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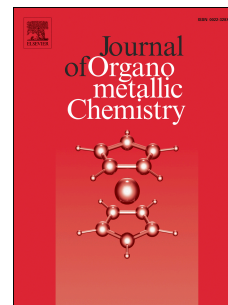
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Organotin(IV) compounds derived from ibuprofen and cinnamic acids, an alternative into design of anti-inflammatory by the cyclooxygenases (COX-1 and COX-2) pathway

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Abstract

New tributyl-, dibutyl- and diphenyl-tin(IV) complexes derived from ibuprofen and cinnamic acids were synthesized. All compounds were structurally characterized by FT-IR, multinuclear ¹H, ¹³C, ¹⁹F and ¹¹⁹Sn NMR and corroborated by 2D spectra. The NMR data in CDCl₃ revealed several hexacoordinated compounds with octahedral geometry. Moreover, in DMSO-d₆ some of these complexes switched to heptacoordination with a pentagonal-bipyramidal geometry due to the inclusion of a solvent's molecule; their ¹¹⁹Sn signals moved up field by around 58 ppm compared to their chemical shifts in non-coordinated solvent CDCl₃. The structural results were supported by Density Functional Theory (DFT) computational calculations. In addition, a docking study was performed to evaluate the ability of ligands to interact within the active site of cyclooxygenases (COX-1 and COX-2). Docking results showed a possible binding of stannoxanes theoretically more selective towards COX-2 than ibuprofen.

Keywords: Stannoxanes; Ibuprofen; Coordination; DFT; Docking; Cyclooxygenase.

1. Introduction

Tin compounds offer a large structural variety due to the range of coordination numbers that the tin atom can adopt (4-8). This leads to different chemical shifts in ¹¹⁹Sn NMR spectroscopy, ranging of +800 to -600 ppm [1-3]. The type and size of the substituents attached to the tin(IV) atom determines the spatial geometry, as well as their industrial and biological applications. It is well known that an increase in the coordination number of the tin atom is related to an increase in electron shielding around the nucleus, that is monitored by the NMR spectra of ¹¹⁹Sn, as well as the ⁿJ(¹³C-^{117/119}Sn) coupling

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