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IR and NMR Spectroscopic Correlation of Enterobactin by DFT.

M. Moreno¹, A. Zacarias², A. Porzel³, L. Velasquez⁴, G. Gonzalez^{5,6}, M. Alegría-Arcos^{7,8}, F. Gonzalez-Nilo⁷, E.K.U. Gross².

¹University of the Basque Country, Barrio Sarriena, s/n, 48940 Leioa, Bizkaia, Spain

²Max Planck Institute of Microstructure Physics, Weinberg 2, D 06120 Halle, Germany and ETSF.

³Leibniz Institute of Plant Biochemistry, Weinberg 3, D 06120, Halle, Germany.

⁴ Universidad Andres Bello, Facultad de Medicina, Center for Integrative Medicine and Innovative Science. Echaurren 183,

Santiago, Chile.

⁵Center for Development of Nanoscience and Nanotechnology, CEDENNA, Casilla 653, Santiago, Chile

⁶Universidad de Chile, Facultad de Ciencias, Departamento de Química, Laboratorio de Sintesis Inorganica y electroquímica. Las Palmeras 3425, Nuñoa, Santiago, Chile.

⁷Universidad Andres Bello, Facultad de Ciencias Biologicas, Center for Bioinformatic and Integrative Biology. Av Republica 239, Santiago, Chile.

⁸Centro Interdisciplinario de Neurociencias de Valparaíso (CINV), Facultad de Ciencias, Universidad de Valparaíso, Valparaíso,

Chile.

Abstract.

Emerging and re-emerging epidemic diseases pose an ongoing threat to global health. Currently, Enterobactin and Enterobactin derivates have gained interest, owing to their potential application in the pharmaceutical field. As it is known [J. Am. Chem. Soc (1979) 101, 20, 6097-6104], Enterobactin (H₆EB) is an efficient iron carrier synthesized and secreted by many microbial species. In order to facilitate the elucidation of enterobactin and its analogues, here we propose the creation of a H₆EB standard set using Density Functional Theory Infrared (IR) and NMR spectra. We used two exchange-correlation (xc) functionals (PBE including long-range corrections – LC-PBE– and mPW1), 2 basis sets (QZVP and 6-31G(d)) and 2 grids (fine and ultrafine) for most of the H₆EB structures dependent of dihedral angles. The results show a significant difference between the O-H and N-H bands, while the C=O amide and O-(C=O)- IR bands are often found on top of each other. The NMR DFT calculations show a strong dependence on the xc functional, basis set, and grid used for the H₆EB structure. Calculated ¹H and ¹³C NMR spectra enable the effect of the solvent to be understood in the context of the experimental measurements. The good agreement between the experimental and the calculated spectra using LC-PBE/QZVP and ultrafine grid suggest the possibility of the systems reported here to be considered as a standard set. The dependence of electrostatic potential and frontier orbitals with the catecholamide dihedral angles of H₆EB is also reported of manner to enrich the knowledge about its reactivity.

Key words: Enterobactin, NMR, FT-IR, DFT and MALDI-TOF MS.

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