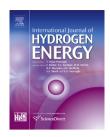
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## Inhibitory effect of chloroform on fermentative hydrogen and methane production from lipidextracted microalgae

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#### ABSTRACT

To improve the sustainability of microalgae as a bioenergy feedstock, lipid-extracted microalgae (LEM) are often further treated by anaerobic digestion (AD). However, the residual chloroform used for extracting lipids as a solvent could inhibit this process, an aspect that has not been studied to date. In this study, the inhibitory effect of chloroform on H<sub>2</sub> and CH<sub>4</sub> production was investigated by performing batch tests. To prepare the feedstock, Chlorella vulgaris was ultrasonicated and the supernatant was discarded after centrifugation. In case of H<sub>2</sub> production, it was found that the H<sub>2</sub> yield fell to almost half that of the control (15.6 mL  $H_2/g$  COD<sub>added</sub>) at 100 mg CHCl<sub>3</sub>/L. The reason for the decrease of the  $H_2$  yield with the increase of chloroform level was due to the change of metabolites from acetate and butyrate to lactate via a non-hydrogenic reaction. In comparison with H<sub>2</sub> production, a much more severe inhibitory effect of chloroform on CH4 production was observed. The inhibitor concentration (IC<sub>30, 60, and 90</sub>) on H<sub>2</sub> production was 138, 319, and 622 mg CHCl<sub>3</sub>/L, respectively, while concentrations of 15, 37, and 86 mg CHCl<sub>3</sub>/L were obtained on CH<sub>4</sub> production. When the chloroform concentration was  $\geq$ 25 mg/L on CH<sub>4</sub> production, more than 2 g COD/L of organic acids remained, resulting in a decrease of CH<sub>4</sub> yield. These findings indicate that the residual chloroform in LEM should be seriously considered to prevent possible microbial inhibition when designing a process for additional energy recovery from microalgae via AD.

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#### Introduction

With increasing interest in sustainable development, huge efforts are now being made to exploit renewable energy sources including biomass. Biomass is a carbon neutral resource during its life cycle and is found throughout the world, which is not subject to global price fluctuations or supply uncertainties [1]. Biomass can be divided into terrestrial and aquatic sources, and the latter has advantages of fast growth rates with high biodegradability [2]. In addition, as a non-edible feedstock it is free from ethical issues, and requires little land and irrigation water. In this regard, microalgae are considered one of the most promising future bioenergy sources [3].

Microalgae are more than two times efficient than terrestrial plants in converting sunlight to biochemical energy and fixing  $CO_2$  [3]. Under certain conditions, they can store neutral lipids, 20–70% of their body, which can be used as an industrial chemical and bioenergy feedstock [4]. In general, to obtain lipids, a microalgae broth is dewatered, cells are ruptured, and finally organic solvents such as methanol and chloroform are added to extract the lipids. During this process, approximately 45–85% of raw microalgae remains as residues, and recent reports have pointed out that attaining more energy from the residues is essential for the feasibility and sustainability of microalgae-based bioenergy systems [5].

One of the viable technologies to convert lipid-extracted microalgae (LEM) to biofuel is anaerobic digestion (AD). AD is a multistage biological conversion route, consisting of hydrolysis, acidogenesis, and methanogenesis, which finally converts organic polymers to methane (CH<sub>4</sub>). Raw and pretreated LEM are often treated by AD, showing a CH<sub>4</sub> yield of 20-70% of their maximum biological potential [6]. In addition, during the acidogenesis step, hydrogen (H<sub>2</sub>) can be obtained, unless the methanogenesis step is active. This process, often called dark fermentation, is attractive in terms of producing carbon-free energy and reacting in a fast manner [7].

Chloroform is used as an organic solvent to extract lipids from microalgae, but it is known to be toxic to microbial cells. It disrupts the microbial cell membrane and compromises cell viability, resulting in inactivation of essential membrane function and denaturation of essential enzymes [8,9]. The inhibitory effects of chloroform have been studied in the field of anaerobic microbial processes. According to Narayanan et al. (1993), chloroform inhibits the activity of methanogenic microorganisms even at a very low concentration of 1.67  $\mu$ M [10]. More recent study has shown that chloroform can inhibit not only the activity of methanogenic archaea but also the activity of acidogenic bacteria and sulfate-reducing bacteria [11]. These findings indicate that special care is required in utilizing LEM as a feedstock for AD.

This study sought to evaluate the inhibitory effect of chloroform on the fermentative  $H_2$  and  $CH_4$  production from LEM. The batch tests were performed using ultrasonicated Chlorella vulgaris as a model feedstock with varying concentration of chloroform in ranges of 0–800 mg CHCl<sub>3</sub>/L and 0–200 mg CHCL<sub>3</sub>/L for  $H_2$  and  $CH_4$  production, respectively. The inhibitor concentrations affecting the microbial activity by "x" percent (IC<sub>x</sub>) were estimated using a non-competitive inhibition model.

#### Materials and methods

#### Feedstock and inoculum preparation

As microalgae, Chlorella vulgaris with carbohydrate content of 12.5%, protein 66.9%, lipid 13%, ash 6%, and others 1.6% was used. The chemical oxygen demand (COD) concentration of C. vulgaris was 1.3 g COD/g dry cell weight (dcw). To prepare LEM, it was diluted with distilled water to 50 g COD/L, and ultrasonicated to disrupt the cell walls at the specific energy input of 100,000 kJ/kg TS using an ultrasonicator with a frequency of 20 kHz and a power of 150 W (VCX-750, Sonics and Materials, USA). The broth was then centrifuged at 7000 rpm for 10 min, and the supernatant was discarded. Finally, the sediment (195 g TS/L) used for the experiment was diluted with distilled water to 108 g TS/L, 91 g VS/L, 115 g COD/L, 22 g carbo./L, and pH 7.1.

The inoculum used in this study was obtained from an anaerobic digester at a local wastewater treatment plant. The pH, alkalinity, and volatile suspended solid (VSS) concentration of the sludge were 7.2, 2.8 g CaCO<sub>3</sub>/L, and 40 g VSS/L, respectively. While raw sludge was used for CH<sub>4</sub> production, heat-treated sludge (90 °C for 20 min) was used for H<sub>2</sub> production to inactivate H<sub>2</sub>-consuming methanogenesis [12].

#### Experiments

Batch tests were carried out using 250 mL serum bottles with a working volume of 150 mL. In the H<sub>2</sub> production experiment, 35 mL of heat-treated sludge and 100 mL of LEM were added to the bottles, corresponding to a substrate concentration of 76 g COD/L and an inoculum-substrate (I/S) ratio of 0.15 [12]. The rest of the working volume was filled with distilled water, and no external nutrients were added. After adding all substances, initial pH was adjusted at 7.5  $\pm$  0.1 by adding 5 N HCl solution and the bottles were purged with N<sub>2</sub> gas for 5 min to provide an anaerobic condition. Finally, specified amounts of 10 N chloroform (CAS no. 67-66-3, CHCl<sub>3</sub>, MW: 119.39) were added to make the desired chloroform concentrations at 0, 10, 25, 50, 200, 400, and 800 mg CHCl<sub>3</sub>/L. All bottles were placed in a shaking incubator controlled at 35 °C and agitated at 100 rpm.

In case of CH<sub>4</sub> production, considering that CH<sub>4</sub>-producers prefer a lower substrate concentration and a higher I/S ratio compared to H<sub>2</sub>-producers [13], the experiment was conducted at a substrate concentration of 10 g COD/L and an I/S ratio of 2 [13]. Chloroform concentration was adjusted at 0, 2.5, 5, 10, 25, 50, 100, and 200 mg/L, and the rest of the experimental procedure was the same as that employed in the H<sub>2</sub> production test. All experiments including a blank (adding only inoculum) were carried out in triplicate, and the results were averaged.

#### Analytical methods

The amounts of CH<sub>4</sub> and carbon dioxide in the biogas were analyzed by a GC (Gow Mac series 580) equipped with a thermal conductivity detector (TCD) and a 2-m  $\times$  2-mm stainless steel column packed with Porapak Q (80/100 mesh). To determine the H<sub>2</sub> content in the biogas, a gas chromatograph

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