



Review

Tailored lanthanide-doped upconversion nanoparticles and their promising bioapplication prospects

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ABSTRACT

Lanthanide-doped upconversion nanoparticles (UCNPs) which could show unique upconversion photoluminescence (UCPL) have attracted considerable attention due to their excellent chemical and optical characteristics. Compared with down-shifting luminescence, UCPL shows its bright future in the bioapplication where sequential absorption of multiple low-energy photons (such as near-infrared (NIR) excitation) by the ladder-like energy levels of the lanthanide ions (Ln^{3+}) leads to the production of higher energy photons (such as ultraviolet, visible light). UCPL shows a variety of advantages such as large anti-Stokes shifts, sharp emissions, long luminescence lifetimes, and high resistance to photobleaching. Furthermore, NIR irradiation has lower photo damage effect, a large penetration depth in tissues, and at the same time, it can avoid the auto-fluorescence interference of biological sample and light scattering phenomenon, which can reduce the background light and improve the signal-to-noise ratio. Thus, UCNPs have emerged as more appropriate nanomaterials for bioapplications. To date, many scientists have focused on the research of the bioapplications of UCNPs, including bioimaging, drug release, and therapies after the special surface modification. In this critical review, recent advances regarding the mechanism, synthesis, modification, and promising bioapplications of UCNPs are reviewed. In particular, the studies related to sensing and bioimaging (UCPL, MR, CT, PET and SPECT), drug release, and therapies (photothermal therapy, photodynamic therapy, and radiotherapy) are presented in detail. For specific bioapplication, tailored UCNPs can be designed and synthesized according to the different synthesis and modification methods as summarized in this review. Finally, we discuss the prospects and challenges of UCNPs in the biomedical and biotechnological fields.

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Contents

1. Introduction	11
2. Synthesis and modification strategies of upconversion nanoparticles	13

Abbreviations: UCNPs, upconversion nanoparticles; UCPL, upconversion photoluminescence; NIR, near-infrared; Ln^{3+} , lanthanide ions; QDs, quantum dots; CDs, carbon dots; CW, continuous-wavelength; EG, ethylene glycol; QY, quantum yield; CT, X-ray computed tomography; MR, magnetic resonance; PET, positron emission tomography; SPECT, positron emission computed tomography; PTT, photothermal therapy; PDT, photodynamic therapy; RT, radiotherapy; ESA, excited state absorption; ETU, energy transfer upconversion; CSU, cooperative sensitization upconversion; CR, cross relaxation; PA, photon avalanche; OA, oleic acid; OM, oleylamine; ODE, 1-octadecene; β -hCG, human chorionic gonadotropin; PEG, polyethylene glycol; CTAB, cetyltrimethylammonium bromide; SiO_2 , silica; mSiO_2 , mesoporous silica; TEOS, tetraethyl orthosilicate; SERS, surface-enhanced Raman scattering; C_{18}TMS , n-octadecyltrimethoxysilane; IBU, ibuprofen; PEI, polyetherimide; HCl, hydrochloric acid; bFGF, basic fibroblast growth factors; FRET, Förster resonance energy transfer; FA, folic acid; $-\text{COOH}$, carboxyl group; $-\text{NH}_2$, amino group; PAA, poly(acrylic acid); $\text{C}_{18}\text{PMH-PEG}$, poly(maleic anhydride-alt-1-octadecene)-poly(ethylene glycol)methyl ethers; PEG-b-PCL, poly(ethylene glycol)-block-poly(caprolactone); PEG-b-PLA, poly((ethylene glycol)-block-lactic acid); 6AA, 6-aminohexanoic acid; EDC, 1-Ethyl-3-(3-dimethylaminopropyl)carbodiimide; NHS, N-Hydroxysuccinimide; NRD, Nile red derivative; GSH, glutathione; MnO_2 , manganese dioxide; BTB, bromothymol blue; GO, graphene oxide; azo, azobenzene; α -TOS, alpha-tocopheryl succinate; DOX, doxorubicin; $^1\text{O}_2$, singlet oxygen; ROS, reactive oxygen species; RB, rose bengal; TiO_2 , titanium dioxide; PpIX, protoporphyrin IX; Ce6, chlorin e6; ZnPc, Zn(II)-phthalocyanine; HA, hyaluronan; CDDP, cisplatin; ATP, Adenosine Triphosphate; PMO, periodic mesoporous organosilica.

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2.1.	Typical synthesis of upconversion nanoparticles	13
2.1.1.	Thermal decomposition method	13
2.1.2.	Hydrothermal/solvothermal method	14
2.1.3.	Co-precipitation method	14
2.2.	Emerging synthesis of upconversion nanoparticles	15
2.3.	Enhancing the upconversion photoluminescence by constructing core/shell structure	15
2.4.	Synthesis of water-soluble upconversion nanoparticles	17
2.4.1.	Inorganic surface silanization	17
2.4.2.	Organic ligand modification method	18
2.5.	Bioconjugation	18
3.	Upconversion nanoparticles for sensing	19
3.1.	Upconversion nanoparticles for anions sensing	19
3.2.	Upconversion nanoparticles for cations sensing	20
3.3.	Upconversion nanoparticles for molecules sensing	20
3.4.	Upconversion nanoparticles for pH sensing	21
3.5.	Upconversion nanoparticles for temperature sensing	21
4.	Upconversion nanoparticles for bioimaging of UCPL, MR, CT, SPECT and PET	21
4.1.	Upconversion nanoparticles for <i>in vitro/in vivo</i> UCPL bioimaging	21
4.2.	Upconversion nanoparticles for dual- and multimode bioimaging	21
4.2.1.	Upconversion nanoparticles for UCPL/MR imaging	21
4.2.2.	Upconversion nanoparticles for UCPL/CT imaging	22
4.2.3.	Upconversion nanoparticles for UCPL/PET or UCPL/SPECT imaging	23
4.2.4.	Upconversion nanoparticles for multi-modal imaging	24
5.	Upconversion nanoparticles for drug release	24
5.1.	Upconversion nanoparticles for lower pH-responsive drug release system	24
5.2.	Upconversion nanoparticles for NIR light-responsive drug release system	25
5.3.	Upconversion nanoparticles for dual-responsive drug release system	26
6.	Upconversion nanoparticles for bioimaging guided therapies	26
6.1.	Upconversion nanoparticles for bioimaging-guided photothermal therapy	26
6.2.	Upconversion nanoparticles for bioimaging-guided photodynamic therapy	28
6.3.	Upconversion nanoparticles for bioimaging-guided radiotherapy	29
7.	Summary and prospects	30
	Acknowledgments	30
	References	30

1. Introduction

Up to now, lanthanide ions (Ln^{3+}) doped luminescent materials via intra- $4f$ or $4f$ - $5d$ transitions have drawn rather consistent interest due to their outstanding features, such as excellent photostabilities, sharp-band emissions, long luminescence lifetimes as well as large anti-Stokes shifts [1,2]. In particular, Ln^{3+} -doped upconversion nanoparticles (UCNPs) have received considerable attention due to their special anti-Stokes photoluminescent properties compared with traditional down-shifting luminescent materials, since Auzel et al. systematically proposed and summed up the mechanism of upconversion materials [3]. Upconversion photoluminescence (UCPL) is a non-linear process where sequential absorption of two or more low-energy photons (such as near-infrared (NIR) light leads to the production of higher energy photons (such as ultraviolet, visible, and NIR light). It is attributed to the intra- $4f$ or $4f$ - $5d$ transitions in the ladder-like, abundant and unique energy levels of the Ln^{3+} [4,5]. So far, although a series of luminescent materials, including fluorescent proteins [6,7], organic dyes [8], metal complex [9], semiconductor quantum dots (QDs) [10], and carbon dots (CDs) [11,12] have been developed for luminescent contrast agents and bioapplications, UCNPs have emerged as more appropriate nanomaterials for bioapplications. In addition, it is available and enforceable that tailored UCNPs can be designed and synthesized for specific bioapplication according to the following aspects.

First, the process of upconversion photoluminescence being carried out by low-power NIR continuous-wavelength (CW) laser (e.g., 808 nm or 980 nm) exhibits several advantages, such as deep penetration of living organisms [13,14], weak autofluorescence [15,16], minimal photobleaching [17,18], and low toxicity

[19,20], suitable for deep tissue optical imaging or light-responsive drug release and therapies. In addition, NIR light poses minimized photo-damage to living organisms compared with UV and visible light. Generally, UCNPs are composed of three parts: inorganic host matrix, sensitizer ions, and activator ions which could also be referred to luminescent centers. Ln^{3+} ions embedded in a suitable inorganic matrix, such as NaYF_4 [21–24], NaYbF_4 [18], $\text{NaGdF}_4/\text{BaGdF}_5$ [25,26], NaLuF_4 [27,28], $\text{Y}_2\text{O}_3/\text{Gd}_2\text{O}_3$ [29,30], Bi_2O_3 [31], and LnF_3 [32,33], etc., are able to emit distinctive and characteristic upconversion photoluminescence. Among these matrices, NaYF_4 is one of the most popular and ideal host materials due to its excellent properties including high tolerance, low phonon energy, high transparency, and outstanding chemical and thermal stability [34,35]. The desired sensitizers possess relatively large photo absorption cross-sections and appropriate energy levels to match the corresponding activators. To date, a number of researches have demonstrated that Yb^{3+} is an efficient sensitizer with high two-photon absorption cross sections (976 nm, ${}^2\text{F}_{7/2} \rightarrow {}^2\text{F}_{5/2}$), which is co-doped with activator (e.g., Er^{3+} , Tm^{3+} or Ho^{3+}) in the host lattice for the energy transfer from Yb^{3+} to activators under the excitation of 980 nm [36,37]. Recently, some groups focused on a new route to construct the Nd^{3+} -sensitized UCNPs with the energy transfer process [${}^4\text{F}_{3/2}$ (Nd^{3+}) \rightarrow ${}^2\text{F}_{5/2}$ (Yb^{3+}) \rightarrow activators] under the excitation of 808 nm [38–43]. In this way, the energy transfer efficiency under the excitation of 808 nm is similar with that of 980 nm, however, the 808 nm excitation shows minimized heating effect to biological tissues [44,45], since 808 nm band with lower water absorbance is expected (Fig. 1) [39]. Generally, Er^{3+} , Tm^{3+} , and Ho^{3+} ions holding ladder-like arrangement energy levels, are frequently used as activators to generate UCPL emission under NIR excitation [46].

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