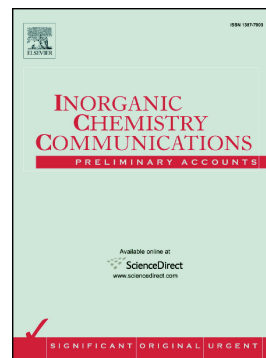


## Accepted Manuscript

Two novel luminescent Cd(II)/Zn(II) coordination polymers based on 4,4'-(1H-1,2,4-triazol-1-yl)methylene-bis(benzonic acid) for sensing organic molecules and Fe<sup>3+</sup> ion

Ya-Ru Huang, Ling-Ling Gao, Jie Zhang, Xiao-Qing Wang, Li-Ming Fan, Tuo-Ping Hu



PII: S1387-7003(18)30098-4

DOI: doi:[10.1016/j.inoche.2018.03.010](https://doi.org/10.1016/j.inoche.2018.03.010)

Reference: INOCHE 6914

To appear in: *Inorganic Chemistry Communications*

Received date: 30 January 2018

Revised date: 23 February 2018

Accepted date: 9 March 2018

Please cite this article as: Ya-Ru Huang, Ling-Ling Gao, Jie Zhang, Xiao-Qing Wang, Li-Ming Fan, Tuo-Ping Hu, Two novel luminescent Cd(II)/Zn(II) coordination polymers based on 4,4'-(1H-1,2,4-triazol-1-yl)methylene-bis(benzonic acid) for sensing organic molecules and Fe<sup>3+</sup> ion. The address for the corresponding author was captured as affiliation for all authors. Please check if appropriate. Inoche(2017), doi:[10.1016/j.inoche.2018.03.010](https://doi.org/10.1016/j.inoche.2018.03.010)

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.



ELSEVIER

## Inorganic Chemistry Communications

journal homepage: [www.elsevier.com/locate/icc](http://www.elsevier.com/locate/icc)

# Two Novel Luminescent Cd(II)/Zn(II) Coordination Polymers Based on 4,4'-(1H-1,2,4-triazol-1-yl)methylene-bis(benzonic acid) For Sensing Organic Molecules and Fe<sup>3+</sup> Ion

Ya-Ru Huang<sup>a</sup>, Ling-Ling Gao<sup>a</sup>, Jie Zhang<sup>a</sup>, Xiao-Qing Wang<sup>a</sup>, Li-Ming Fan<sup>a</sup>, Tuo-Ping Hu<sup>a\*</sup><sup>a</sup> Department of Chemistry, College of Science, North University of China, Taiyuan 030051, China.**ARTICLE INFO**

Article history:

Received xxxxxx

Accepted xxxxxx

Available online xxxxxx

Keywords:

Coordination polymers

Luminescent detection

Organic solvent molecules

Metal ions

**ABSTRACT:** Two novel coordination polymers,  $\{[Cd(TZMB)(1,4-bib)_{0.5}]\cdot(H_2O)_{1.5}\}_n$  (**1**) and  $\{[Zn(TZMB)(1,4-bib)_{0.5}]\}_n$  (**2**), have been synthesized under solvothermal conditions. Complex **1** displays a 3D structure with the  $(4^7.8^4)(4.7^2.8^3)$  topology and Complex **2** is a 3D structure with the point symbol of  $\{4.8^2\}\{4.8^5\}$ . And complex **1** shows good luminescent quenching effect for organic solvent molecules (actone and p-nitrophenol) and Fe<sup>3+</sup>.

In recent years, nitrobenzene derivatives as raw materials are widely used in the syntheses of the dyes and pesticides, but casual emissions of them also pollute the environment [1–2]. Meanwhile, the random disposal of heavy metal ions poses a great threat to mankind. Hence, sensitive and selective detection of them become a pressing issue in protecting environment. And some methods have been used to detect the above-mentioned pollutants, such as ion mobility spectroscopy (IMS), Raman spectroscopy, X-ray methods and so on [3–5]. But these measurements are not always convenient and available because of the high cost and low sensitivity of instruments. In view of the above situation, so it is a realistic job exploring a simple, inexpensive, and sensitive method to detect these substances. A large number of references have been reported on the use of fluorescent complexes as probes, which are superior to traditional methods in terms of cost and sensitivity [6–7].

As we all known, during the process of synthesizing luminescent coordination polymers (LCPs), the selection of metal salts and organic ligands is two key factors, which affect properties of LCPs [8–9]. Firstly, d<sup>10</sup> metal ions (Zn(II)/Cd(II)) are one of very important candidates on account of their excellent optical properties; Secondly, a ligand containing imidazole and carboxyl groups can adopt diverse coordination modes forming structure-stable and property-excellent LCPs. Inspired by the above mentioned

facts, the ligand 4,4'-(1H-1,2,4-triazol-1-yl)methylene-bis(benzonic acid) (H<sub>2</sub>TZMB) was chosen as a main linker, and two novel CPs, namely  $\{[Cd(TZMB)(1,4-bib)_{0.5}]\cdot(H_2O)_{1.5}\}_n$  (**1**) and  $\{[Zn(TZMB)(1,4-bib)_{0.5}]\}_n$  (**2**), have been constructed with the help of transition metal ions and 1,4-bib ligands (Scheme S1) under hydrothermal condition and characterized by single-crystal X-ray diffraction analysis, elemental analysis, infrared spectrum analysis (IR), thermogravimetric analysis (TG), topological analysis and powder X-ray diffraction (PXRD). Furthermore, luminescent properties of **1** and **2** were also investigated.

The experimental sections, including instruments and structure refinement, have been listed in the Electronic supplementary information (ESI). Crystallographic data for complexes **1** and **2** are given in Table S1. Selected bond lengths and angles of title complexes are listed in Table S2.

A mixture of H<sub>2</sub>TZMB (0.024 mmol, 7.8 mg), 1,4-bib (0.024 mmol, 5.0 mg) and Cd(NO<sub>3</sub>)<sub>2</sub>·6H<sub>2</sub>O (0.024 mmol, 7.40 mg) was dissolved in 3 ml H<sub>2</sub>O and 0.05 mL NaOH (0.25 mol·L<sup>-1</sup>) aqueous solution, and then they were sealed in a 25 ml Teflon-lined stainless steel vessel. After that, the reaction vessel was heated at 110°C for 3000 min. Finally, colorless block crystals of **1** were obtained by cooling to room temperature slowly. And the synthesis process of complex **2** was similar to that of **1**.

\*Corresponding author.

E-mail address: [hutuopingsx@126.com](mailto:hutuopingsx@126.com).

Download English Version:

<https://daneshyari.com/en/article/7748517>

Download Persian Version:

<https://daneshyari.com/article/7748517>

[Daneshyari.com](https://daneshyari.com)