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Deflagrations of localised homogeneous and inhomogeneous hydrogen-air mixtures in enclosures

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ABSTRACT

Two original models for use as novel tools for the design of hydrogen-air deflagration mitigation systems for equipment and enclosures are presented. The first model describes deflagrations of localised hydrogen-air mixtures in a closed space such as a pressure vessel or a well-sealed building while the second model defines safety requirements for vented deflagrations of localised mixtures in an enclosure. Examples of localised mixtures include 'pockets' of gas within an enclosure as well as stratified gas distributions which are especially relevant to hydrogen releases. The thermodynamic model for closed spaces is validated against experiments available from the literature. This model is used to estimate the maximum hydrogen inventory in a closed space assuming the closed space can withstand a maximum overpressure of 10 kPa without damage (this is typical of many civil structures). The upper limit for hydrogen inventory in a confined space to prevent damage is found to be equivalent to 7.9% of the closed space being filled with 4% hydrogen. If the hydrogen inventory in a closed space is above this upper limit then the explosion has to be mitigated by the venting technique. For the first time an engineering correlation is presented that accounts for the phenomena affecting the overpressure from localised vented deflagrations, i.e. the turbulence generated by the flame front itself, the preferential diffusion in stretched flames, the fractal behaviour of the turbulent flame front surface, the initial flow turbulence in unburnt mixture, and the increase of the flame surface area due to the shape of an enclosure. Validation of the new vented deflagration model developed at Ulster has been carried out against 25 experiments with lean stratified hydrogen-air mixtures performed by the Health and Safety Executive (UK) and Karlsruhe Institute of Technology (Germany).

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Introduction

Deflagration of inhomogeneous, e.g. stratified, fuel-air mixtures in an enclosure, where fuel concentration varies

mainly in vertical direction, is a realistic accident scenario for both industrial environment and domestic premises [1,2]. This scenario may result from a slow release of heavier or lighter than air flammable gas, or a spill of

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Nomen	clature	Φ^{*}	volumetric fraction of the fastest burning fuel-air
А	fraction of vent area occupied by combustion		mixture in enclosure
	products, or coefficient in equation (49)	γ	adiabatic index, $\gamma = c_p/c_v$
В	coefficient in equation (49)	φ	volumetric fraction of fuel in localised fuel-air
Br _t	turbulent Bradley number, $Br_t =$		mixture, and number of moles in equations
·	$(\sqrt{E_{\rm c}}/\alpha \mu E_{\rm c})((36\pi_{\rm o})^{1/3}EV^{2/3}S_{\rm c}(E_{\rm c}-1))$		(9)-(11)
Dr*	$(\nabla E_i / \mu E_{ui})((500)) = \nabla E_{ui} (E_i - E_i)$	μ	discharge coefficient
DI _t		π	non-dimensional pressure $\pi = p/p_i$
	(00)	π_0	"Pi" number, 3.14159
Cp	and (12) (Ure/K)	Ξ/μ	deflagration-outflow interaction (DOI) number, $\Xi/$
	and (12) $(J/Kg/K)$		$\mu = \Xi_{\rm K} \Xi_{\rm LP} \Xi_{\rm FR} \Xi_{\rm u'} \Xi_{\rm AR} \Xi_{\rm O}$
Cυ	(I/m a)/V)	Ψ	empirical coefficient
	(1/1101/K)	Ξ_{K}	wrinkling factor to account for the turbulence
C _{ui}	speed of sound in unburnt mixture (m/s), $c_{ui} = \sqrt{\frac{1}{2}}$		generated by the flame front itself
_	$\sqrt{\gamma RT/M}$	Ξ_{LP}	wrinkling factor to account for the leading point
D	fractal dimension		flame acceleration mechanism
е	energy per unit mass (J/kg)	Ξ_{FR}	wrinkling factor to account for the fractal increase
E_i	combustion products expansion coefficient,		of the flame surface area
	$E_i = M_{ui} T_{bi} / M_{bi} T_{ui}$	Ξu,	wrinkling factor to account for the initial flow
F	vent area (m²)		turbulence in unburnt mixture
G	mass flow rate (kg/s)	Ξ_{AR}	wrinkling factor to account for the aspect ratio of
Н	height (m), or enthalpy (J)		the enclosure
ΔH_c	heat of combustion (J/mol)	Ξo	wrinkling factor to account for the presence of
$\Delta h_{ m c}$	heat of combustion (J/kg)		obstacles
h	enthalpy per unit mass (J/kg)	ρ	density (kg/m ³), $\rho = pM/(RT)$
L	length (m)	σ	relative density, $\sigma = ho / ho_{ m i}$
М	molecular mass (g/mol)	au	non-dimensional time, $ au = tS_{u_i}/c_{u_i}$
m	mass (kg), or burning velocity temperature index	ω	volume fraction in enclosure
	(Table 7)		
		Subari	eta .
n	mass fraction, or burning velocity baric index	Subscrip	pts
n	(Table 7) mass fraction, or burning velocity baric index (Table 7)	Subscrij air b	pts air burnt minture
n p	(Table 7) mass fraction, or burning velocity baric index (Table 7) pressure (Pa abs)	Subscrij air b	pts air burnt mixture
n p q	(Table 7) mass fraction, or burning velocity baric index (Table 7) pressure (Pa abs) heat per unit mass (J/kg)	Subscrij air b corr	pts air burnt mixture correlation value
n p q R	(Table 7) mass fraction, or burning velocity baric index (Table 7) pressure (Pa abs) heat per unit mass (J/kg) flame radius (m), or universal gas constant,	Subscrip air b corr exp	pts air burnt mixture correlation value experimental value
n p q R	(Table 7) mass fraction, or burning velocity baric index (Table 7) pressure (Pa abs) heat per unit mass (J/kg) flame radius (m), or universal gas constant, R = 8314 (J/K/kmol)	Subscrip air b corr exp f	pts air burnt mixture correlation value experimental value fuel
n p q R R#	<pre>(Table 7) mass fraction, or burning velocity baric index (Table 7) pressure (Pa abs) heat per unit mass (J/kg) flame radius (m), or universal gas constant, R = 8314 (J/K/kmol) non-dimensional venting parameter, R[#] =</pre>	Subscrip air b corr exp f H ₂	pts air burnt mixture correlation value experimental value fuel hydrogen
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n p q R R [#] R ₀	(Table 7) mass fraction, or burning velocity baric index (Table 7) pressure (Pa abs) heat per unit mass (J/kg) flame radius (m), or universal gas constant, R = 8314 (J/K/kmol) non-dimensional venting parameter, $R^{\#} =$ $\{ (2\gamma)/(\gamma - 1) \pi\sigma [(1/\pi)^{2/\gamma} - (1/\pi)^{1+1/\gamma}] \}^{1/2}$ critical radius for transition from laminar to fully turbulent flame propagation regime (m) turbulent burning velocity (m/s)	Subscrip air b corr exp f H ₂ i MAX MIN m	air air burnt mixture correlation value experimental value fuel hydrogen initial conditions maximum minimum flammable mixture
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n p q R R # R ₀ S _t S _u T t	(Table 7) mass fraction, or burning velocity baric index (Table 7) pressure (Pa abs) heat per unit mass (J/kg) flame radius (m), or universal gas constant, R = 8314 (J/K/kmol) non-dimensional venting parameter, $R^{\#} =$ { $(2\gamma)/(\gamma - 1) \pi \sigma [(1/\pi)^{2/\gamma} - (1/\pi)^{1+1/\gamma}] \}^{1/2}$ critical radius for transition from laminar to fully turbulent flame propagation regime (m) turbulent burning velocity (m/s) laminar burning velocity (m/s) temperature (K) time (s)	Subscrip air b corr exp f H ₂ i MAX MIN m t u 0	air air burnt mixture correlation value experimental value fuel hydrogen initial conditions maximum minimum flammable mixture turbulent unburnt mixture initial thermodynamic state (before combustion)
n p q R R # R ₀ S _t S _u T t	(Table 7) mass fraction, or burning velocity baric index (Table 7) pressure (Pa abs) heat per unit mass (J/kg) flame radius (m), or universal gas constant, R = 8314 (J/K/kmol) non-dimensional venting parameter, $R^{\#} =$ { $(2\gamma)/(\gamma - 1) \pi \sigma [(1/\pi)^{2/\gamma} - (1/\pi)^{1+1/\gamma}]$ } ^{1/2} critical radius for transition from laminar to fully turbulent flame propagation regime (m) turbulent burning velocity (m/s) laminar burning velocity (m/s) temperature (K) time (s) internal energy per unit mass (J/kg)	Subscrip air b corr exp f H ₂ i MAX MIN m t u 0 1	air air burnt mixture correlation value experimental value fuel hydrogen initial conditions maximum minimum flammable mixture turbulent unburnt mixture initial thermodynamic state (before combustion) thermodynamic state 1 (on completion of
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n p q R R [#] R ₀ S _t S _u T t u V v w W w x _{H2} Z	(rable 7) mass fraction, or burning velocity baric index (Table 7) pressure (Pa abs) heat per unit mass (J/kg) flame radius (m), or universal gas constant, R = 8314 (J/K/kmol) non-dimensional venting parameter, $R^{\#} =$ { $(2\gamma)/(\gamma - 1) \pi \sigma [(1/\pi)^{2/\gamma} - (1/\pi)^{1+1/\gamma}] \}^{1/2}$ critical radius for transition from laminar to fully turbulent flame propagation regime (m) turbulent burning velocity (m/s) laminar burning velocity (m/s) laminar burning velocity (m/s) temperature (K) time (s) internal energy per unit mass (J/kg) volume (m ³) specific volume (m ³ /kg) width (m), or non-dimensional venting parameter, $W = \mu Fc_{ui}/[(36 \pi_0)^{1/3} \sqrt{\gamma_u} V^{2/3} S_{ui}]$ mechanical work of gas per unit mass (J/kg) volumetric fraction of hydrogen in entire enclosure, $x_{H_2} = \phi \cdot \Phi$ non-dimensional number, $Z = \gamma_b [E_i - (\gamma_u/\gamma_b)(\gamma_b - 1)/(\gamma_u - 1)] \pi^{(1-\gamma_u)/\gamma_u} + (\gamma_u - \gamma_b)/(\gamma_u - 1)$	Subscrip air b corr exp f H ₂ i MAX MIN m t u 0 1 2 Supersc ' Acronyr BOS CFD DOI	<pre>pts air air air air burnt mixture correlation value experimental value fuel hydrogen initial conditions maximum minimum flammable mixture turbulent unburnt mixture initial thermodynamic state (before combustion) thermodynamic state 1 (on completion of adiabatic isochoric combustion) thermodynamic state 2 (on completion of adiabatic expansion) ripts value in flammable mixture ms Background Oriented Schlieren Computational Fluid Dynamics Deflagration-Outflow Interaction</pre>
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n p q R R T t u V v W w x _{H2} Z Greek Φ	(rable 7) mass fraction, or burning velocity baric index (Table 7) pressure (Pa abs) heat per unit mass (J/kg) flame radius (m), or universal gas constant, R = 8314 (J/K/kmol) non-dimensional venting parameter, $R^{\#} =$ $\{ (2\gamma)/(\gamma - 1) \pi \sigma [(1/\pi)^{2/\gamma} - (1/\pi)^{1+1/\gamma}] \}^{1/2}$ critical radius for transition from laminar to fully turbulent flame propagation regime (m) turbulent burning velocity (m/s) laminar burning velocity (m/s) laminar burning velocity (m/s) temperature (K) time (s) internal energy per unit mass (J/kg) volume (m ³) specific volume (m ³ /kg) width (m), or non-dimensional venting parameter, $W = \mu Fc_{ui}/[(36 \pi_0)^{1/3} \sqrt{\gamma_u} V^{2/3} S_{ui}]$ mechanical work of gas per unit mass (J/kg) volumetric fraction of hydrogen in entire enclosure, $x_{H_2} = \phi \cdot \Phi$ non-dimensional number, $Z = \gamma_b [E_i - (\gamma_u/\gamma_b)(\gamma_b - 1)/(\gamma_u - 1)] \pi^{(1-\gamma_u)/\gamma_u} + (\gamma_u - \gamma_b)/(\gamma_u - 1)$ volumetric fraction of localised flammable fuel-air mixture in anchorume	Subscrip air b corr exp f H ₂ i MAX MIN m t u 0 1 2 Supersc ' Acronyr BOS CFD DOI LES RHS SGS	<pre>pts air air air burnt mixture correlation value experimental value fuel hydrogen initial conditions maximum minimum flammable mixture turbulent unburnt mixture initial thermodynamic state (before combustion) thermodynamic state 1 (on completion of adiabatic isochoric combustion) thermodynamic state 2 (on completion of adiabatic expansion) rtipts value in flammable mixture ms Background Oriented Schlieren Computational Fluid Dynamics Deflagration-Outflow Interaction Large Eddy Simulation Right-Hand Side Sub-Grid Scale</pre>

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