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Trinuclear clusters containing 2-aminopyridinate/pyrimidinate ligands as electrocatalysts for proton reduction

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## ABSTRACT

Triiron and triruthenium clusters containing capping 2-aminopyridinate/pyrimidinate ligands are developed as functional models of the [FeFe]-hydrogenase for electrocatalytic reduction of protons to hydrogen. The 48-electron clusters  $Fe_3(CO)_9(\mu_3-pyNH)(\mu-H)$  (1),  $Fe_3(EO)_9(\mu_3-pyNH)(\mu-H)$  (1),  $Fe_3(EO)_9(\mu-H)$  (1),  $Fe_3(EO)_9(\mu-H)$ pymNH)(µ-H) (2), Ru<sub>3</sub>(CO)<sub>9</sub>(µ<sub>3</sub>-pyNH)(µ-H) (3) and Ru<sub>3</sub>(CO)<sub>9</sub>(µ<sub>3</sub>-pymNH)(µ-H) (4) (pyNH = 2-aminopyridinate, pymNH = 2-aminopyrimidinate) are prepared from reactions of  $M_3(CO)_{12}$  (M = Fe or Ru) with the corresponding heterocyclic amine at elevated temperatures. Each contains a hydride and a residual amino hydrogen in close proximity (ca. 2.8 Å). The triiron 2-aminopyridinate cluster 1 does not protonate by TsOH·H<sub>2</sub>O (used as the proton source during catalysis), whereas its ruthenium analogue 3 undergoes slow protonation across a ruthenium-ruthenium bond. The 2-aminopyrimidinate clusters 2 and 4 undergo rapid protonation at the ring nitrogen. In MeCN, the triiron clusters show a single irreversible reduction wave ( $E_p = -1.61$  V for 1;  $E_p = -1.47$  V for 2) in the cathodic region of their CVs, while the triruthenium clusters display a pair of irreversible cathodic waves ( $E_p$  = -2.01 and -2.15 V for 1;  $E_p = -1.93$  and -2.09 V for 2). All catalyze proton reduction in the presence of TsOH·H<sub>2</sub>O but different mechanisms are proposed. The triiron clusters are more efficient toward proton reduction and operate at reduced overpotentials as compared to their triruthenium analogues. Introduction of a potential proton relay in these clusters neither increases the efficiency nor reduces the overpotential of catalysis which is in sharp contrast with the results observed for hydrogenase biomimetics.

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