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Communication

Organocatalytic cascade 1,6-conjugate addition/annulation/ tautomerization of functionalized *para*-quinone methides: Access to chiral 2-amino-4-aryl-4*H*-chromenes

Cong Duan^a, Ling Ye^b, Wenqin Xu^a, Xinying Li^a, Feng Chen^a, Zhigang Zhao^a, Xuefeng Li^{a,*}

^a College of Chemistry and Environment Protection Engineering, Southwest Minzu University, Chengdu 610041, China ^b Faculty of Geosciences and Environmental Engineering, Southwest Jiaotong University, Chengdu 610031, China

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ABSTRACT

A novel organocatalytic cascade process initiated by 1,6-conjugated addition has been successfully developed. A range of pharmaceutically active 2-amino-4-aryl-4*H*-chromenes were readily obtained in high yields (88%-99%) and excellent enantiopurities (86%-99% *ee*). The functionalized *para*-Quinone methides (*p*-OMs) could be facilely obtained.

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2-Amino-4H-chromene is a class of privileged structural motif frequently appearing in natural products and pharmaceutically active compounds [1]. In particular, 2-amino-4H-chromene possessing a cyano group at the 3-position attracted considerable attention due to its powerful antitumor activity [1a,2]. Among all these compounds, 2-amino-3-cyano-4H-chromenes containing an aryl group at the 4-position were identified as potent apoptosis inducers [3]. These compounds were found to be potent inhibitors of tubulin polymerization as well and to effectively inhibit the binding of colchicine to tubulin [3]. MX-58151 (Fig. 1) retained activity in tumor cells resistant towards current antimitotic agents, taxanes (including Taxol and Taxotere) and Vinca alkaloids [2a,3-4]. Therefore, it represented a new class of potential therapeutic agent for the treatment of drug-resistant cancer. In addition, several 4-aryl-4H-chromenes were also determined to be effective vascular disrupting agents (VDA) [2b,5]. One of the lead VDA candidates, Crolibulin (Fig. 1), also known as EPC2407, is currently in phase I/II clinical trials [2b]. Notably, the antitumor activity of Crolibulin is closely correlated with its absolute configuration. It is revealed that the R-isomer is approximately 50-100 times more active than its S-enantiomer [6].

Considering these appealing pharmaceutical activities, elegant approaches have been proposed to prepared these enantioen-riched 2-amino-3-cyano-4*H*-chromene [7]. However, to the best of

our knowledge, current protocol mainly focuses on the synthesis of 4-alkyl-substituted ones [7a-g]. The catalytic enantioselective synthesis of 4-aryl chromenes was relatively less explored [7h-n]. Meanwhile, the reported strategies for the synthesis of 4-aryl chromenes suffered from several drawbacks such as unsatisfactory enantiopurity [7i,k] and specialized substrate scope [7h,7j-m]. Therefore, developing a highly enantioselective and general strategy for the construction of chiral 2-amino-3-cyano-4-aryl-4H-chromene is still highly desirable.

para-Quinone methides (p-QMs) have been increasingly utilized as a class of prevailing Michael acceptor for 1,6-conjugate additions [8]. As a result, a range of diarylmethine compounds were obtained with high degrees of optical purities [9]. In contrast, the functionalized *p*-QMs have been less explored and the related domino reaction has not been well investigated, despite the great potential in the construction of complex diarylmethine-containing natural product and functional molecules [10]. In this context, Enders group presented a subtle domino oxa-Michael/1,6-conjugated addition process in a highly stereoselective manner, employed the functionalized ortho-hydroxyl-substituted p-QMs [11]. More recently, Jiang and his co-operator utilized this versatile precursor to construct densely functionalized spiro[chromane-2,1'-isochromene] scaffold in an achiral fashion [12]. Regretfully, the tedious prior preparation procedure of this functionalized starting material might limit their wide application. Based on our continuous effort in the asymmetric organocatalytic cascade reactions [7g], herein we would like to present a highly efficient tandem 1,6-conjugate addition/annulation/tautomerization of

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^{*} Corresponding author. E-mail address: lixuefeng@swun.edu.cn (X. Li).

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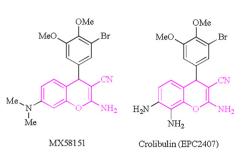


Fig. 1. Structures of the pharmaceutically active moleculars.

functionalized *p*-QMs [13]. A variety of 2-amino-3-cyano-4-aryl-4*H*-chromenes were afforded with high optical purities. Notably, the functionalized *ortho*-hydroxyl-substituted *p*-QMs could be feasibly accessed *via* one-step transformation (see Supporting information), using the simple commercially available precursor.

We commenced our investigation with utilizing ortho-hydroxyl p-QM 1a and malononitrile 2a as the model substrates. Initially, the catalytic effect of various bifunctional thiourea based on cinchona alkaloids [14] was evaluated in chloroform. Gratifyingly, catalysts C1-C4 (Fig. 2) properly promoted the designed 1,6-conjugated addition/annulation/tautomerization domino process, and the desired 4-aryl-substituted chromene 3aa was generated in acceptable yield and promising optical purity (Table 1, entries 1-4). Further solvent screening revealed that chloroform was the favorable one in terms of reactivity and enantioselectivity (Table 1, entries 5-7 vs. entry 1). In order to further improve the enantioselectivity, the quinine-based squaramides C5 and C6 were then investigated. Notably, an obvious enhancement of enantiomeric purity was observed in the presence the squaramide **C5** [15] in comparison with the corresponding thiourea **C1** (Table 1, entry 8 vs. entry 1). The quinine-based squaramide C5 proved to be the preferred catalyst in view of reactivity and enantioselectivity (Table 1, entry 8 vs. entry 9). The catalyst loading exerted an impact on this cascade process to a certain extent. The model reaction proceeded equally efficiently in the presence of diminished amount of squaramide C5, albeit slightly prolonged time was required (Table 1, entries 10 and 11). In contrast, superior enantiomeric excess (90% ee) was achieved in the presence of 5 mol% of C5 (Table 1, entry 11 vs. entries 8 and 10).

Having established the optimal reaction conditions, we successively surveyed the generality and scope of this tandem 1,6-conjugate addition/annulation/tautomerization process of various *p*-OMs with a range of cyano-containing compounds (Table 2). For electron-rich *p*-OMs **1b-e**, the titled cascade process occurred smoothly and afforded the desired chromenes **3ab-ae** in excellent yields and satisfactory enantiopurities (Table 2, entries 2-5). The sterical hindrance of substituent on the aromatic ring had a slight influence on the enantioselectivity of this asymmetric

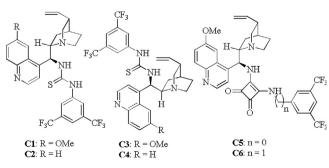
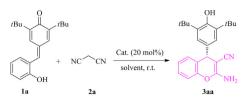


Fig. 2. The structures of bifunctional organocatalysts.

Table 1

Optimization of reaction conditions.^a.



Entry	Cat.	Solvent	Time (h)	Yield (%) ^b	ee (%) ^c
1	C1	CHCl ₃	24	92	74
2	C2	CHCl ₃	48	84	70
3	C3	CHCl ₃	24	92	-70
4	C4	CHCl ₃	48	80	-69
5	C1	Toluene	120	62	69
6	C1	EtOAc	24	60	34
7	C1	Ether	168	50	74
8	C5	CHCl ₃	15	95	83
9	C6	CHCl ₃	24	92	51
10 ^d	C5	CHCl ₃	30	99	89
11 ^e	C5	CHCl ₃	30	99	90

 $^{\rm a}$ Unless otherwise noted, the reaction was performed with 0.1 mmol of 1a, 0.1 mmol of malononitrile and 20 mol% of catalyst in 1 mL of solvent at room temperature (r.t.).

^b Isolated yields.

^c Determined by chiral HPLC analysis.

^d Performed with 10 mmol% of **C5**.

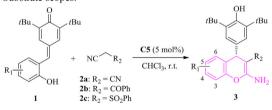
^e Performed with 5 mmol% of **C5**.

process. The sterically more bulky 3-ethoxyl-substituted **1b** offered **3ab** with relatively poorer enantioselectivity (86% *ee*) (Table 2, enty 2). This domino reaction was independent of the

Table 2

Substrate scopes ^a.

Substrate scopes.^a



Entry	R ₁	R ₂	Time (h)	3	Yield (%) ^b	ее (%) ^с
1	H (1a)	CN	30 (36) ^d	3aa	99 (92) ^d	90(90) ^d
2	3-EtO (1b)	CN	48	3ab	98	86
3	4-MeO (1c)	CN	48	3ac	92	96
4	4-Et ₂ N (1d)	CN	48	3ad	98	90
5	5-MeO (1e)	CN	48	3ae	93	96
6	4-Cl (1f)	CN	48	3af	99	90
7	5-Br (1 g)	CN	48	3ag	88	86
8	H (1a)	PhCO	24	3ba	92	99
9	3-EtO (1b)	PhCO	48	3bb	97	97
10	4-MeO (1c)	PhCO	48	3bc	95	98
11	5-Me (1 h)	PhCO	48	3bh	98	98
12	4-Cl (1f)	PhCO	48	3bf	92	98
13	5-Br (1 g)	PhCO	48	3bg	90	98
14	H (1a)	PhSO ₂	48	3ca	90	99
15	4-MeO (1c)	PhSO ₂	120	3cc	95	99
16	4-Cl (1f)	PhSO ₂	120	3cf	92	99

^a Unless otherwise noted, the reaction was performed with 0.1 mmol of **1**, 0.1 mmol of malononitrile and 5 mol% of **C5** in 1 mL of CHCl₂ at r.t. for due time

b Isolated yields.

Enantiomeric excesses were determined by chiral HPLC analysis.

 $^{\rm d}$ Performed with 1.1 g of **1a** (3.3 mmol) under the optimal conditions.

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