



Label-free electrochemiluminescent immunosensor for detection of prostate specific antigen based on mesoporous graphite-like carbon nitride

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ABSTRACT

In this paper, mesoporous graphite-like carbon nitride (mpg-C₃N₄) combined with gold nanoparticles (Au NPs) was conducted to construct a luminol-hydrogen peroxide (H₂O₂) based electrochemiluminescent (ECL) immunosensor for prostate specific antigen (PSA) detection. The mpg-C₃N₄ exhibited large specific surface area and high porosity was synthesized from a simple precursor, which possessed abundant active sites to load Au NPs and luminol (mpg-C₃N₄/Au-luminol). After that, Au NPs linked mpg-C₃N₄ (mpg-C₃N₄/Au) could hatch with the primary antibody (Ab₁) via Au-NH₂ bond, which reinforced the sensitivity of immunosensor. The mpg-C₃N₄ and Au shows excellent catalytic enhancement effect on the ECL intensity of luminol. Under optimal conditions, the constructed ECL immunosensor exhibited sensitive response to PSA in a wide linear range of 0.001 ~ 15 ng mL⁻¹ with a lower detection limit of 0.927 pg mL⁻¹. Meanwhile, the proposed ECL immunosensor endows the good reproducibility and stability. Therefore, the results could open another avenue for detection of PSA and other biomarkers.

1. Introduction

Biomarkers have been utilized as important diagnostic tools for cancers and other diseases [1]. Recently, a lot of methods have been focused on the detection of cancer markers, such as fluorescence (FL) [2–5], surface plasmon resonance [6–8], chemiluminescence (CL) [9,10], mass spectrometry [11]. Prostate cancer as one of the most common malignant diseases occurs in the male prostate tissue, which is caused by abnormal growth of the prostate gland cells. Prostate specific antigen (PSA) as tumor markers is commonly used to detect prostate and breast cancer whose concentration range in normal human body is 0 ~ 4 ng mL⁻¹ [12]. PSA has important clinical significance for early diagnosis, clinical staging, postoperative curative effect observation and follow-up of prostate cancer [13].

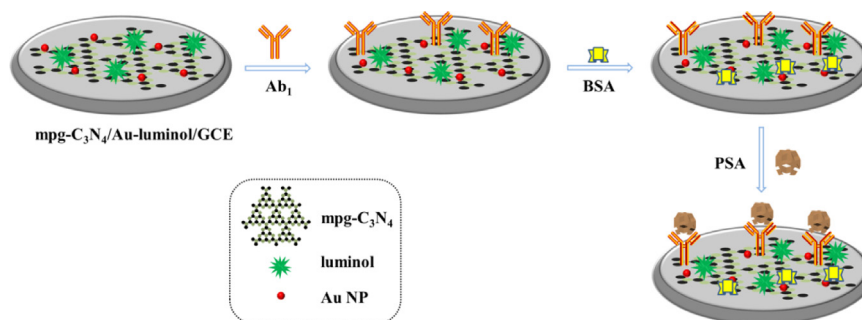
Electrochemiluminescent (ECL) as a simple, sensitive, and powerful analytical technique combines the advantages of electrochemical and luminescent techniques [14,15], involving a light emission process in a redox reaction of electrogenerated reactants. In the field of ECL, semiconductor quantum dots (QDs), Ru(bpy)₃²⁺, luminol and their analogues have been usually used as luminophores [16,17]. Unlike some other luminophores, luminol/H₂O₂ is widespread studied in virtue of its high luminescence yield, chemical stability and low potential. To the best of our knowledge, most of the reported system of

luminol is bound up with ROS, including superoxide radicals (O₂^{•-}), hydroxyl radical (OH[•]) and HO₂⁻. In order to promote the ECL efficiency of luminol/H₂O₂ system, a great deal of efforts have been devoted to produce highly reactive O₂^{•-} and OH[•] through catalyzing H₂O₂ [18]. Thus, biosensors based on luminol systems have been widely applied to the analysis filed [19]. Ju and his co-worker [20] had constructed a quenched immunosensor which based nitrogen doped graphene catalytic reduction of oxygen to produce hydroxide, thereby blocking the ECL process of CdS dots. Taking into account the advantages of luminol, ECL immunosensor based luminol has been constructed for PSA detection in our work.

To date, graphitic-like carbon nitride (g-C₃N₄) has given rise to a great deal of attention owing to its special structure and properties, and it is widely applied in catalysis [21–25], photoelectronic devices [26–28] and chemical sensors [29–34]. As we all know, g-C₃N₄ possessed good visible-light-driven photocatalytic performance and could produce hydrogen or oxygen by water splitting under visible light, which was firstly reported [27]. Compared with g-C₃N₄, mesoporous graphitic-like carbon nitride (mpg-C₃N₄) displays much higher surface area with abundant mesoporous. mpg-C₃N₄ can expose more active sites to enhance its application in catalytic reaction performance [35]. In order to improve the photoactivity of mpg-C₃N₄, researchers have developed several methods, such as doping with metal or nonmetallic

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Scheme 1. The fabrication process of the ECL modified electrodes.

elements [36], and coupling with inorganic semiconductors [37,38]. Although the ECL property of g-C₃N₄ was better than mpg-C₃N₄, mpg-C₃N₄ possessed good adsorption efficiency. Herein, mpg-C₃N₄ can be set as base material to load the luminescent materials in ECL sensor construction.

In this work, mpg-C₃N₄ was synthesized via a facile one-step method to construct a luminol-H₂O₂ based ECL immunosensor for PSA detection. The fabrication process of the developed ECL immunosensor for PSA is illustrated in Scheme 1. Due to the larger specific surface area and adsorption efficiency, mpg-C₃N₄ could combine with more Au NPs and luminol. Au NPs could accelerate the electron transport to realize the sensitive detection of PSA in human serum. Furthermore, mpg-C₃N₄ as a favorable substrate material could enhance the stability of the ECL signal remarkably. Notably, mpg-C₃N₄/Au could interact with luminol/H₂O₂ system to accelerate the ECL reaction for significantly enhancing the signal of the luminol/H₂O₂ system.

2. Experimental section

2.1. Synthesis of mpg-C₃N₄

The mpg-C₃N₄ was prepared according to previous reported method [35] through pyrolysis of urea and dicyandiamide in air atmosphere. Briefly, 3.5 g urea and 1.5 g dicyandiamide was mixed and thoroughly ground in a mortar (at least 50 min until the material was in ultrafine powder). Following, the abraded powder was transferred into a crucible and then put it in a Muffle Furnace. The certain parameter of calcination was setting as follows: from room temperature to 530 °C, and the heating rate was 5 °C min⁻¹. Then, maintain for 4 h; Final, declined to room temperature with -1 °C min⁻¹ and the yellow mpg-C₃N₄ powders was obtained. The g-C₃N₄ was prepared by pyrolysis of melamine under the same method.

2.2. Synthesis of Au NPs, mpg-C₃N₄/Au and mpg-C₃N₄/Au-luminol hybrids

The synthesis of Au NPs was conducted according previous methods [39]. Subsequently, 1.0 mL of mpg-C₃N₄ solution and 1.0 mL of Au NPs solution were mixed together sufficiently and then oscillation 24 h at 4 °C. After that the product was obtained by centrifugation. Finally, the product was dispersed in ultrapure water. 1.0 mL luminol was added into the above mpg-C₃N₄/Au solution and continued vibration 24 h at 4 °C. Afterward, the solution was also centrifuged and washed with ultrapure water. Finally, mpg-C₃N₄/Au-luminol hybrid were obtained and kept at 4 °C for future use.

2.3. Preparation of ECL modified electrodes

Firstly, the GCE (4 mm) was respectively polished using alumina slurry into mirror and then washed with ultrapure water. The process of modifying the GCE was shown in Scheme 1. Next, 6 μL of mpg-C₃N₄/Au-luminol hybrid was dropped and dried in room temperature.

Subsequently, 6 μL of anti-PSA solution (10 μg mL⁻¹) immobilized and incubated via Au-NH₂ bond, followed by washing to remove physically absorbed Anti-PSA. Then 3 μL 1% BSA solution was incubated with the GCE to block the nonspecific binding sites and washed thoroughly. After that step, different concentrations of PSA were added to the electrode and reacted with the Anti-PSA. After rinsing thoroughly, the ECL immunosensor was fabricated successfully.

2.4. Measurement procedure

To measure the UV-vis spectra, Au NPs were obtained from the reduction AuCl₄⁻. And g-C₃N₄/Au hybrid was prepared by a compound centrifugation of 1 mL of Au NPs solution and 1 mL of mpg-C₃N₄ solution. The concentration of mpg-C₃N₄/Au-luminol composed by 1 mL of mpg-C₃N₄/Au solution and 1 mL of 5 mmol L⁻¹ luminol solution.

The immunosensors incubated with different concentrations of PSA were put in 10 mL of pH 10.0 CBS (1/15 mol L⁻¹, Carbonate buffer solution contained Na₂CO₃ and NaHCO₃) and 30 mmol L⁻¹ H₂O₂ by single-step cycle pulse method. Pulse potential, pulse time, initial potential and pulse period were set at 0.55 V, 0.1 s, -0.45 V and 6 s, respectively. And the scanning rate was 0.1 V s⁻¹. The ECL signal was recorded via the MPI-F flow-injection chemiluminescence detector, and the voltage of the photomultiplier tube (PMT) was set at 500 V.

3. Result and discussion

3.1. Characterization of materials

The surface compositions of the obtained mpg-C₃N₄ were characterized by XPS. The XPS survey scan spectrum (Fig. 1C) showed that the main elements of mpg-C₃N₄ were carbon and nitrogen. As shown in Fig. 1A, the C 1s spectra could be divided into three peaks at 284.5 eV, 286.1 eV and 288 eV. It could be seen that a major C peak at 288 eV, which was identified as sp²-bonded C (C-N). The lower peak at 284.5 eV was attributed to C-C which was ordinarily observed on the XPS characterization for carbon nitrides [35]. The N 1s spectra (Fig. 1B) could be divided into 3 peaks at 398.4 eV, 400.7 eV and 404.3 eV. The higher N1s peak at 398.4 eV was attributed to N bonded to C (sp² hybridized aromatic). The lower peak at 400.7 eV was owing to quaternary N-C₃ in the aromatic cycles. While the additional weak peak at 404.3 eV was arisen from the amino functional groups with hydrogen (C-N-H), which might be related to incomplete condensation [40]. Furthermore, the crystalline structure of the prepared materials was measured by XRD in Fig. 1D. The strong XRD (curve a) peak at 27.6° (002) was consistent with the reported mpg-C₃N₄ [35]. After combined with Au NPs (curve b), three new diffraction peaks appears at 38.2°, 44.4°, and 64.5°, which corresponds to the (111), (200) and (220) plane of Au.

The morphological structure of the prepared materials was investigated by the measurement of SEM. As shown in Fig. 2A, the SEM image of mpg-C₃N₄ reveals that mpg-C₃N₄ was porous structure.

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