



# Investigations on the structural, electronic, elastic and thermodynamic properties of niobium silicide under high temperature and pressure

Ning Xu <sup>a,\*</sup>, Yue Xu <sup>b</sup>, Jun Ma <sup>c</sup>

<sup>a</sup> Key Laboratory for Advanced Technology in Environmental Protection of JiangSu Province Yancheng Institute of Technology, Yancheng 224051, China

<sup>b</sup> School of Material, The University of Manchester, Manchester M13 9PL, UK

<sup>c</sup> Institute of Nuclear Physics and Chemistry, China Academy of Engineering Physics, Mianyang 621900, China



## ARTICLE INFO

Available online 11 October 2014

### Keywords:

NbSi<sub>2</sub>

Structural

Electronic

Elastic

Thermodynamic

## ABSTRACT

A first principles investigation on the structural, electronic and elastic properties of NbSi<sub>2</sub> has been calculated using the plane wave pseudo-potential density functional theory (DFT) method. The lattice constants and elastic constants are obtained. The characters of the band structure and the density of states of niobium silicide (NbSi<sub>2</sub>) are analyzed, which shows that NbSi<sub>2</sub> is metallic. The bulk, shear and Young's modulus for NbSi<sub>2</sub> are also calculated. The Debye temperature is calculated from the average elastic wave velocity obtained from the shear and bulk modulus. The thermodynamic properties including the heat capacity  $C_V$  and  $C_P$ , thermal expansion  $\alpha$ , entropy  $S$  and Grüneisen parameter have been calculated at temperatures from 0 to 2100 K and pressures from 0 to 30 GPa using the quasi-harmonic Debye model. We also found that the present work can give a reference to those not to be experimentally investigated.

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## 1. Introduction

Transition metal disilicide are of much interest due to the variety of their physical properties and practical applications. These have also two crystal structures: i.e. the tetragonal C11<sub>b</sub> structure and the hexagonal C40 structure. The C11<sub>b</sub> structure has a comprehensive mechanical properties at high temperatures, but low fracture toughness at low temperatures [1,2]. C40 structure compounds are attractive material for microelectronic devices, due to their desirable physical properties, such as high melting temperature, low density and high temperature anomalous yield behavior [3,4]. Among these C40 compounds,

NbSi<sub>2</sub> is a typical transition metal disilicide compounds. In recent years, there are many experimental and theoretical studies on the NbSi<sub>2</sub>. In experimental studies, Chu et al. have studied the elastic properties of hexagonal C40 transition metal disilicides NbSi<sub>2</sub> and TaSi<sub>2</sub> at room temperature and low temperature using Resonant Ultrasound Spectroscopy (RUS) [5,6]. Laborde et al. have studied the phonon spectra of metallic disilicides VSi<sub>2</sub>, NbSi<sub>2</sub> and TaSi<sub>2</sub> at 300 K and specific heat measurements between 10 K and 250 K by inelastic neutron scattering [7]. Antonov et al. have investigated the optical properties of NbSi<sub>2</sub> experimentally and theoretically [8]. Lasjaunias et al. have been also extensively studied the specific heat of NbSi<sub>2</sub> [9].

However, there are few reports on the elastic and thermodynamic properties of C40 structure NbSi<sub>2</sub> with space group P6<sub>4</sub>22( $D_{6h}^5$ ) at high temperature and high pressure in theory. Therefore, the aim of this work is to

\* Corresponding author.

E-mail address: [xuning196402@163.com](mailto:xuning196402@163.com) (N. Xu).

give a systematic theoretical study of elastic and thermodynamic properties of NbSi<sub>2</sub>.

## 2. Method of calculations

The present first principles DFT calculations are done by means of the CASTEP code [10,11]. The interactions between the electrons and core ions are treated by using ultrasoft pseudopotentials [12]. For the geometry optimizations and property calculations, the exchange–correlation effects are mainly taken into account using the Ceperley–Alder data as parameterized by Perdew–Zunger (CA-PZ) of local density approximation (LDA) [13]. Kinetic-energy cutoff is 600 eV for the plane waves and Brillouin zone integration is performed using a 8 × 8 × 8 Monkhorst–Pack *k*-point sampling for a primitive cell. The self-consistent convergence of the total energy is 1 × 10<sup>−5</sup> eV atom<sup>−1</sup>. Both the *k*-point sampling and cutoff have been tested to provide sufficient precision in our calculations.

In order to investigate the thermal properties of NbSi<sub>2</sub> at high temperature and high pressure we have used the quasi-harmonic Debye model as implemented in the Gibbs code [14]. The non-equilibrium Gibbs function *G*<sup>\*</sup>(*V*; *P*, *T*) takes the form of

$$G^*(V; P, T) = E(V) + PV + A_{vib}(\theta(V); T) \quad (1)$$

where *E*(*V*), *PV*, and *θ*(*V*) are the total energy per unit cell, the constant hydrostatic pressure condition, the Debye temperature, respectively. The Helmholtz free energy *A*<sub>vib</sub> can be written as [15]

$$A_{vib}(\theta; T) = nkT \left[ \frac{9\theta}{8T} + 3 \ln(1 - e^{-\theta/T}) - D(\theta/T) \right] \quad (2)$$

where *D*(*θ*/*T*) and *n* are the Debye integral and the number of atoms per formula unit, respectively. *θ* is expressed by [16]

$$\theta = \frac{\hbar}{k} [6\pi^2 V^{1/2} n]^{1/3} f(\sigma) \sqrt{\frac{B_S}{M}} \quad (3)$$

*M* being the molecular mass per unit cell and *B<sub>S</sub>* is the adiabatic bulk modulus, approximated by the static compressibility

$$B_S \cong B(V) = V \left\{ \frac{d^2 E(V)}{dV^2} \right\} \quad (4)$$

*f*(*σ*) is given by Refs. [17,18], and *σ* is the Poisson ratio. Therefore, the non-equilibrium Gibbs function *G*<sup>\*</sup>(*V*; *P*, *T*) as a function of (*V*; *P*, *T*) can be minimized with respect to volume *V*

$$\left( \frac{\partial G^*(V; P, T)}{\partial V} \right)_{P, T} = 0 \quad (5)$$

By solving the Eq. (5), one can get the thermal equation of state *V*(*P*, *T*).

The constant volume heat capacity *C<sub>V</sub>* and the constant pressure capacity *C<sub>P</sub>*, the thermal expansion coefficient *α* and entropy *S* are taken as

$$C_V = 3nk \left[ 4D(\theta/T) - \frac{3\theta/T}{e^{\theta/T} - 1} \right] \quad (6)$$

$$C_P = C_V(1 + \alpha\gamma T) \quad (7)$$

$$\alpha = -\frac{\gamma C_V}{B_S V} \quad (8)$$

$$S = nk[4D(\theta/T) - 3 \ln(1 - e^{-\theta/T})] \quad (9)$$

*γ* is the Grüneisen parameter, which is defined as

$$\gamma = -\frac{d \ln \theta(V)}{d \ln V} \quad (10)$$

## 3. Results and discussions

### 3.1. Structural properties

NbSi<sub>2</sub> was refined to be a hexagonal crystal with the space group P6<sub>4</sub>22(*D*<sub>6h</sub><sup>5</sup>). The atomic positions are Nb 3c (0.5, 0, 0), Si 6i (0.167, 0.334, 0) [19]. The crystal structure is shown in Fig. 1.

All physical properties are in connected with the total energy. For instance, the equilibrium structural of a crystal is the lattice constants when the total energy obtains its minimum. Once the total energy is obtained, any physical property concerning the total energy can be determined. In a first step a set of calculations were done around optimized minima to obtain the equilibrium ground state energy–volume (*E*–*V*) data, where *V*<sub>0</sub> is the equilibrium volume at zero pressure, which are fitted to the Murnaghan equation of state [20]

$$\Delta E(V) = E - E_0 = B_0 V_0 \left[ \frac{V_n}{B_0'} + \frac{1}{1 - B_0'} + \frac{V_n^{1-B_0'}}{B_0'(B_0' - 1)} \right] \quad (11)$$

where *E*<sub>0</sub> is the equilibrium energy at zero pressure.

The pressure *P* versus the normalized volume *V<sub>n</sub>* = *V*/*V*<sub>0</sub> (where *V*<sub>0</sub> is the equilibrium volume at zero pressure) is obtained through the following relationship:

$$P = -\frac{dE}{dV} = \frac{B_0}{B_0'} [V_n^{-B_0'} - 1] \quad (12)$$

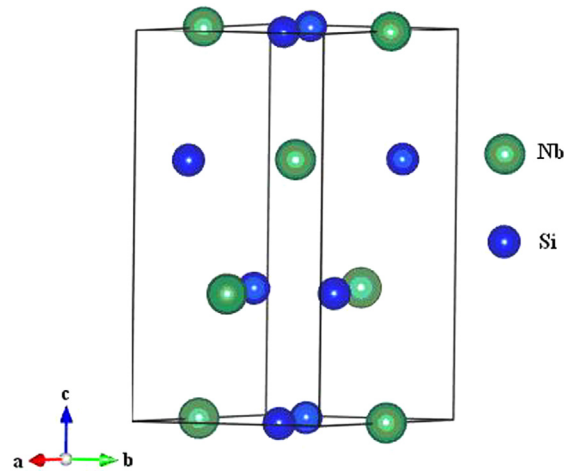


Fig. 1. Crystal structure of NbSi<sub>2</sub>.

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