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Communication

Insight into the influence of high temperature annealing on the onset potential of Ti-doped hematite photoanodes for solar water splitting

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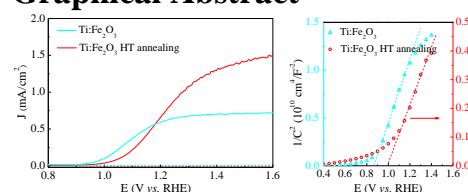
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Graphical Abstract



The unfavorable onset potential shift of Ti:Fe₂O₃ photoanode induced by high temperature post-annealing was studied. The underlying mechanism is the surface Ti/Fe atomic ratio increase, leading to adsorption capacity change of surface hydroxide ions and the positive shift of flatband potential (*i.e.*, theoretical onset potential).

ABSTRACT

For Ti-doped hematite photoanodes, high temperature annealing drastically increases the water oxidation plateau photocurrent, but also induces an anodic shift of onset potential by about 100 mV, thus hindering the performance under low applied bias. To the best of our knowledge, the effects of high temperature annealing on the onset potential have been rarely studied. Herein, both X-ray photoelectron spectroscopy (XPS) measurements and theoretical calculations indicated that the increase of surface Ti/Fe atomic ratio after high temperature annealing decreased the adsorption capacity of hydroxide ions on the hematite surface. Subsequently, the flatband potential (*i.e.*, the theoretical onset potential) of Ti doped hematite photoanodes positively shifted, which was supported by the Mott-Schottky measurements.

Keywords:

Onset potential

Hematite

Ti doping

High temperature annealing

Flatband potential

DFT calculation

Photoelectrochemical (PEC) water splitting has been considered as a promising technology for storing solar energy [1]. Since the start-up work of Fujishima and Honda on a TiO₂ photoanodes, many visible-light responsive semiconductor materials, such as α -Fe₂O₃, Ta₃N₅, and BiVO₄, have attracted intensive attention from researchers [2-4]. Among these materials, α -Fe₂O₃ is one of the most promising photoanode materials due to its high theoretical efficiency (about 16%), excellent photochemical stability, non-toxicity and low cost [5]. However, the PEC performance of α -Fe₂O₃ is hindered by some intrinsic drawbacks, including short hole diffusion length, poor electrical conductivity, severe surface states recombination and so on.

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