



Determination of fracture toughness of propylene polymers at different operating temperatures



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ARTICLE INFO

Article history:

Received 27 May 2013

Received in revised form 17 February 2014

Accepted 25 April 2014

Available online 17 May 2014

Keywords:

Propylene polymers

Fracture Mechanics parameters

Temperature

Molecular weight

Ethylene content

ABSTRACT

The influence of the temperature on the fracture parameters of three types of polypropylenes under static loading conditions has been analyzed. Single-phase polypropylene homopolymer and random ethylene–propylene copolymer were tested slightly above their glass transition temperatures (within the ductile-to-brittle transition region), and well below and above them; heterophasic ethylene–propylene block copolymers were tested on the range of temperatures from above the glass transition temperature of propylene matrix to below the glass transition temperature of the elastomer phase. Attending to the mechanical response, suitable Fracture Mechanics approaches were used. Fracture toughness increased with temperature for single-phase polymers but block copolymers were not temperature sensitive for the range of temperatures above the glass transition temperature of the elastomeric phase despite their variation on the mechanical properties.

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1. Introduction

Low manufacturing costs, ease of recycling and possibility of tailoring properties are some of the reasons why isotactic polypropylene, PP, has been enjoying the fastest growth in consumption since its discovery [1]. There are three typical types of PP: homopolymers, random copolymers and block (or impact/heterophasic) copolymers. The copolymers are usually of ethylene and butane. Random copolymers usually contain up to 6 wt% of ethylene or other comonomers inserted randomly within the chain, reducing the crystallinity and the melting point by introducing irregularities into the chain. They are used where clarity, lower melting point or lower modulus is desirable. The low content in ethylene provides a continuous and monophase structure with one single glass transition temperature, T_g , slightly below that of the PP homopolymers because of the presence of ethylene. Block copolymers, also known as impact or heterophasic copolymers, contain up to 40% ethylene–propylene rubber (EPR), uniformly dispersed within the homopolymer matrix [1–3]. They show two distinct glass transition temperatures, associated with the PP matrix, $\sim 23^\circ\text{C}$, and with the elastomeric phase, $\sim -50^\circ\text{C}$. As implied in the name, those copolymers are used where impact strength is important, especially at low temperatures [1]. The toughening mechanism is related to the generation of many small crazes, which absorb significant energy, and the interruption of a propagating fracture in the matrix [4–11]. This mechanism operates above the glass transition temperature of the dispersed phase, and its effectiveness is determined by one side, the nature of the elastomeric phase, which defines the deformation

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Nomenclature

a_0	initial crack length
a	crack length
ASTM	American Society for Testing and Materials
B	specimen thickness
DMTA	Dynamic Mechanical Thermal Analysis
DSC	Differential Scanning Calorimetry
E	Young's modulus
EPBC	ethylene–propylene block copolymer
EPR	ethylene–propylene random copolymer
ESIS	European Structural Integrity Society
EWf	Essential Work of Fracture
$F(J)$	cumulative probability
G_{IC}	critical energy release rate
GPC	Gel Permeation Chromatography
J	J -integral
J_{Cmin}	lowest fracture toughness
J_{Cmax}	highest fracture toughness
J_{IC}	fracture toughness at crack initiation
J_{med}	mean fracture toughness
J_0	threshold toughness
$J_{0.2}$	crack resistance at 0.2 mm of the total crack growth
$K_{I_{max}}$	critical stress intensity factor
LEFM	Linear Elastic Fracture Mechanics
m	shape parameter
M_n	number average molecular weight
M_w	weight average molecular weight
M_z	z-average molecular weight
NLEFM	Non Linear Elastic Fracture Mechanics
NMR	Nuclear Magnetic Resonance
P	cumulative probability
PP	polypropylene homopolymer
SENB	single edge notched bend specimens abbreviation
T_g	glass transition temperature
W	specimen width
Δ_a	crack extension
σ_Y	yield stress
χ	degree of crystallinity
3P-W	three-parameter Weibull model

behaviour of the dispersed particles and their ability to absorb and redistribute the energy; and by the other side, the concentration and particle size, which establishes the probability of both initiating crazes and intercepting the fracture line.

The influence of the structural and morphological parameters (content, size and interparticle distance of the rubber modifiers [4–14], molecular weight, tacticity, crystallinity index, shape, size and distribution of crystalline regions of the PP matrix) on the fracture behaviour of PPs has been widely analyzed specially under high loading rates and at low temperature [2–4,6–17,18–25]. The main reason is that these polymeric materials, especially the block copolymers, exhibit a wide range of properties and structures with a great performance in the high strain field.

Apart from the typical high strain rate applications, the propylene polymers are also particularly suitable for hot and cold water distribution systems, heating and air conditioning for residential and industrial networks and transport of agricultural and industrial fluids as well as food substances. In service and in most of these applications, the polymers are to bear temperatures that go from $-30\text{ }^{\circ}\text{C}$ to even $100\text{ }^{\circ}\text{C}$ under static loading conditions. Previous works by Pooler [26] and Kobbe [27] illustrated vastly different behaviours over a practical in-service range of temperatures for HDPE and polypropylene formulations under compressive, tensile and flexural testing configurations, respectively. However, under quasi-static conditions there is lack of systematic studies in the literature of the evolution of the fracture parameters of bulk PPs with the temperature.

Santarelli and Frontini showed that it was very difficult to determine the fracture toughness of PP homopolymers at room temperature and under static loading conditions [28]. The fracture toughness data were highly scattered and showed a strong sensitivity to notch [19–31] and geometry dimensions [4,5,28] and strain rate [5,12,15,16,30,31]. In addition, some

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