



New rhenium carbonyl cluster complexes containing bridging hydrocarbyl and bridging mercury groups



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ABSTRACT

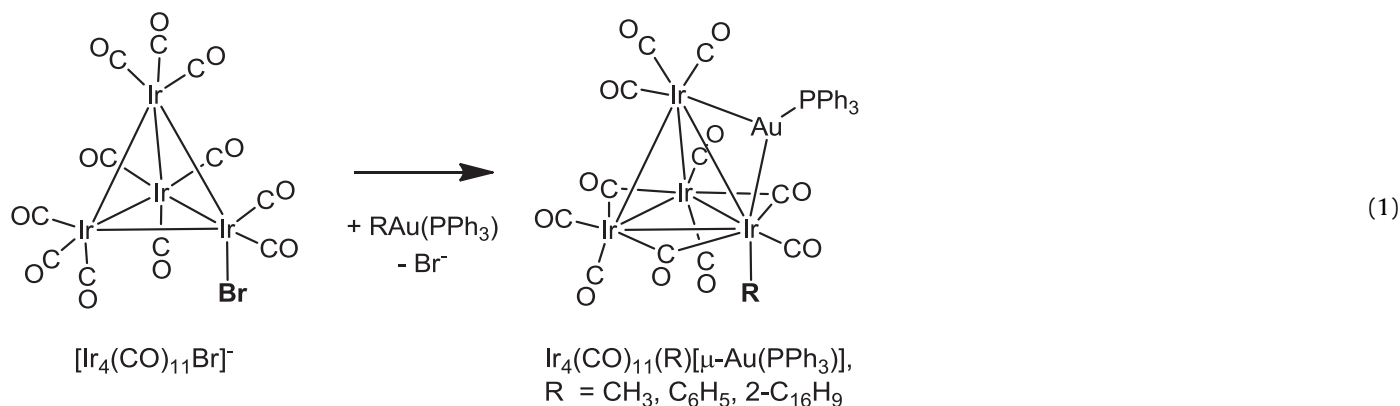
The new rhenium–mercury complexes $[\text{Re}_2(\text{CO})_8(\mu\text{-HgI})(\mu\text{-}\eta^1\text{-C}_6\text{H}_5)]_2$, **2** and $[\text{Re}_2(\text{CO})_8[\mu\text{-HCC(H)C}_4\text{H}_9](\mu\text{-H})]$, **3** were obtained from the reactions of $\text{Re}_2(\text{CO})_8[\mu\text{-Au}(\text{PPh}_3)](\mu\text{-}\eta^1\text{-C}_6\text{H}_5)$, **1** with HgI_2 and of $\text{Re}_2(\text{CO})_8[\mu\text{-HCC(H)C}_4\text{H}_9](\mu\text{-H})$ with $\text{Hg}(\text{C}_6\text{H}_5)_2$, respectively. In the solid state compound **2** is dimer of $\text{Re}_2(\text{CO})_8(\mu\text{-HgI})(\mu\text{-}\eta^1\text{-C}_6\text{H}_5)$ that held together by iodide ligands that asymmetrically bridge between the two mercury atoms. Each dirhenium group is formally electronically unsaturated and contains one bridging $\eta^1\text{-C}_6\text{H}_5$ ligand. Compound **3** contains two $\text{Re}_2(\text{CO})_8[\mu\text{-HCC(H)C}_4\text{H}_9]$ groups held together by a quadruply bridging *spiro*-structured mercury atom.

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Introduction

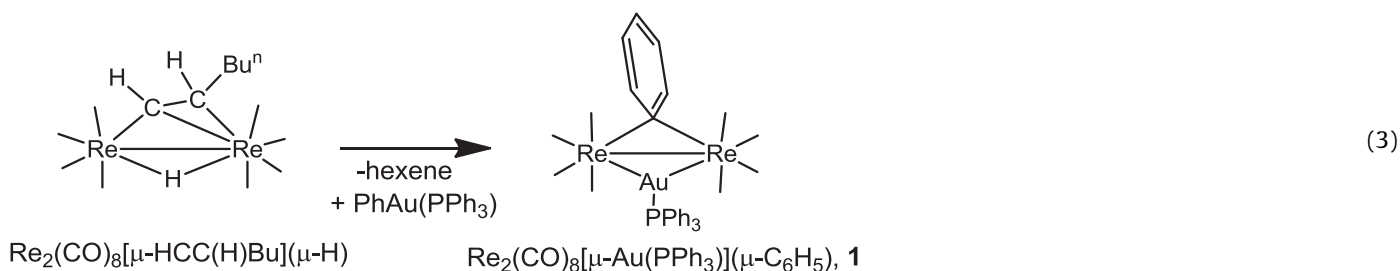
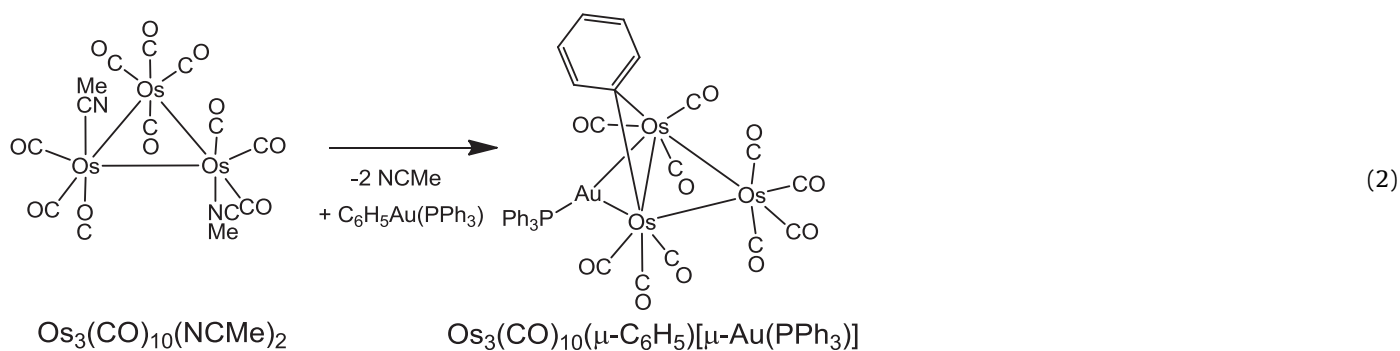
In recent studies it has been shown that arylgoldphosphine compounds, such as $(\text{PPh}_3)\text{Au}(\text{C}_6\text{H}_5)$, can be readily oxidatively added to activated 3rd row polynuclear metal carbonyl cluster complexes to yield metalcarbonyl cluster complexes containing

aryl ligands and bridging gold phosphine groupings, e.g. Eqs. (1)–(3) [1–3]. In some cases, the aryl ligands have adopted unusual bridging coordination modes that can result in interesting physical and chemical properties, such a hindered rotation about the metal–metal bond [4].



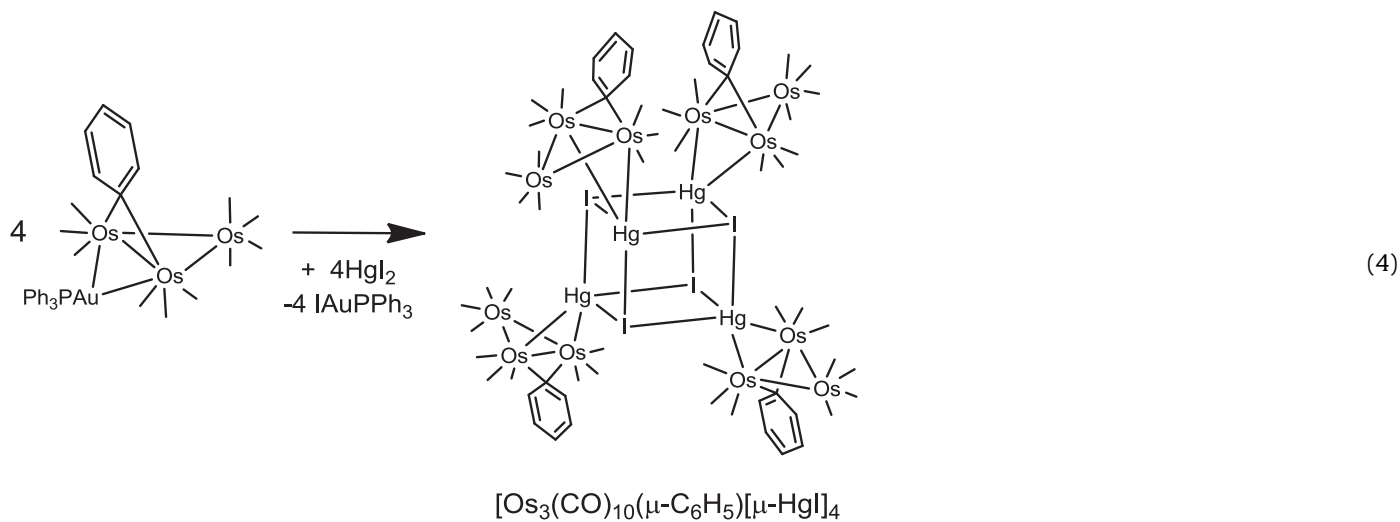
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We have found that the gold phosphine group can be replaced in some of these complexes in reactions with mercuric halides, Eq. (4) [5].

We have now investigated the reaction of $\text{Re}_2(\text{CO})_8[\mu\text{-Au}(\text{PPh}_3)](\mu\text{-}\eta^1\text{-C}_6\text{H}_5)$, **1** with HgI_2 . We have found that the AuPPh_3 group is replaced by a bridging HgI group with concomitant formation of $\text{IAu}(\text{PPh}_3)$ and the dirhenium product $\text{Re}_2(\text{CO})_8(\mu\text{-HgI})(\mu\text{-}$



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