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Efficiencies of perovskite hybrid solar cells influenced by film thickness and morphology of $CH_3NH_3PbI_{3-x}Cl_x$ layer

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ABSTRACT

Perovskite hybrid solar cells (pero-HSCs) have been intensively investigated due to their promising photovoltaic performance. However, the correlations between the efficiencies of pero-HSCs and thin film thicknesses and morphologies of $CH_3NH_3PbI_{3-x}CI_x$ perovskite layers are rarely addressed. In this study, we report the correlation between the efficiencies of "planar heterojunction" (PHJ) pero-HSCs and the thin film thicknesses and morphologies of solution-processed CH₃NH₃PbI_{3-x}Cl_x perovskite layers. Investigation of absorption spectra, X-ray diffraction patterns, atomic force microscopy and scanning electron microscopy images of CH₃NH₃PbI₃ _xCl_x layers indicate that the efficiencies of PHJ pero-HSCs are dependent on the film thickness, as the thickness of $CH_3NH_3PbI_{3-x}Cl_x$ is less than 400 nm; whereas the efficiencies are significantly dependent on the film morphologies of $CH_3NH_3PbI_{3-x}Cl_x$ layers as the thickness is larger than 400 nm. Our studies provide a promising pathway for fabricating high efficiency PHJ pero-HSCs.

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1. Introduction 42

43 Perovskite materials ABX₃, where A is organic cuboctahedral, B is inorganic octahedral, and X is halogens, for example CH₃NH₃-44 45 PbX_3 (X = I, Br and Cl), have recently emerged as a remarkable light harvester for thin-film photovoltaics [1-5]. CH₃NH₃PbX₃ are 46 47 hybrid materials with an ideal bandgap for single junction solar cells [6] due to its broad absorption spectrum with high extinction 48 49 coefficients [7], large electron-hole diffusion length [8], good elec-50 trical transport properties [9]. Currently, over 17% of power conversion efficiency (PCE) has been reported from perovskites 51 hybrid solar cells (pero-HSCs) [10]. 52

The pero-HSCs based on CH₃NH₃PbX₃ thin films processed by 53 54 various methods show quite different device performances [11-14]. High PCEs was observed from pero-HSCs based on high quality 55 56 vapor-deposited CH₃NH₃PbX₃ thin films [15]. Moderate PCEs was 57 observed from pero-HSCs based on low guality solution-processed 58 $CH_3NH_3PbI_{3-x}Cl_x$ layer [16–19]. However, the correlations between 59 the efficiencies of pero-HSCs and the film thicknesses and morphologies of CH₃NH₃PbI_{3-x}Cl_x layers are rarely addressed. 60

In this study, we investigate the device performance of "planar 61 heterojunction" (PHJ) pero-HSCs correlated with the film thick-62

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http://dx.doi.org/10.1016/j.orgel.2015.02.023 1566-1199/© 2015 Published by Elsevier B.V. nesses and film morphologies of solution-processed CH₃NH₃PbI₃- $_{x}Cl_{x}$ layers by investigations of absorption spectra properties, X-ray diffraction (XRD) patterns, atomic force microscopy (AFM) and scanning electron microscopy (SEM) images of CH₃NH₃PbI_{3-x}Cl_x films. The PCEs ranging from 4.88% to 11.88% are observed from PHJ pero-HSCs with the thicknesses of $CH_3NH_3PbI_{3-x}Cl_x$ layers from 270 nm to 670 nm.

2. Experimental section

2.1. Materials

Poly(3,4-ethylenedioxythiophene):polystyrene sulfonate (PED-OT:PSS) and phenyl-C₆₁-butyric acid methyl ester (PC₆₁BM) were purchased from Clevious and 1-Material Inc., respectively and used as received without further treatment. Lead chloride (PbCl₂), anhydrous N,N-dimethylformamide (DMF), hydroiodic acid (99.99%), methylamince were purchased from Sigma-Aldrich and used as received without further purification.

2.2. $CH_3NH_3PbI_{3-x}Cl_x$ precursor preparation

Methyl ammonium iodide (CH₃NH₃I) was prepared following 80 the previous literatures [20,21]: hydroiodic acid (114 mmol, 81 15 mL, 57 wt%) and methylamine (140 mmol, 70 mL, 2.0 M in 82

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K. Wang et al. / Organic Electronics xxx (2015) xxx-xxx

83 methanol) were reacted at 0 °C with stirring under nitrogen atmo-84 sphere for 120 min. The resultant solution was evaporated to give a 85 white precipitate, and then the precipitate was washed with 86 diethyl ether for several times until the diethyl ether was com-87 pletely colorless. The white precipitate was dried under vacuum 88 for 48 h and used without further purification. And the precursor 89 solutions of $CH_3NH_3PbI_{3-x}Cl_x$ was prepared by dissolving $PbCl_2$ 90 and CH₃NH₃I (with a molar ratio of 1:3) in DMF with different concentrations of 0.44, 0.66, and 0.88 mol/L. 91

92 2.3. Thin film preparation

93 The $CH_3NH_3PbI_{3-x}Cl_x$ films were prepared by spin-casting the precursor solution on the top of PEDOT:PSS coated indium tin 94 95 oxide (ITO) substrate (for device fabrication and AFM measure-96 ment), quartz glass substrate (for absorption measurement) and 97 polvethylene terephthalate (PET) substrate (for XRD measure-98 ment), respectively, followed with thermal annealing at 90 °C for 2 h. The thickness of $CH_3NH_3PbI_{3-x}Cl_x$ perovskite films was con-99 trolled by using different precursor concentrations and spin-speed. 100

101 2.4. Thin film characterization

102 The absorption spectra of perovskite films were measured by HP 8453 UV-vis spectrophotometer. The SEM images were 103 104 obtained by using a field emission scanning electron microscope 105 (JEOL-7401). The AFM images were obtained by using a NanoScope 106 NS3A system (Digital Instrument) to observe the surface mor-107 phologies of various thin films. The film thicknesses were measured by Dektak 150 surface profilometer with scan rate 108 109 0.06 mm/s. XRD patterns of perovskite films coated on PET substrates were obtained by using a Bruker AXS Dimension D8 X-ray 110 111 System.

112 2.5. PHJ pero-HSCs fabrication

113 All PHI pero-HSCs were fabricated on pre-cleaned ITO coated 114 glass substrates. After ITO glass substrates were treated with UV-o-115 zone for 20 min under an ambient atmosphere, a \sim 40 nm thick 116 film of PEDOT:PSS was spin-coated on the top of ITO glass sub-117 strates, followed with thermal annealing at 150 °C for 10 min. After that, the CH₃NH₃PbI_{3-x}Cl_x thin films with different thickness were 118 spin-coated on the top of PEDOT:PSS from CH₃NH₃PbI_{3-x}Cl_x pre-119 120 cursor solution, followed with thermal annealing at 90 °C for 2 h. The thickness was controlled by using $CH_3NH_3PbI_{3-x}Cl_x$ precursor 121 122 solution with different concentrations (0.44 mol/L, 0.66 mol/L, 123 and 0.88 mol/L) and different spin-rates (2000 rpm, 2500 rpm and 3000 rpm). The thicknesses of $CH_3NH_3PbI_{3-x}Cl_x$ thin films are 124 ranged from 270 nm to 668 nm. Then \sim 200 nm thick PC₆₁BM 125 126 was deposited from the chlorobenzene solution. Finally, ~100 nm 127 thick Aluminum (Al) was thermally deposited on the top of PC₆₁BM layer in the vacuum with a base pressure of 6×10^{-6} mbar through 128 a shadow mask. The device area was measured to be 0.16 cm^2 . 129

130 2.6. PHJ pero-HSCs characterization

131 The current densities versus voltage (I-V) characteristics of pero-HSCs were obtained by using a Keithley model 2400 source 132 measure unit. A Newport Air Mass 1.5 Global (AM1.5G) full-spec-133 134 trum solar simulator was applied as the light source. The light intensity was 100 mW cm⁻², which was calibrated by utilizing a 135 136 monosilicon detector (with a KG-5 visible color filter) of National 137 Renewable Energy Laboratory to reduce the spectral mismatch. 138 The incident photon-to-electron conversion efficiency (IPCE) was 139 measured through the IPCE measurement setup in use at ESTI for 140 cells and mini-modules: a 300 W steady-state xenon lamp provides the source light; up to 64 filters (8–20 nm width, range from 141 300 to 1200 nm) are available on four filter-wheels to produce the 142 monochromatic input, which is chopped at 75 Hz, superimposed 143 on the bias light and measured via the usual lock-in technique; 144 bias light is necessary to put the device under examination close 145 the operating irradiance condition. After collecting the IPCE data, 146 the software also integrates the data with the AM1.5G spectrum 147 and gives the calculated *J*_{SC} value, which is helpful for checking 148 the accuracy of the measurement. 149

3. Results and discussion

The PHJ pero-HSCs with a p-i-n device structure of ITO/PED-151 OT:PSS/CH₃NH₃PbI_{3-x}Cl_x/PC₆₁BM/Al is applied to investigate the 152 device performance influenced by film thicknesses and film mor-153 phologies of $CH_3NH_3PbI_{3-x}Cl_x$ layers. In the p-i-n pero-HSCs, the 154 $CH_3NH_3PbI_{3-x}Cl_x$ film being the intrinsic semiconductor acts as 155 the light absorber, PEDOT:PSS being the p-type layer acts as the 156 hole transport layer, and PC₆₁BM being the n-type layer acts as 157 the electron transport layer, respectively. The I-V characteristics 158 of PHJ pero-HSCs are shown in Fig. 1a. Table 1 summarizes the 159 device performance parameters of all PHJ pero-HSCs. For each 160 thickness, over 50 devices were tested for the statistics. The confi-161 dence interval is 80% for standard errors. All PHJ pero-HSCs possess 162 a similar fill factor (FF) of \sim 0.61, and an open-circuit voltage (V_{OC}) 163 of ~0.93 V, but with significantly different short-circuit current 164 density (J_{SC}) , consequently significantly different PCEs. As the 165 thicknesses of $CH_3NH_3PbI_{3-x}Cl_x$ layers are gradually increased from 166 273 nm to 575 nm, and then further increased to 668 nm, the J_{SC} 167 values are gradually increased from 8.45 mA/cm² to 19.79 mA/ 168 cm², and then decreased to 17.26 mA/cm², respectively; corre-169 spondingly, the PCEs are increased from 4.88% to 11.88%, and then 170 decreased to 9.42%, respectively. The decreased J_{SC} in thicker CH₃-171 $NH_3PbI_{3-x}Cl_x$ films are likely due to the inferior surface quality of 172 $CH_3NH_3PbI_{3-x}Cl_x$ films [22], which will be addressed late (Figs. 5) 173 and 7). 174

To gain insight into the influence of film thickness on *Isc*, the 175 incident IPCE spectra of pero-HSCs were measured and displayed 176 in Fig. 1b. The IPCE specifies the ratio of extracted electrons to inci-177 dent photons at a given wavelength. The IPCE spectra of pero-HSCs 178 are in good agreement with the absorption spectra of CH₃NH₃-179 $PbI_{3-x}Cl_x$ with different film thicknesses (Fig. 3). The IPCE spectra 180 of all pero-HSCs show a wide photoresponse from 300 nm to 181 800 nm, which is consistent to absorption spectra of CH₃NH₃PbI₃-182 _xCl_x. As the film thicknesses are gradually increased from 270 nm 183 to 607 nm, the photoresponses of pero-HSCs are gradually 184 enhanced from \sim 40% to \sim 80%, correspondingly. In specific, as the 185 thicknesses of $CH_3NH_3PbI_{3-x}Cl_x$ are gradually increased from 186 270 nm to 607 nm, the photoresponses in the region from 187 550 nm to 800 nm are dramatically enhanced while the photore-188 sponses at the wavelength below 500 nm are moderately 189 enhanced. The different shapes of IPCE spectra are probably due 190 to interactions among photocontribution from PC₆₁BM and from 191 $CH_3NH_3PbI_{3-x}Cl_x$ of different thicknesses. It has been reported that 192 a small amount of photoinduced excitons can also be generated in 193 PC₆₁BM at the wavelength below 500 nm [23]. When the thickness 194 of $CH_3NH_3PbI_{3-x}Cl_x$ is small, the photocontribution from $PC_{61}BM$ is 195 more obvious than that from $CH_3NH_3PbI_{3-x}Cl_x$; while the thickness 196 $CH_3NH_3PbI_{3-x}Cl_x$ is large, the photocontribution from $PC_{61}BM$ is 197 trivial compared with that from CH₃NH₃PbI_{3-x}Cl_x [24,25,23]. The 198 estimated J_{SC} from pero-HSCs with different thicknesses of CH₃-199 $NH_3PbI_{3-x}CI_x$ layers are 7.24 mA/cm² (273 nm), 8.59 mA/cm² (296 nm), 11.82 mA/cm² (365 nm), 14.05 mA/cm² (387 nm), 200 201 15.88 mA/cm² (401 nm), 17.21 mA/cm² (468 nm), 18.13 mA/cm² 202 (575 nm), 18.45 mA/cm² (607 nm), 15.76 mA/cm² (668 nm). These 203

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