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Quantitative Proton Nuclear Magnetic Resonance evaluation and total assignment of the capsular polysaccharide *Neisseria meningitidis* serogroup X

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ABSTRACT

Neisseria meningitidis constitutes the main cause of meningococcal disease in infants. Serogroups A, B, C, W135, Y, and X have the higher incidence in young children and teenagers. The use of polyvalent conjugate carbohydrate-based vaccines has decreased the meningococcal infection around the world. Recently, the serogroup X has been found to be responsible of different outbreaks of meningococcal diseases, mainly in "Meningitis Belt" of Africa and the structure of the repetitive unit of the capsular polysaccharide has been confirmed through a monodimensional ¹³C NMR study. No further characterization studies have been carried out, especially with the use of other nuclei. In this paper a novel method for quantification of the *N. meningitidis* serogroup X by proton qNMR is reported. Deep characterization of the serogroup X polysaccharide was also carried out by combination of correlation experiments involving ¹³C, ¹H, and ³¹P nuclei.

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1. Introduction

Neisseria meningitidis is one of the main pathogens responsible of meningococcal disease in infants and adolescents. Thirteen serogroups have been reported, however six of them (A, B, C, W135, Y and X) are the main cause of outbreaks, endemic and epidemic meningococcal diseases [1]. The serogroup X was first described in 1966 by Bories et al. [2] This strain causes substantial diseases in the "Meningitis Belt" of Africa but only rarely causes meningococcal disease in other parts of the world. Different outbreaks by this group have been reported in Niger, Ghana, Kenya, Burkina Faso, Togo, and Uganda since 2006 [3–5]. Several vaccines are employed to prevent the meningococcal infections from serogroups A, B, C, W135, and Y but none of them induce

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protection against X serogroup [6]. This serogroup has become a critical target for the next generation of meningococcal vaccines. Importantly, in 1974 Bundle et al. confirmed the structure for the repetitive unit of native polysaccharide through a monodimensional ¹³C NMR study [7]. The serogroup X polysaccharide (PsX) is a homopolymer of linked 2-acetamido-2-deoxyglucosyl phosphate $\alpha 1 \rightarrow 4 [\rightarrow 4) - \alpha - D - GlcpNAc - (1 \rightarrow OPO_3 \rightarrow);$ (see Fig. 1) No further characterization studies have been carried out especially with the use of other nuclei. However, many NMR correlations experiments developed since 1980 and some capabilities of modern NMR spectrometers can now expand the earlier studies [8,9]. Hence, a comparative structural characterization of PsX is recommended. In the last decade, the application of quantitative Proton Nuclear Magnetic Resonance had an increasing impact on the pharmaceutical industry. The qNMR method involves NMR experiments with several modified parameters in order to obtain spectra with quantifiable signals (99.9% recovery of thermal equilibrium magnetization). These parameters depend essentially on the relaxation times of the nuclei to be evaluated for the analyte and the reference [10,11]. This paper discusses a novel Proton Nuclear Magnetic Resonance method for the quantification of N. meningitidis serogroup X capsular polysaccharide. In addition, the total NMR assignment of the PsX as result of combining correlation experiments of ¹H, ¹³C, and ³¹P nuclei is reported as a complementary study.

Abbreviations: ¹H NMR, Proton Nuclear Magnetic Resonance; ¹³C NMR, Carbon 13 Nuclear Magnetic Resonance; qNMR, Quantitative Nuclear Magnetic Resonance; QNP, Quadruple Nucleus Probe; HSQC, Heteronuclear Single Quantum Coherence; PENDANT, Polarization Enhancement During Attached Nucleus Testing; DQF-COSY, Double Quantum Filtered Correlation Spectroscopy; GARP, Globally optimized Alternating-phase Rectangular Pulses; LB, Line Broadening.

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Fig. 1. Chemical structure of PsX repetitive unit.

2. Materials and methods

2.1. Sample preparation

The capsular polysaccharide of N. meningitidis serogroup X was supplied by the Technological Development Department from the Finlay Institute, Havana, Cuba as a result of collaborative scientific project between Norwegian Institute of Public Health (NIPH), Norway and Finlay Institute. One aliquot was taken from a stock solution to determine the PsX content by the colorimetric assay. From the colorimetric result several aliquots were also taken to prepare three sets of samples for gNMR with seven different quantities of PsX per set [(A) 5.84 mg, (B) 7.61 mg, (C) 11.12 mg, (D) 15.55 mg, (E) 16.46 mg, (F) 19.69 mg, and (G) 21.96 mg]. The aliguots were diluted in 0.6 ml of deuterium oxide for a lyophilization step. This stage allows the exchange of hydroxyl protons by deuterium. The resultant powder was again dissolved in 0.6 ml of deuterium oxide with 3-(trimethylsilyl) 2,2,3,3-tetra-deuteropropionic acid sodium salt (TPS-d₄) (1 mg/ml) as reference. For phosphorus spectra, one capillary with triphenylphosphine (PPh₃) as reference was inserted into the 5 mm NMR tube. All the solvents and reagents were of spectroscopic quality.

2.2. Colorimetric assay

Primary quantitative evaluation was carried out for a stock solution of PsX according to the total phosphorus determination proposed by Chen et al. [12].

2.3. Instrument settings

NMR analyses were carried out on a Bruker Avance DPX 250-MHz instrument operating with a 5 mm QNP *z*-axis gradient probe at 27 °C. The ¹³C NMR edited spectrum was obtained through a PENDANT experiment. The edited HSQC (¹H–¹³C) delay for single quantum evolution was optimized for ¹J_{H–C} = 150 Hz. The HSQC (¹H–³¹P) delay for developing the heteronuclear correlation was optimized for ³J_{H–P} = 7 Hz. DQF-COSY experiments were run in the magnitude mode [13]. The TSP-d₄ non-deuterated methyl signal was referenced at δ 0.00 ppm to calibrate the spectra. The phosphorus spectra were calibrated referencing the PPh₃ signal at δ –5.00 ppm [13].

The optimized proton qNMR experiment was set without sample spinning, a recycled time of 8 s, pulse duration P1 adjusted as reported by Ernst and Anderson [14] and corresponding to 88°, and ¹³C decoupling using the GARP sequence. To obtain the spectra, each Free Induction Decay (FID) was processed with a Fourier Transform apodizated by an exponential function (LB 0.3 Hz). To guarantee the signal integration process baseline adjustment was required. The spectral data were acquired and transformed under TopSpin 1.3 software pack.

2.4. Statistical analysis

The statistical evaluation of the proposed method was tested by the study of the following parameters: specificity, linearity, precision (intra and inter day), accuracy, and ruggedness. All statistical calculations were carried out according to International Conference on Harmonization (ICH) guidelines [15,16].

3. Results and discussion

Unlike the only approach for the PsX assignment carried out by ¹³C NMR by Bundle et al. [7], we have made a total assignment of the PsX repeating unit using also ¹H and ³¹P NMR. In addition, we have developed an assay for the quantitative evaluation of the polysaccharide optimizing several parameters of the ¹H NMR spectrum.

3.1. Assignment of the polysaccharide by ¹H, ¹³C, and ³¹P NMR

The ¹³C PENDANT (Fig. 2a) spectrum shows some peaks as doublets due to the heteronuclear coupling between carbon and phosphorus nuclei [7,13]. The singlet of the carbonyl group is a negative peak at low field (δ 175.1 ppm). The other negative peak is the methylene carbon (C-6) at δ 60.9 ppm. The doublet of the anomeric carbon (C-1) appears in the expected region at δ 94.6 ppm and ${}^{2}J_{C-P}$ = 5.9 Hz. The methyl of the N-acetyl methyl group shows a singlet at high field (δ 22.7 ppm). Carbon C-2 is a doublet at δ 54.2 ppm (${}^{3}J_{C-P}$ = 8.5 Hz). The other carbons (C-3, C-4, and C-5) show close doublets at δ 70.6, 74.5, and 72.6 ppm with coupling constants of ${}^{3}J_{C-P}$ = 1.5 Hz, ${}^{2}J_{C-P}$ = 6.0 Hz and ${}^{3}J_{C-P}$ = 5.1 Hz, respectively. These last assignments were compared with the literature [7] and confirmed through the combination of the edited HSQC (${}^{1}H-{}^{13}C$), HSQC (${}^{1}H-{}^{31}P$) and the DQF-COSY experiments; see Table 1.

The ¹H NMR spectrum shows the anomeric proton (H-1) as doublet of doublet at δ 5.57 ppm due to the heteronuclear $({}^{3}J_{H-P}=6.4 \text{ Hz})$ and the homonuclear coupling constants $(J_{H-1-H-2} = 2.9 \text{ Hz})$ [7]. These values confirm the stereochemistry of the glycosidic bond [17]; see Fig. 2b. The singlet at δ 2.09 ppm belongs to the methyl of N-acetyl methyl group. The assignments of the region between δ 4.20 and 3.70 ppm were carried out step by step combining the bidimensional experiments. The only negative cross peak in the edited HSQC $(^{1}H^{-13}C)$ is the heteronuclear correlation of the methylene $(H-6a \rightarrow C-6 \text{ and } H-6b \rightarrow C-6)$ [13]. The shift of H-2 (δ 4.07 ppm) was also assigned from the edited $HSQC(^{1}H-^{13}C)[13];$ (Fig. 3a). The signal of H-4 was unambiguously assigned at δ 4.11 ppm using the HSQC (¹H–³¹P) spectrum. The shift of H-5 at δ 4.05 ppm was provided by the COSY experiment through the scalar coupling cross peak between $H-6 \rightarrow H-5$. The analogous cross peak for $H-3 \rightarrow H-4$ correlation shows the final assignment of H-3 at δ 3.99 ppm (data not shown) [17]. H-3, H-4, and H-5 assignments are shown in Fig. 3b. The phosphorus assignment at δ -2.0 ppm was also made; (see Fig. 3c).

3.2. Specificity, structural analysis and spin lattice relaxation time (T_1) calculation

We have selected the region between δ 1.99 and 2.19 ppm (R-1) for quantitative evaluation of PsX due to this intense signal without interference. In addition, the region between δ –0.10 and 0.10 ppm (R-2) was selected for evaluating the reference; (Fig. 4). The specificity was demonstrated checking the no interference of any other component in the evaluated regions ensuring the specific contribution of the analyzed compounds PsX and TSP-d₄; see Fig. 3.

The conditions for acquiring the FID of qNMR were optimized using an inversion-recovery experiment to measure spin lattice relaxation times of evaluated protons (T_1) and to calculate the Ernst angle in order to provide a quantitative magnetization recovery. Download English Version:

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