidative Regeneration	287
3.6.1.	Oxidative Regeneration with O ₂	287
3.6.1.1.	Thermal Oxidative Regeneration	287
3.6.1.2.	Catalytic Thermal Oxidative Regeneration	288
3.6.1.3.	Ozone Regeneration	289
3.6.1.4.	Wet Air Regeneration (WAR)	289
3.6.2.	Oxidative Regeneration without O ₂	289
3.6.2.1.	Photocatalytic Regeneration	289
3.6.2.2.	H ₂ O ₂ Regeneration	290
3.6.2.3.	Persulfate Regeneration	290
4.	Microbiological Regeneration	292
4.1.	Regeneration of BACs	292
4.1.1.	Adsorption, biodegradation and bioregeneration pathways	292
4.2.	Bioregeneration	292
4.2.1.	Regeneration pathways	293
4.2.2.	Influence of experimental variables on Bioregeneration	293
4.2.3.	RE of Bioregeneration	293
5.	Vacuum Regeneration	293
5.1.	VSA	293
5.1.1.	Thermal Vacuum Swing Adsorption (TVSA)	294
5.1.2.	VSA and TVSA's applications	294
6.	Conclusions	294
	Acknowledgements	295
	Appendix A. Supplementary data	295
	References	295

1. Introduction

The excellent adsorbent properties of porous carbonaceous materials, such as activated carbons (ACs), are responsible for their extensive use in many fields. Polluted gaseous and liquid stream purification is the most common application and consumes the most ACs. During these treatments, pollutants continuously accumulate on the adsorbent surface. As a result, the adsorption capacity is progressively reduced until the material is finally exhausted. After saturation, the exhausted ACs are replaced with fresh materials and the used ACs are burnt or disposed of in landfills, but this option entails important environmental and economic setbacks. The regeneration of these materials is an alternative to disposal that allows them to be re-used. Regeneration attempts to remove pollutants retained on the ACs' surfaces to restore adsorptive capacity without modifying porosity or causing adsorbent mass losses. By this way, Regeneration produces valuable products and avoids the contamination that disposal of this solid residue would cause.

The relevance of this process is reflected in the numerous methods developed and published investigations concerning the regeneration of carbonaceous adsorbents. Despite this interest, to our knowledge, no article has yet strictly classified or deeply described existing regeneration methods. Therefore, the main goals of this review are as follows: (i) establishing criteria to classify existing regeneration methods in a clear and rigorous manner and (ii) describing all of these methods in-depth.

This review maintains the traditional division of regeneration methods, which distinguishes among three major groups: Thermal, Chemical and Microbiological Regeneration. A fourth major group, Vacuum Regeneration, is added. To classify regeneration methods included within each of the four main groups properly, the following double criterion focused on two fundamental aspects is proposed [1]:

1. The mechanism of regeneration: the regeneration pathways involved in the removal of the adsorbates retained on the ACs.
2. The type of regenerating agent: the primary agent that induces regeneration is not considered in isolation. The presence of a secondary agent, such as a catalyst or ultraviolet radiation, modifies the regeneration process and consequently, represents a different method.

Within the four major groups, each method represents a unique “regeneration mechanism-regeneration agent” combination that differentiates it from the others. Table 1 displays the general classification of the regeneration methods along with the mechanisms and regeneration agents included within each major group. This criterion also allows any future methods to be properly classified.

Thermal Regeneration, which involves heating the spent adsorbents, was analyzed in a previous review [1]. Chemical, Microbiological and Vacuum Regeneration are considered in this review.

Chemical Regeneration employs an extensive variety of reagents to remove adsorbates. The main characteristic of Microbiological Regeneration is the employment of microorganisms as the regeneration agents. Vacuum Regeneration shifts the adsorption equilibrium towards desorption through pressure changes in adsorption-regeneration cycles.

2. Important definitions

It is necessary to describe a number of concepts that will appear throughout this study:

- AC forms: different types of AC have differing morphology and particle size. These materials include granular activated carbon (GAC), powdered activated carbon (PAC), activated carbon fibers (ACFs), carbon nanotubes (CNTs), cloths, monoliths, composites and extruded or pellet forms.

Download English Version:

<https://daneshyari.com/en/article/72920>

Download Persian Version:

<https://daneshyari.com/article/72920>

[Daneshyari.com](https://daneshyari.com)