Accepted Manuscript

The complexity of condensed tannin binding to bovine serum albumin – An isothermal titration calorimetry study

Rachel L. Kilmister, Peta Faulkner, Mark O. Downey, Samuel J. Darby, Robert J. Falconer

PII: S0308-8146(15)00732-3

DOI: http://dx.doi.org/10.1016/j.foodchem.2015.04.144

Reference: FOCH 17557

To appear in: Food Chemistry

Received Date: 10 December 2014
Revised Date: 26 March 2015
Accepted Date: 29 April 2015



Please cite this article as: Kilmister, R.L., Faulkner, P., Downey, M.O., Darby, S.J., Falconer, R.J., The complexity of condensed tannin binding to bovine serum albumin – An isothermal titration calorimetry study, *Food Chemistry* (2015), doi: http://dx.doi.org/10.1016/j.foodchem.2015.04.144

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.

ACCEPTED MANUSCRIPT

The complexity of condensed tannin binding to bovine serum albumin – An isothermal titration calorimetry study

Rachel L. Kilmister^{1*}, Peta Faulkner¹, Mark O. Downey¹, Samuel J. Darby², Robert J. Falconer²

- Department of Environment and Primary Industries, Victoria, PO Box 905, Mildura, VIC, 3502,
 Australia
- 2. Department of Chemical and Biological Engineering, ChELSI Institute, University of Sheffield, Sheffield, S1 3JD, England
- * To whom the correspondence should be addressed. Telephone +61 3 5051 4541; Fax +61 3 5051 4523; Email: rachel.kilmister@depi.vic.gov.au

Keywords: Proanthocyanidin, oligomer, thermodynamics, hydrogen bonding, hydrophobic interaction

Abstract

Isothermal titration calorimetry was applied to study the binding of purified proanthocyanidin oligomers to bovine serum albumin (BSA). The molecular weight of the proanthocyanidin oligomer had a major impact on its binding to BSA. The calculated change in enthalpy (ΔH) and association constant (K_a) became greater as the oligomer size increased then plateaued at the heptameric oligomer. These results support a model for precipitation of proteins by proanthocyanidin where increased oligomer size enhanced the opportunity for cross linkages between proteins ultimately forming sediment-able complexes. The authors suggest tannin binding to proteins is opportunistic and involves multiple sites, each with a different K_a and ΔH of binding. The ΔH of binding comprises both an endothermic hydrophobic interaction and exothermic hydrogen bond component. This

Download English Version:

https://daneshyari.com/en/article/7590337

Download Persian Version:

https://daneshyari.com/article/7590337

<u>Daneshyari.com</u>