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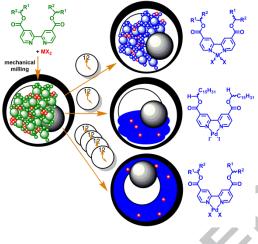
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Mechanochemical synthesis of zinc and palladium complexes of dialkyl 2,2'-bipyridine-4,4'-dicarboxylate and analysis of solid-state reaction kinetics.

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Mechanochemical synthesis of a series of Zn and Pd dialkyl 2,2'-bipyridine-4,4'-dicarboxylate complexes (Lⁿ, where n specifies the number of carbons in the α - or β -substituted alkyl chains) including ZnL^{β 12}Cl₂, ZnL^{β 14}Cl₂, ZnL^{α 14}Cl₂, ZnL^{α 14}Cl₂, ZnL^{α 14}Cl₂, and PdL^{α 14}Cl₂ is explored through a variety of linear and sigmoidial reaction models. The Johnson-Mehl-Avrami-Yerofeev-Kolmogrov (JMAYK) and Finke-Watzky (FW) models are largely equivalent as demonstrated through analysis of Akaike weights. This analysis was extended to the previously reported syntheses of PdL^{β 10}I₂, PdL^{α 16}I₂, and PdL^{β 16}I₂ with similar results,

except that the JMAYK model is significantly superior to the FW model for $PdL^{\beta 16}I_2$. The physical properties of the reaction mixture, most prominently whether the product phase is easily mixed, is a major factor in determining the rate of reaction.

Keywords: mechanochemistry; ball mill; solid-state synthesis; kinetics; transition metal complexes

1. Introduction

Mechanochemical synthesis, although not a new technology, is currently being used to prepare many more types of compounds than the oxides, alloys, and compounds that were the previously the focus of the technique [1, 2]. Sustainability drives much of this research as the mechanical force used in these syntheses can be applied in the absence of solvent, drastically reducing the waste in these reactions [3]. From the perspective of researchers at less well-resourced institutions, mechanochemistry also provides a method for highly cost-effective preparation of transition metal complexes; an area of study only recently emerged in mechanochemical synthesis [4]. However, in order for mechanochemical synthesis to be utilized for routine small molecule and Download English Version:

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