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Infrared spectral properties for α -NaYF₄ single crystal of various Er³⁺ doping concentrations

Cheng Wang^a, Haiping Xia^{a,*}, Zhigang Feng^a, Zhixiong Zhang^a, Dongsheng Jiang^a, Jian Zhang^a, Shinan He^a, Qingyang Tang^a, Qiguo sheng^a, Xuemei Gu^a, Yuepin Zhang^a, Baojiu Chen^b, Haochuan Jiang^c

^a Key laboratory of Photo-electronic Materials, Ningbo University, Ningbo 315211, China

^b Department of Physics, Dalian Maritime University, Dalian 116026, China

^c Ningbo Institute of Materials Technology and Engineering, the Chinese Academy of Sciences, Ningbo 315211, China

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ABSTRACT

The α -NaYF₄ single crystals doped with 0.2 mol%, 0.5 mol%, 1 mol%, 2 mol% and 3 mol% Er³⁺ ions were fabricated by an improved Bridgman method. The $\sim 1.5 \mu\text{m}$ and $\sim 2.7 \mu\text{m}$ emission properties were investigated in detail through the measured absorption and emission spectra. A Judd–Ofelt analysis of Er³⁺ in α -NaYF₄ samples is performed. The Judd–Ofelt parameters are obtained ($\Omega_2=2.71 \times 10^{-20} \text{ cm}^2$, $\Omega_4=2.28 \times 10^{-20} \text{ cm}^2$, $\Omega_6=0.84 \times 10^{-20} \text{ cm}^2$) and the radiative lifetimes of Er³⁺ energy levels and the branching ratios of Er³⁺ transitions are calculated. The calculated maximum emission cross section by McCumber theory for $\sim 1.5 \mu\text{m}$ and $\sim 2.7 \mu\text{m}$ reached $1.35 \times 10^{-20} \text{ cm}^2$ and $2.2 \times 10^{-20} \text{ cm}^2$, respectively. The gain cross section spectra were calculated based on the absorption and emission cross section spectra. All these spectral properties indicated that this kind of fluoride crystal has potential application as host material for infrared lasers.

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1. Introduction

In recent decades, the infrared solid state lasers have attracted much attention because of their many applications including environment monitoring, laser processing, distance measurement, military and many other fields [1–3]. It is noted that rare-earth ions in the low phonon energy of the host material in the infrared multiphonon relaxation transition rate is greatly reduced, so that the infrared radiation transitions become possible [4]. Erbium ion (Er³⁺) is one of the most important rare earth ions due to the efficient emission band from UV to mid-infrared. Up to know, a lot of Er³⁺ doped crystals such as Er³⁺:Y₃Al₅O₁₂ [5], Er³⁺:Ca₉Y(VO₄)₇ [6], Er³⁺:La₃Ga_{5.5}Ta_{0.5}O₁₄ [7], Er³⁺:LiNbO₃ [8] and Er³⁺:LiYF₄ [9] have shown efficient spectral properties, and the lasers at about $\sim 1.5 \mu\text{m}$ via $^4\text{I}_{13/2} \rightarrow ^4\text{I}_{15/2}$ have been outputted successfully from activator Er³⁺-based solid materials [10–14]. Recently, the development of mid-infrared laser at $\sim 2.7 \mu\text{m}$ via $^4\text{I}_{11/2} \rightarrow ^4\text{I}_{13/2}$ transitions of Er³⁺ doped solid materials becomes considerable interest because of their special application in military and environment monitoring.

It is well known that the luminescence of Er³⁺ in traditional oxide crystals tend to be quenched easily by the rapid multi-

phonon effect due to their high phonon energies. Furthermore, in order to obtain high luminescent efficiency at $\sim 2.7 \mu\text{m}$, matrix with low phonon energy should be chosen because of the narrow energy difference between $^4\text{I}_{11/2}$ and $^4\text{I}_{13/2}$ levels. Thus it becomes vital and important to explore new host materials with low phonon energies.

The cubic α -NaYF₄ has been widely studied because of its low maximum phonon energy and the applications of these crystals as excellent hosts for up-conversion [15]. More recently, the α -NaYF₄ single crystals doped with Ho³⁺/Yb³⁺ [16], and Tm³⁺ [17] were successfully grown in our laboratory by an improved flux-Bridgman method. The transmittance in the range of infrared and mid-infrared, and Raman spectra of α -NaYF₄ single crystal have been already recorded in our laboratory [18]. It shows high transmittance in range of infrared and mid-infrared and the maximum phonon energy of α -NaYF₄ single crystal about 390 cm^{-1} . Compared with LiYF₄ single crystal [9], the α -NaYF₄ shows further lower phonon energy and higher transparency in range of infrared and mid-infrared [16,17]. The excellent properties of α -NaYF₄ single crystal in optical behaviors and physical-chemical properties are beneficial for infrared and mid-infrared laser application to obtain high luminous efficiency and high stability. However, there are few reports about Er³⁺ ion doped α -NaYF₄ for infrared and mid-infrared laser because of the difficulty of the crystal growth.

In this paper, α -NaYF₄ samples with different Er³⁺ doping

* Corresponding author.

E-mail address: hpxcm@nbu.edu.cn (H. Xia).

concentrations (0.2, 0.5, 1, 2, 3 mol%) were grown successfully by an improved Bridgman method using KF as flux. The absorption spectrum was detected and the optical parameters were calculated. The emission spectrum of $\sim 1.5 \mu\text{m}$ and $\sim 2.7 \mu\text{m}$ for Er^{3+} singly doped $\alpha\text{-NaYF}_4$ crystal were investigated. It is expected that this study provides a good theoretical basis for the further development and implementation of infrared devices.

2. Experimental

The Er^{3+} -doped $\alpha\text{-NaYF}_4$ single crystals in 10 mm diameter investigated in our study were fabricated by the Bridgman method with molar compositions of $30\text{NaF}-18\text{KF}-(52-\chi)\text{YF}_3-\chi\text{ErF}_3$ ($\chi=0.2, 0.5, 1.0, 2.0, 3.0$). The mixtures were treated by fluoridated method with anhydrous HF at $750\text{--}800^\circ\text{C}$ for 8–10 h to remove the influence of the hydroxyl ion OH^- concentration in these crystals. A low OH^- content is important in order to eliminate the non-radiative energy transfer from the excited level of Er^{3+} ion to OH^- . The specific details and processes of crystal growth were described in Ref. [19].

The polished slices of various concentrations are shown in the inset of Fig. 1(a). The X-ray diffraction patterns of samples were recorded with a XD-98X diffractometer (XD-3, Beijing). The concentration of Er^{3+} in this sample was measured with inductively coupled plasma atomic emission spectrum (ICP-AES, PerkinElmer Inc., Optima 3000). The absorption spectrum was detected by a Cary 5000 UV/VIS/NIR spectrophotometer in the wavelength region from 200 nm to 2200 nm. The emissions spectra were investigated with a Traix 320 type spectrometer in the range of 1000–1800 nm, and 2500–3000 nm excited by 980 nm light. All measurements were carried out at room temperature. A FLSP 920 type spectrometer (Edinburgh Co., England) excited by 980 nm LD was used to obtain the fluorescence lifetime.

3. Results and discussion

3.1. X-ray diffraction

Although we have obtained $\alpha\text{-NaYF}_4$ crystals with different doping concentrations of Er^{3+} , the higher doping level has a greater impact on the structure of crystal, and the XRD diffraction pattern of the sample with 3 mol% is plotted in Fig. 1(b). By comparing with the JCPDS card of $\alpha\text{-NaYF}_4$ (No.77-2042) of cubic

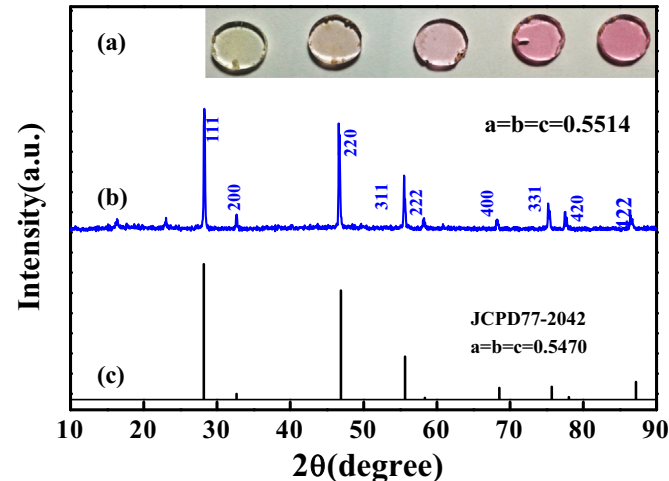


Fig. 1. (a) XRD curve of Er^{3+} singly-doped $\alpha\text{-NaYF}_4$ crystal with molar concentration of 3 mol%. (b) The standard line pattern of $\alpha\text{-NaYF}_4$ (JCPDS 77-2042).

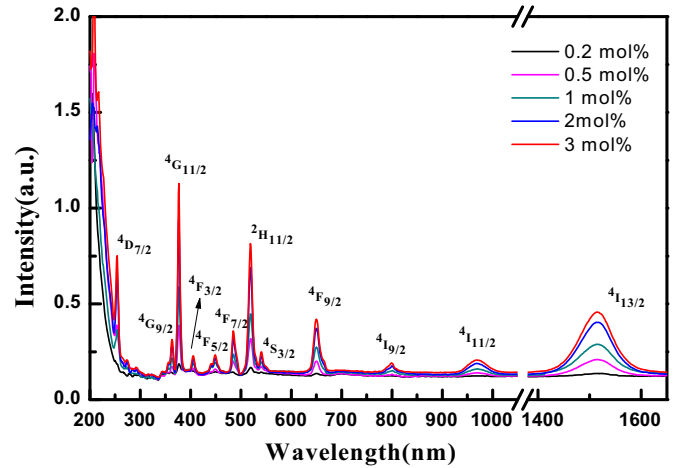


Fig. 2. Absorption spectra of various concentrations of Er^{3+} -doped $\alpha\text{-NaYF}_4$ single crystals.

system [16–18] illustrated in Fig. 1(c), it can be seen that the current doping level does not cause any obvious peak shift or another phase, indicating that this transparent crystal has pure cubic phase and Er^{3+} ions perfectly entered the Y^{3+} sites. The calculated cell parameters from the XRD pattern for Er^{3+} singly doped $\alpha\text{-NaYF}_4$ crystal are $a=b=c=0.5514 \text{ nm}$. As labeled in Fig. 1, the lattice parameters of the Er^{3+} -doped one are a bit larger than the blank matrix, it is well in agreement with the fact that the radius of Er^{3+} is greater than Y^{3+} .

3.2. Absorption spectra and Judd–Ofelt analysis

Fig. 2 presents the absorption spectra of $\alpha\text{-NaYF}_4$ samples with different erbium doping concentrations in the wavelength region of 200–1050 nm and 1400–1650 nm at room temperature. The absorption bands at 377, 487, 517, 650, 792, 980, 1517 and 2700 nm of Er^{3+} , which are closely related to the transitions from the ground state $^4\text{I}_{15/2}$ to $^4\text{G}_{11/2}$, $^4\text{F}_{7/2}$, $^2\text{H}_{11/2}$, $^4\text{F}_{9/2}$, $^4\text{I}_{9/2}$, $^4\text{I}_{11/2}$ and $^4\text{I}_{13/2}$ excited states. It can be seen from Fig. 2 that the absorption intensity gradually increases as the doping concentration changes from 0.2 to 3 mol%, and there are no miscellaneous peaks appearing or any peak shifts. A broad absorption peak concentrated at 980 nm indicates that Er^{3+} ion can act as an excellent activator to absorb the 980 nm pump light. Therefore, the excellent infrared transmission property provides potential application for mid-infrared laser material.

Judd–Ofelt (J–O) theory [20,21] is extensively used in the analysis of spectroscopic characters of rare earth elements in laser materials. Based on J–O theory, the absorption spectra can be used to calculate the oscillator strength parameter Ω_t ($t=2, 4, 6$), radiative lifetime τ_{rad} , fluorescence branching ratio β and transition probability A . To the best of our knowledge, there are rarely Judd–Ofelt analyses of Er^{3+} in hosts isomorphous with $\alpha\text{-NaYF}_4$ published in the literature. In this work, the J–O intensity parameters of Er^{3+} -doped $\alpha\text{-NaYF}_4$ were calculated.

According to the absorption spectrum, the experimental oscillator strengths are obtained by the following expression [22]:

$$f_{\text{exp}} = \frac{mc^2}{\pi e^2 N} \int \frac{2.303 OD(\lambda)}{\lambda^2 d} d\lambda \quad (1)$$

where m is the electron mass, e is the electron charge, N is the number of the absorbing ions in unit volume, $OD(\lambda)$ is the optical density, c is the speed of light and d is the thickness of the sample.

According to Judd–Ofelt theory, the calculated oscillator strengths are obtained by the following expression:

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