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New deep-red heteroleptic iridium complex with 3-hexylthiophene for solution-processed organic light-emitting diodes emitting saturated red and high CRI white colors

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ABSTRACT

The exploitation of soluble and efficient deep-red phosphorescent emitters is of paramount importance for solution-processed organic light-emitting diodes (OLEDs) applied in both high-quality RGB displays and high color-rendering-index (CRI) solid-state lighting source. In this work, a new deep-red heteroleptic iridium(III) complex, i.e. bis[2,5-di(4-hexylthio-phen-2-yl)pyridine][acetylacetonate]iridium(III) [Ir(ht-5ht-py)₂(acac)], has been synthesized and successfully used to fabricate solution-processed saturated red and white organic light-emitting diodes (WOLEDs). The long alkyl side-chains of Ir(ht-5ht-py)₂(acac) render its excellent solubility in common organic solvents and good compatibility with common host materials. The solution-processed red OLED based on Ir(ht-5ht-py)₂(acac) exhibited a decent external quantum efficiency of 8.2% and a power efficiency of 6.5 lm/W, with satisfactory Commission International de L'Eclairage (CIE) coordinates of (0.68, 0.31) for saturated red emission. Furthermore, the prepared multiple-phosphors-doped WOLED with Ir(ht-5ht-py)₂(acac) as the red emitter showed an excellent high color rendering index (CRI) value of 89 as well as low color-correlated temperature (CCT) of 2331 K, which can meet the call for physiologically-friendly indoor illumination.

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4950 **1. Introduction**

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Solution-processed phosphorescent organic lightemitting diodes (OLEDs) have drawn great attention in the past decades since they hold great potential in large-area and cost-effective manufacturing of flat panel displays and solid-state lighting sources. The efficient phosphorescent emitters that can harvest both singlet and triplet excitons for radiative decay are of paramount importance for

http://dx.doi.org/10.1016/j.orgel.2015.02.016 1566-1199/© 2015 Published by Elsevier B.V. achieving high device performance [1–4]. Efficient and solution-processible deep red phosphors and devices are indispensible in various kinds of organic electroluminescent (EL) devices such as red–green–blue (RGB) full color displays and white lighting devices [5]. It not only functions as a primary color for high color quality RGB displays, but also plays an important role in determining the light-emitting efficiency and color quality of white light-emitting devices (WOLEDs) [6–7]. Moreover, the widely used physio-logically-friendly "candle-like" white lighting devices also require efficient deep red phosphors to achieve high color rendering index (CRI) and low color correlated temperature (CCT) merits [5]. Although great progress has been achieved

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71 in solution-processed red OLEDs by using novel host mate-72 rials in combination with typical light-red dopants [9–11], 73 deep-red phosphors and devices with CIE- $x \ge 0.67$ are still 74 scarce up to now [12.13].

75 Several saturated red phosphors have been developed 76 previously to improve the light-emitting efficiencies of 77 the solution-processed red OLEDs. For example, Park 78 et al. synthesized an efficient red phosphor [2-(9,9-diethyl-79 9H-fluoren-2-yl)-4-phenylquinoline]₂ iridium(III) picolinic acid N-oxide [(FPQ)₂Ir(pic-N-O)] and successfully applied it 80 in solution-processed red OLEDs, achieving a luminous 81 efficiency (LE) of 9.9 cd/A, a power efficiency (PE) of 82 3.9 lm/W and an external quantum efficiency (EQE) of 83 84 8.9%, with the CIE coordinates of (0.660, 0.338) [14]. We have reported an efficient saturated deep-red iridium den-85 86 drimer containing arylamine units as periphery dendron to realize superior efficiency/color purity trade-offs [10]. The 87 88 resultant device exhibited a high EQE of 11.65% and a PE of 89 3.65 lm/W with a CIE coordinate of (0.70, 0.30). In contrast to the dendritic route to develop solution-processible red 90 phosphors [15,16], Chao et al. proposed to modify red 91 92 tris(1-phenylisoquinoline) iridium $[Ir(piq)_3]$ with long side 93 chain to improve its solubility [17], thus facilitating the 94 miscibility with common poly(vinylcarbazole) (PVK) host 95 and leading to a distinct efficiency increase from 0.74 cd/ A to ca. 6 cd/A. In spite of these improvements, highly-96 97 soluble efficient deep-red phosphors remain scarce and limit the development of high color quality solution-98 99 processed red OLEDs and WOLEDs [18-20]. As an example, the CRI values of these WOLEDs were typically less than 80 100 and thus did not meet the practical requirement for white 101 color lighting. 102

Here, we synthesized a new solution-processible red 103 104 heteroleptic iridium(III) complex, i.e. bis[2,5-di(4-hexylthiophen-2-yl)pyridine][acetylacetonate]iridium(III) [Ir(ht-5ht-105 106 py)₂(acac)], which showed pure red photoluminescence (PL) emission with peak located at 628 nm, corresponding 107 108 to the CIE coordinates of (0.68, 0.31). The red phosphor [Ir(ht-5ht-py)₂(acac)] exhibits good solubility in common 109 110 organic solvents and excellent miscibility with host in solution process. The corresponding solution-processed red 111 OLED showed a promising EQE of 8% and a PE of 6.5 lm/W. 112 113 Furthermore, by combining it with a blue phosphorescent iridium(III) [bis(4,6-difluorophenyl)-pyridinato-N,C²]-pico-114 linate [FIrpic] [21], green phosphorescent G0 [22] and 115 orange phosphorescent dopant containing 5-trifluorometh-116 yl-2-(9,9-diethylfluoren-2-yl)pyridine ligand [Ir(Flpy-CF₃)₃] 117 118 [20] to prepare four-color white emissive layer, the result-119 ing WOLEDs show a high CRI of 89 and low CCT of 2331 K, making it a physiologically-friendly white lighting source 120 [8]. This is among a few reports that solution-processed 121 WOLEDs achieve such ideal warm white light emission. 122

123 2. Experimental

2.1. General information 124

125 All chemicals and reagents were purchased from 126 Aldrich Chemicals Company. The solvents were carefully 127 dried and distilled with appropriate drying agents prior 128 to use. Commercially available reagents were used without further purification unless otherwise stated. The ¹H and ¹³C NMR spectra were recorded with Bruker Advanced 400 MHz NMR spectrometer. Thermal gravimetric analysis (TGA) was performed on Perkin-Elmer-TGA 7 thermal 132 gravimetric analyzer under nitrogen flow at a heating rate 133 of 10 °C/min. MALDI-TOF-TOF was carried out using Bruk-134 er autoflex III smart beam mass spectrometer. 135

2.2. Synthesis of ligand

2,5-Dibromopyridine (2.0 g, 7.55 mmol), 3-hexylthio-137 phene boronic acid (4.0 g, 18.9 mmol) and $Pd(PPh_3)_4$ 138 (262.0 mg) were added into a mixture of THF (30 mL) 139 and 2 M Na_2CO_3 (8 mL) under N_2 atmosphere. The reaction 140 mixture was heated to 90 °C for 48 h with stirring. Then 141 the reaction mixture was cooled down to room tem-142 perature and extracted with ethyl acetate (EA). The com-143 bined organic phase was washed with water. The organic 144 phase was separated and dried over MgSO₄. The solvent 145 was removed under reduced pressure and the residue 146 was purified by column chromatography eluting with 147 CH₂Cl₂/hexane. The product was obtained as a white crys-148 talline solid. ¹H NMR (400 MHz, CDCl₃): δ (ppm) 8.80 (d, 149 J = 1.9 Hz, 1H,Ar), 7.84 (d, J = 6.8 Hz, 1H, Ar), 7.61 (d, 150 I = 8.3 Hz, 1H, Ar), 7.51 (d, I = 4.5 Hz, 1H, Ar), 7.20 (d, 151 *J* = 1.0 Hz, 1H, Ar), 6.99 (d, *J* = 11.8 Hz, 1H, Ar), 6.94 (s, 1H, 152 Ar), 2.63 (t, *J* = 7.7 Hz, 4H, hexyl), 1.72–1.59 (m, 4H, hexyl), 153 1.42-1.23 (m, 12H, hexyl), 0.90 (dd, / = 7.0, 5.6 Hz, 6H, 154 hexyl). ¹³C NMR (101 MHz, CDCl₃): δ (ppm) 151.31, 155 146.39, 144.70, 144.47, 143.93, 140.06, 133.26, 128.69, 156 125.96, 125.19, 122.48, 120.42, 118.55 (Ar), 31.70, 30.65, 157 30.58, 30.11, 29.01, 22.64, 14.12 (hexyl). 158

2.3. Preparation of Ir(ht-5ht-py)₂(acac)

The phosphorescent iridium complexes were prepared 160 according to the well-established two-step strategy from 161 the cyclometalation of IrCl₃·3H₂O with the corresponding 162 organic ligand to form, initially, the µ-chloro-bridged 163 dimer, followed by coordination of the acetylacetone 164 (acac) anion in the presence of Na₂CO₃ [23]. The reaction 165 mixture was extracted with CH₂Cl₂. The combined organic 166 phase was washed with water. The organic phase was 167 separated and dried over MgSO₄. The product was purified 168 by silica gel column chromatography with CH₂Cl₂/hexane 169 as an eluent and a dark red solid was obtained. ¹H NMR 170 (400 MHz, CDCl₃): δ (ppm) 8.49 (d, *J* = 1.8 Hz, 2H, Ar), 171 7.68 (dd, / = 8.5, 2.1 Hz, 2H, Ar), 7.36 (d, / = 8.4 Hz, 2H, 172 Ar), 7.04 (t, J = 5.4 Hz, 2H, Ar), 6.81 (s, 2H, Ar), 6.69 (s, 2H, 173 Ar), 5.21 (d, J = 4.1 Hz, 1H, acac), 2.52 (dd, J = 15.2, 7.4 Hz, 174 4H, alkyl), 1.85-1.66 (m, 9H, alkyl), 1.66-1.44 (m, 12H, 175 alkyl), 1.34–1.09 (m, 28H, alkyl), 1.09–0.93 (m, 11H, alkyl), 176 0.93-0.72 (m, 19H, alkyl), 0.72-0.60 (m, 8H, alkyl). ¹³C 177 NMR (101 MHz, CDCl₃): δ (ppm) 183.33 (acac), 163.07, 178 150.44, 146.59, 144.04, 143.56, 138.46, 135.54, 132.75, 179 124.08, 123.4, 121.51, 118.85, 115.52, 99.86 (Ar), 30.64, 180 30.48, 29.52, 29.36, 29.09, 28.68, 28.41, 27.95, 27.51, 181 21.66, 21.56, 13.07, 13.01 (alkyl + acac). Calcd for C₅₅H₇₇ 182 IrN₂O₂S₄: C, 59.37; H, 6.43; N, 2.52, found: C, 59.33; H, 183 6.92; N, 2.52. MALDI-TOF-TOF-MS: m/z found 1112.40, 184 calcd 1112.66. 185

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