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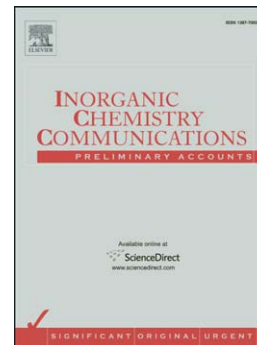
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# A new self-penetrating amine-decorated microporous metal–organic framework: crystal structure, adsorption selectivity, and luminescence properties

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## ABSTRACT

A new microporous metal–organic framework,  $[\text{Cd}(\text{bpdc})_{0.5}(\text{atz})(\text{DMF})] \cdot 0.5\text{DMF}$  (**1**) ( $\text{H}_2\text{bpdc}$  = 4,4'-biphenyldicarboxylic acid,  $\text{Hatz}$  = 3-amino-1,2,4-triazole, DMF = N,N-dimethylformamide) has been solvothermally synthesized by employing the mixed  $\text{H}_2\text{bpdc}$  and  $\text{Hatz}$  ligands. **1** is a 3D pillared-layer framework, consisting of Cd-triazolate layer and dicarboxylate pillar, which exhibits a 6-connected ( $4^6 \cdot 6^8 \cdot 8$ ) self-penetrating net. Because of the pores system functionalized by amino groups and open metal sites, this material shows high  $\text{CO}_2$  adsorption selectivity over  $\text{H}_2$  and  $\text{N}_2$ . In addition, **1** exhibits blue emission at ambient temperature.

*Keywords:* Metal-organic framework, Self-penetrating, Sorption, Luminescence

Porous metal–organic frameworks (MOFs) have gained great interests due to their fascinating topologies and promising applications as materials [1]. Particularly, MOFs have been extensively studied for selective  $\text{CO}_2$  capture in recent years due to their advantages such as high porosity, modifiable pore surface, and flexible structures [2]. Commonly, the interactions between the framework and adsorbate play a primary role for enhancing the MOF's sorption affinity and capacity, especially in low pressure. Thus, some strategies, including generation of open metal sites, control of pore size, and incorporation of specific polar functional groups ( $-\text{NH}_2$ ,  $-\text{CF}_3$ ,  $-\text{OH}$ , etc.) on the ligands, have been investigated to augment the  $\text{CO}_2$  adsorptive affinity and selectivity [3]. Grafting of amines onto the pore surfaces of MOFs not only allows for the increase of polarity but also could adjust the pore size, which has been investigated to augment the  $\text{CO}_2$  adsorptive affinity and selectivity [4].

On the other hand, Self-penetration, in contrast to interpenetrating nets, features a single network having the peculiarity that the smallest topological circuits from the same network are catenated with each other. Compared to the fruitful productions of interpenetrating networks,

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