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Ning Liu, Jingyu Wang, Fengying Wang, Jun Liu



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The promoting effect of tantalum and antimony additives on deNO_x performance of Ce₃Ta₃SbO_x for NH₃-SCR reaction and DRIFT studies

Ning LIU^a, Jingyu WANG^a, Fengying WANG^{a*}, Jun LIU^b

^a School of Environment and Safety Engineering, North University of China, Taiyuan 030051, China

^b Key Laboratory of coal Science and Technology, Ministry of Education and Shanxi Province, Taiyuan University of Technology, Taiyuan 030024, China

Abstract

A superior Ce-Ta-Sb composite oxide catalyst prepared using homogeneous precipitation method exhibited excellent deNO_x efficiency and nearly 100% N₂ selectivity with broad operation temperature window and better resistance to higher space velocity, meanwhile strong resistance to H₂O and SO₂. This catalyst was systematically characterized using XRD, N₂ adsorption, SEM, TEM, XPS, ESR, Raman, H₂-TPR, NH₃-TPD and *in situ* DRIFTS. There exists a synergistic effect between Ce, Ta and Sb species. It is further indicated that the prominent deNO_x performance of the Ce₃Ta₃SbO_x catalyst is attributed to the elevated Ce³⁺ concentrations, abundant active surface oxygen species, as well as surface acidity and reducibility, which is closely linked with the synergistic effect between Ce, Sb and Ta species. Results from DRIFTS reveal that the reaction mechanism of surface-adsorbed NH₃ and NO_x species is linked to temperature, the L-H mechanism mainly occurs at low temperature (<300 °C), while the E-R mechanism occurs at high temperature(>300 °C). Overall, these findings indicate that Ce₃Ta₃SbO_x is promising for NO_x practical abatement.

Keywords: Ce-Ta-Sb mixed oxides; Homogeneous precipitation method; NH₃-SCR; Synergistic effect; NO_x adsorbed species; Rare earths

1. Introduction

The selective catalytic reduction (SCR) of NO with NH₃ is an efficient and economical technique for the removal of nitrogen oxides (NO_x) which are major environmental pollutants^[1-7]. To date, the commercial V₂O₅-WO₃(MoO₃)/TiO₂ catalysts are active within a temperature window ranging from 300 to 400 °C. Such commercial catalysts are placed upstream of the de-sulfurizer and electrostatic precipitator, where the catalysts suffered from the poisoning effect of SO₂ and dust^[8]. In addition, the poor deNO_x efficiency at high temperature limits their development. Therefore, the research and development of other catalysts with high activities is highly desirable.

It has been found that some transition-metal oxides have catalytic activity for the low-temperature NH₃-SCR of NO, among which manganese oxides exhibit the highest catalytic performance^[9,10]. Meanwhile, cerium oxides have already been widely used as promising candidates owing to their prominent oxygen storage capacity and redox cycle (Ce⁴⁺→Ce³⁺) in the NH₃-SCR process^[11-13]. Their acid-base properties, low cost and inherently environmentally friendly property also attract much attention^[14]. Recently, much research has been focused on the ceria-based NH₃-SCR catalysts, such as Ce-Ti^[15,16], Ce-W-Ti^[17], Ce-Mo-Ti^[18,19], Ce-Mo^[20], Ce-Nb^[21], Ce-Mn^[22-26], Ce-W^[27], Ce-Cu^[28], and Ce-Zr^[29]. Compared with single component CeO_x, these Ce-based composite oxide catalysts exhibit better deNO_x performance, N₂ selectivity, and the specific surface area. Besides that, Sb has been used as a promoter of some NH₃-SCR catalysts due to its better electric conduction. Zhang et al.^[30] studied a Ce_aTa_bO_x catalyst using co-precipitation method, and it exhibited excellent catalytic performance and good resistance to SO₂ and H₂O. Liu et al.^[31] studied a series of Ce-Sb binary oxide catalysts prepared by the citric acid method for the NH₃-SCR reaction, and proposed that the strong interaction between Sb and Ce species not only enhances the redox property of the catalyst but also increases the surface

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***Corresponding author:** WANG Fengying (**E-mail:** fengyingwang@126.com; **Tel.:** +86-351-3921560)

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