



## Research paper

# Phosphorus-nitrogen compounds. Part 40. The syntheses of (4-fluorobenzyl) pendant armed cyclotetraphosphazene derivatives: Spectroscopic, crystallographic and stereogenic properties, DNA interactions and antimicrobial activities

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## ABSTRACT

The Cl substitution reaction of octachlorocyclotetraphosphazene,  $N_4P_4Cl_8$  (**1**), with 1.2 equimolar amounts of sodium salt of N/O donor-type bidentate ligand (**2**) gave four different new (4-fluorobenzyl) pendant armed cyclotetraphosphazene derivatives; namely, mono-(4-fluorobenzyl)-spiro- (spiro; **3**), mono-(4-fluorobenzyl)-2-*cis*-4-dichloro-ansa- (2,4-ansa; **4**), bis-(4-fluorobenzyl)-2-*trans*-6-dispiro (**5**) and bis-(4-fluorobenzyl)-2-*trans*-4-dispiro (**6**) cyclotetraphosphazenes of which **3** was the major product (yield 55%). Compound **3** was treated with mono and difunctional reagents to prepare the fully substituted products (**3a–3j**) due to its very high yield. However, tetrapyrrolidino-2-*cis*-4-dichloro-ansa- product (**4a**) was obtained from the reaction of 2,4-ansa (**4**) with excess pyrrolidine. The new cyclotetraphosphazenes were characterized by elemental analyses, mass spectrometry (ESI-MS), Fourier transform infrared (FTIR),  $^1H$ ,  $^{13}C$ , and  $^{31}P$  NMR techniques. The structures of **3** and **3g** were determined by X-ray crystallography. 2,4-Ansa compounds (**4** and **4a**) have two stereogenic P atoms. The stereogenic property of **4** was identified by  $^{31}P$  NMR spectra in the addition of the chiral solvating agent, (R)-(+)-2,2,2-trifluoro-1-(9'-anthryl)-ethanol (CSA). The tetraspiro compounds (**3i** and **3j**) look similar to a propeller, and they may have P and M atropisomers. The antimicrobial activities of the compounds were screened against some G(−)/G(+) bacteria and yeast strains. The interactions of the compounds with supercoiled plasmid pBR322 DNA and their inhibited DNA restriction were examined.

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## 1. Introduction

Octachlorocyclotetraphosphazene [tetramer,  $N_4P_4Cl_8$  (**1**)] is a highly reactive inorganic heterocyclic ring system which is a starting material used in the syntheses of new substituted cyclotetraphosphazene derivatives [1,2]. The investigations of the substituent exchange reaction patterns of Cl atoms in tetramer with monodentate ligands, such as amines and alkoxides, are substantially examined in the literature [3–5]. However, the reactions of  $N_4P_4Cl_8$  with the bifunctional (diamines, dialkoxides and aminoalkoxides) and polyfunctional (polyamines, polyaminoalkoxides and polyalkoxides) reagents are quite limited [6–10]. The partly and fully substituted products, e.g. spiro, ansa, bino, dispiro,

spiroansa, spirobino, trispiro, tetraspiro, spiroansa-spiro and ansa-spiroansa, were produced as the result of these condensation reactions depending upon the substituents and reaction conditions [8,10,11,12]. Furthermore, as a result of the reactions, the geometrical (geminal and *non*-geminal *cis/trans*) and optical isomers ought to be obtained. The chiralities of the cyclotetraphosphazenes have also been the exciting topic in recent years [8,13–15]. But, the separations, purifications and characterizations of the geometrical and optical isomers of these products are very difficult in some cases [14]. In the literature, there are a pretty small number of papers investigating the stereogenic properties of the tetrameric phosphazenes using  $^{31}P$  NMR spectroscopy in the addition of (R)-(+)-2,2,2-trifluoro-1-(9'-anthryl)-ethanol (CSA) and by the chiral HPLC techniques [8,15].

Besides, some of the organocyclophosphazenes have attracted much interest in their diverse chemical, technological and biological applications, eg. the utilization as ligands and strong bases in

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coordination and organometallic chemistry [16], preparation of the phosphazenes polymers because of a great deal of the flexibility of phosphazene ring [17], dendrimers [18], liquid crystalline materials [19], ionic liquids [20], flame retardant additives [21], fluorescence chemosensors [22], photosensitizers [23], Langmuir–Blodgett thin films [24], and antibacterial, antifungal, anti-cancer and antituberculosis agents [8,15,25,26].

On the other hand, the N/O donor-type bidentate ligand, sodium 3-(4-fluorobenzylamino)-1-propanoxide (**2**), used as the starting material in the present study treated with hexachlorocyclotriphosphazene (trimer,  $\text{N}_3\text{P}_3\text{Cl}_6$ ) to obtain only the spirocyclotriphosphazenes, regioselectively [27]. But, it was observed in this study that  $\text{N}_4\text{P}_4\text{Cl}_8$  and **2** gave four kinds of compounds; namely, mono-(4-fluorobenzyl)-spiro- (**3**), mono-(4-fluorobenzyl)-2-*cis*-4-dichloro-ansa- (**4**), bis-(4-fluorobenzyl)-2-*trans*-6-dispiro (**5**) and bis-(4-fluorobenzyl)-2-*trans*-4-dispiro (**6**) cyclotetraphosphazenes.

Eventually, the present work essentially focuses on the Cl replacement reactions of  $\text{N}_4\text{P}_4\text{Cl}_8$  with 1.2 equimolar amounts of **2** (Scheme 1) for assessment of the spectroscopic, crystallographic and stereogenic properties, antimicrobial activities and the pBR322 DNA interactions of the cyclotetraphosphazenes.

## 2. Experimental part

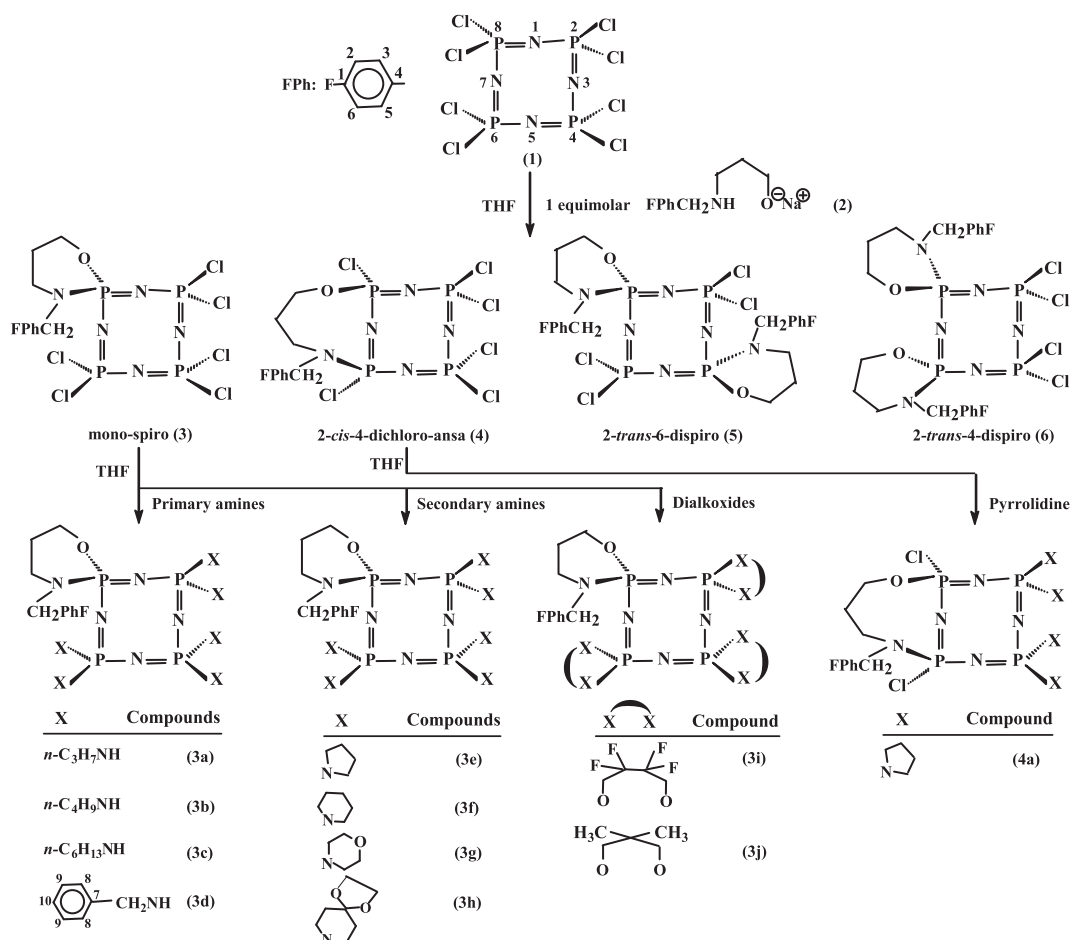
### 2.1. Materials and method

The condensation reactions were pursued by thin-layer chromatography on Merck DC Alufolien Kieselgel 60  $\text{B}_{254}$  sheets

in suitable solvents. The column chromatography was realised on the Merck Kieselgel 60 (230–400 mesh ATSM) silica gel. The melting points of the obtained compounds were determined with a Gallenkamp apparatus using a capillary tube. The elemental analyses were carried out using the Leco CHNS-932 instrument. ESI mass spectra of the products were enrolled with a Waters 2695 Alliance Micromass ZQ spectrometer. The FTIR spectra were recorded on a Jasco FT/IR-430 spectrometer in KBr discs.  $^1\text{H}$  and  $^{13}\text{C}$   $\{^1\text{H}\}$  NMR spectra of all the cyclotetraphosphazenes were monitored on a Varian Mercury FT-NMR (400 MHz) spectrometer using  $\text{SiMe}_4$  as an internal standard operating at 400.13 MHz and 100.62 MHz, respectively.  $^{31}\text{P}$   $\{^1\text{H}\}$  NMR spectra were saved on a Bruker Avance III HD (600 MHz) spectrometer using  $\text{H}_3\text{PO}_4$  (85%) as an external standard operating at 242.94 MHz. Experiments involving the CSA were done by the addition of small aliquots of a concentrated solution of the CSA in the solvent used in NMR spectroscopy, and the  $^{31}\text{P}$   $\{^1\text{H}\}$  NMR spectra were recorded for each addition. The spectrometers were fitted with the 5 mm PABBO BB inverse-gradient probe, and the standard Bruker pulse programs [28] were employed. The microanalyses were designated by the microanalytical service of Ankara University (FTIR, ESI-MS,  $^1\text{H}$  and  $^{13}\text{C}$   $\{^1\text{H}\}$  NMR) and İnönü University ( $^{31}\text{P}$   $\{^1\text{H}\}$  NMR).

### 2.2. Materials used in the syntheses

The Cl replacement reactions were performed under Ar atmosphere. The organic solvents were dried and purified using the standard methods.  $\text{N}_4\text{P}_4\text{Cl}_8$  (Otsuka and recrystallized from hot



**Scheme 1.** Phosphazene derivatives obtained from the reactions of  $\text{N}_4\text{P}_4\text{Cl}_8$  (**1**) with **2**, and amino and/or dialkoxo derivatives of the compounds (**3** and **4**).

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