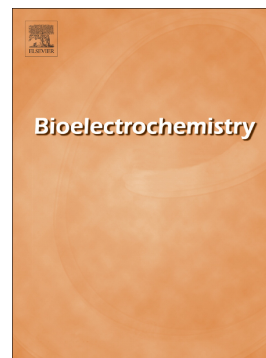


Accepted Manuscript

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PII: S1567-5394(17)30388-2
DOI: doi:[10.1016/j.bioelechem.2018.03.017](https://doi.org/10.1016/j.bioelechem.2018.03.017)
Reference: BIOJEC 7136
To appear in: *Bioelectrochemistry*
Received date: 28 July 2017
Revised date: 28 March 2018
Accepted date: 28 March 2018

Please cite this article as: Myreisa Morales-Cruz, Marjorie López-Nieves, Roberto Morales-Hernández, Gian C. Rivera-Crespo, Gary A. Toranzos, Ileana González-González, Carlos R. Cabrera , Proteus vulgaris - Pt electrode system for urea to nitrogen conversion in synthetic urine. The address for the corresponding author was captured as affiliation for all authors. Please check if appropriate. Biojec(2017), doi:[10.1016/j.bioelechem.2018.03.017](https://doi.org/10.1016/j.bioelechem.2018.03.017)

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Proteus vulgaris - Pt Electrode System for Urea to Nitrogen Conversion in Synthetic Urine

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Abstract: One of the most challenging problems when trying to recycle urine for different purposes is the removal of urea. In this project we studied an ureolysis system using the bacterium *Proteus vulgaris* for the transformation of urea to ammonia and its subsequent oxidation to nitrogen at a Pt working electrode. Our system was tested under different pH, microbial reaction times, and urea and bacteria concentrations. Our results indicate that a pH 8 is optimal for the combined *Proteus vulgaris* urease activity and the ammonia oxidation reaction at a Pt electrode. The reaction time and concentration dependence on the ammonia oxidation reaction current densities was also studied. Results showed limited ammonia oxidation under high urea concentrations in $\sim 2.5 \times 10^9$ cfu/mL *Proteus vulgaris* in synthetic urine.

Keywords: ammonia oxidation reaction; *Proteus vulgaris*; ureolysis; urine; urea

1. Introduction

In 2015, the United Nations reported that more than 40% of the global population is affected by water scarcity and that this number will continue to increase. For this reason, water recovery from wastewater is an essential area ~~problem~~ to study ~~and that must be solved in that~~ could be of help to the modern world [1]. One of the major sources of nutrients in wastewaters is urine [2], (80% nitrogen, 50% phosphorus and 9% potassium) [3]. Hence, human waste treatment could benefit tremendously by separating urine in new toilet systems that produces electricity and clean water. This would bring the added advantages of reducing the amount of water used during flushing cycles, allowing for high nutrient recycling, ~~and would~~ therefore reducing contamination in bodies of water ~~bodies~~ [3, 4]. Apart from urine recycling to obtain water, the removal of urea is important in order to protect the environment. Urea is a key pollutant in agricultural waste due to its stimulation of algal growth, increasing the pH when it is converted to ammonia [5, 6].

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