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³ New deep-red heteroleptic iridium complex with ⁴ 3-hexylthiophene for solution-processed organic light-emitting ₅ diodes emitting saturated red and high CRI white colors

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abstract

The exploitation of soluble and efficient deep-red phosphorescent emitters is of paramount 31 importance for solution-processed organic light-emitting diodes (OLEDs) applied in both 32 high-quality RGB displays and high color-rendering-index (CRI) solid-state lighting source. 33 In this work, a new deep-red heteroleptic iridium(III) complex, i.e. bis[2,5-di(4-hexylthio- 34 phen-2-yl)pyridine][acetylacetonate]iridium(III) [Ir(ht-5ht-py)₂(acac)], has been synthe-
sized and successfully used to fabricate solution-processed saturated red and white sized and successfully used to fabricate solution-processed saturated red and white organic light-emitting diodes (WOLEDs). The long alkyl side-chains of Ir(ht-5ht-py)₂(acac) 37 render its excellent solubility in common organic solvents and good compatibility with 38 common host materials. The solution-processed red OLED based on Ir(ht-5ht-py)₂(acac) 39
exhibited a decent external quantum efficiency of 8.2% and a power efficiency of 6.5 lm/ 40 exhibited a decent external quantum efficiency of 8.2% and a power efficiency of 6.5 lm/ W, with satisfactory Commission International de L'Eclairage (CIE) coordinates of (0.68, 41 0.31) for saturated red emission. Furthermore, the prepared multiple-phosphors-doped 42 WOLED with Ir(ht-5ht-py)₂(acac) as the red emitter showed an excellent high color render- 43 ing index (CRI) value of 89 as well as low color-correlated temperature (CCT) of 2331 K, 44 which can meet the call for physiologically-friendly indoor illumination. 45

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49 50 1. Introduction

 Solution-processed phosphorescent organic light- emitting diodes (OLEDs) have drawn great attention in the past decades since they hold great potential in large-area and cost-effective manufacturing of flat panel displays and solid-state lighting sources. The efficient phosphorescent emitters that can harvest both singlet and triplet excitons for radiative decay are of paramount importance for

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achieving high device performance $[1-4]$. Efficient and 58 solution-processible deep red phosphors and devices are 59 indispensible in various kinds of organic electrolumines- 60 cent (EL) devices such as red–green–blue (RGB) full color 61 displays and white lighting devices $[5]$. It not only functions 62 as a primary color for high color quality RGB displays, but 63 also plays an important role in determining the light- 64 emitting efficiency and color quality of white light-emitting 65 devices (WOLEDs) $[6-7]$. Moreover, the widely used physio- 66 logically-friendly "candle-like" white lighting devices also 67 require efficient deep red phosphors to achieve high color 68 rendering index (CRI) and low color correlated temperature 69 (CCT) merits $[5]$. Although great progress has been achieved 70

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 in solution-processed red OLEDs by using novel host mate- rials in combination with typical light-red dopants [\[9–11\]](#page--1-0), 73 deep-red phosphors and devices with CIE- $x \ge 0.67$ are still scarce up to now [\[12,13\]](#page--1-0).

 Several saturated red phosphors have been developed previously to improve the light-emitting efficiencies of the solution-processed red OLEDs. For example, Park et al. synthesized an efficient red phosphor [2-(9,9-diethyl-79 9H-fluoren-2-yl)-4-phenylquinoline \vert_2 iridium(III) picolinic 80 acid N-oxide $[(FPQ)_2Ir(pic-N-O)]$ and successfully applied it in solution-processed red OLEDs, achieving a luminous efficiency (LE) of 9.9 cd/A, a power efficiency (PE) of 3.9 lm/W and an external quantum efficiency (EQE) of 8.9%, with the CIE coordinates of (0.660, 0.338) [\[14\]](#page--1-0). We have reported an efficient saturated deep-red iridium den- drimer containing arylamine units as periphery dendron to 87 realize superior efficiency/color purity trade-offs [\[10\].](#page--1-0) The resultant device exhibited a high EQE of 11.65% and a PE of 3.65 lm/W with a CIE coordinate of (0.70, 0.30). In contrast to the dendritic route to develop solution-processible red 91 phosphors [\[15,16\],](#page--1-0) Chao et al. proposed to modify red 92 tris(1-phenylisoquinoline) iridium $[Ir(piq)_3]$ with long side 93 chain to improve its solubility $[17]$, thus facilitating the miscibility with common poly(vinylcarbazole) (PVK) host and leading to a distinct efficiency increase from 0.74 cd/ A to ca. 6 cd/A. In spite of these improvements, highly- soluble efficient deep-red phosphors remain scarce and limit the development of high color quality solution-99 processed red OLEDs and WOLEDs [\[18–20\].](#page--1-0) As an example, the CRI values of these WOLEDs were typically less than 80 101 and thus did not meet the practical requirement for white color lighting.

 Here, we synthesized a new solution-processible red heteroleptic iridium(III) complex, i.e. bis[2,5-di(4-hexylthio- phen-2-yl)pyridine][acetylacetonate]iridium(III) [Ir(ht-5ht-106 py \vert ₂(acac)], which showed pure red photoluminescence (PL) emission with peak located at 628 nm, corresponding to the CIE coordinates of (0.68, 0.31). The red phosphor $[Ir(ht-5ht-py)_{2}(acac)]$ exhibits good solubility in common organic solvents and excellent miscibility with host in solu- tion process. The corresponding solution-processed red OLED showed a promising EQE of 8% and a PE of 6.5 lm/W. Furthermore, by combining it with a blue phosphorescent 114 iridium(III) [bis(4,6-difluorophenyl)-pyridinato-N,C²]-pico- linate [FIrpic] [\[21\]](#page--1-0), green phosphorescent G0 [\[22\]](#page--1-0) and orange phosphorescent dopant containing 5-trifluorometh-117 yl-2-(9,9-diethylfluoren-2-yl) pyridine ligand $[Ir(Flpy-CF₃)₃]$ [\[20\]](#page--1-0) to prepare four-color white emissive layer, the result- ing WOLEDs show a high CRI of 89 and low CCT of 2331 K, making it a physiologically-friendly white lighting source [\[8\].](#page--1-0) This is among a few reports that solution-processed WOLEDs achieve such ideal warm white light emission.

123 2. Experimental

124 2.1. General information

 All chemicals and reagents were purchased from Aldrich Chemicals Company. The solvents were carefully dried and distilled with appropriate drying agents prior to use. Commercially available reagents were used without

further purification unless otherwise stated. The $\rm ^1H$ and $\rm ^{13}C$ 129 NMR spectra were recorded with Bruker Advanced 130 400 MHz NMR spectrometer. Thermal gravimetric analysis 131 (TGA) was performed on Perkin–Elmer-TGA 7 thermal 132 gravimetric analyzer under nitrogen flow at a heating rate 133 of $10 °C$ /min. MALDI-TOF-TOF was carried out using Bruk- 134 er autoflex III smart beam mass spectrometer. 135

2.2. Synthesis of ligand 136

2,5-Dibromopyridine (2.0 g, 7.55 mmol), 3-hexylthio- 137 phene boronic acid $(4.0 \text{ g}, 18.9 \text{ mmol})$ and $Pd(PPh_3)_4$ 138 (262.0 mg) were added into a mixture of THF (30 mL) 139 and 2 M Na_2CO_3 (8 mL) under N_2 atmosphere. The reaction 140 mixture was heated to 90 \degree C for 48 h with stirring. Then 141 the reaction mixture was cooled down to room tem- 142 perature and extracted with ethyl acetate (EA). The com- 143 bined organic phase was washed with water. The organic 144 phase was separated and dried over MgSO₄. The solvent 145 was removed under reduced pressure and the residue 146 was purified by column chromatography eluting with 147 $CH₂Cl₂/$ hexane. The product was obtained as a white crys- 148 talline solid. ¹H NMR (400 MHz, CDCl₃): δ (ppm) 8.80 (d, 149 $J = 1.9$ Hz, 1H,Ar), 7.84 (d, $J = 6.8$ Hz, 1H, Ar), 7.61 (d, 150 $J = 8.3$ Hz, 1H, Ar), 7.51 (d, $J = 4.5$ Hz, 1H, Ar), 7.20 (d, 151 $J = 1.0$ Hz, 1H, Ar), 6.99 (d, $J = 11.8$ Hz, 1H, Ar), 6.94 (s, 1H, 152 Ar), 2.63 (t, $J = 7.7$ Hz, 4H, hexyl), 1.72–1.59 (m, 4H, hexyl), 153 1.42–1.23 (m, 12H, hexyl), 0.90 (dd, $J = 7.0$, 5.6 Hz, 6H, 154 hexyl). ¹³C NMR (101 MHz, CDCl₃): δ (ppm) 151.31, 155 146.39, 144.70, 144.47, 143.93, 140.06, 133.26, 128.69, 156 125.96, 125.19, 122.48, 120.42, 118.55 (Ar), 31.70, 30.65, 157 30.58, 30.11, 29.01, 22.64, 14.12 (hexyl). 158

2.3. Preparation of Ir(ht-5ht-py)₂(acac) 159

The phosphorescent iridium complexes were prepared 160 according to the well-established two-step strategy from 161 the cyclometalation of IrCl₃.3H₂O with the corresponding 162 organic ligand to form, initially, the μ -chloro-bridged 163 dimer, followed by coordination of the acetylacetone 164 (acac) anion in the presence of $Na₂CO₃$ [\[23\].](#page--1-0) The reaction 165 mixture was extracted with $CH₂Cl₂$. The combined organic 166 phase was washed with water. The organic phase was 167 separated and dried over $MgSO₄$. The product was purified 168 by silica gel column chromatography with CH_2Cl_2/h exane 169 as an eluent and a dark red solid was obtained. 1 H NMR -170 $(400 \text{ MHz}, \text{ CDCl}_3)$: δ (ppm) 8.49 (d, J = 1.8 Hz, 2H, Ar), 171 7.68 (dd, J = 8.5, 2.1 Hz, 2H, Ar), 7.36 (d, J = 8.4 Hz, 2H, 172 Ar), 7.04 (t, $J = 5.4$ Hz, 2H, Ar), 6.81 (s, 2H, Ar), 6.69 (s, 2H, 173 Ar), 5.21 (d, $J = 4.1$ Hz, 1H, acac), 2.52 (dd, $J = 15.2$, 7.4 Hz, 174 4H, alkyl), 1.85–1.66 (m, 9H, alkyl), 1.66–1.44 (m, 12H, 175 alkyl), 1.34–1.09 (m, 28H, alkyl), 1.09–0.93 (m, 11H, alkyl), 176 0.93–0.72 (m, 19H, alkyl), 0.72–0.60 (m, 8H, alkyl). ¹³C 177 NMR (101 MHz, CDCl₃): δ (ppm) 183.33 (acac), 163.07, 178 150.44, 146.59, 144.04, 143.56, 138.46, 135.54, 132.75, 179 124.08, 123.4, 121.51, 118.85, 115.52, 99.86 (Ar), 30.64, 180 30.48, 29.52, 29.36, 29.09, 28.68, 28.41, 27.95, 27.51, 181 21.66, 21.56, 13.07, 13.01 (alkyl + acac). Calcd for $C_{55}H_{77}$ 182 IrN₂O₂S₄: C, 59.37; H, 6.43; N, 2.52, found: C, 59.33; H, 183 6.92; N, 2.52. MALDI-TOF–TOF-MS: m/z found 1112.40, 184 calcd 1112.66. 185

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