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# Research on the synergistic doped effects and the catalysis properties of $\text{Cu}^{2+}$ and $\text{Zn}^{2+}$ co-doped $\text{CeO}_2$ solid solutions

Guofang Zhang\*, Yiming Li, Zhonghui Hou, Jianyi Xv, Qingchun Wang, Yanghuan Zhang

*School of Material and Metallurgy, Inner Mongolia University of Science and Technology, Baotou, 014010, China*

\*Corresponding author. Address: School of Material and Metallurgy, Inner Mongolia University of Science and Technology, No.7 Arding Street, 014010, Baotou, China. Tel.: +86 472 5951572; fax: +86 472 5951572. E-mail address: afang1001@126.com (G. Zhang).

## Abstract

The  $\text{Cu}^{2+}$  and  $\text{Zn}^{2+}$  co-doped  $\text{CeO}_2$ -based solid solutions were synthesized via hydrothermal method. The microstructure and the spectra features of the solid solutions were characterized systematically. The XRD results showed that the dopants were incorporated into the  $\text{CeO}_2$  lattice to form  $\text{Ce}_{1-x}\text{Cu}_{0.5x}\text{Zn}_{0.5x}\text{O}_2$  solid solutions when  $x$  was lower than 0.14. The cell parameters and the crystalline size decreased linearly, and the lattice strain gradually increased with increasing the doping level. The TEM patterns showed that the particle size in the solid solution was lower than 10 nm which is in accordance with the XRD results. The ICP analysis indicated that the real doped content in the solid solution was close to the nominal proportion. XPS proved that the  $\text{Ce}^{3+}$  component was increased by doping. The Raman and PL spectra indicated that the lattice distortion and the oxygen vacancies also increased following the same trend. At the same time, the synergistic effects of two ions co-doped solid solutions were studied by comparing them with that of single ions doped samples. The catalysis effects of  $\text{Cu}^{2+}$  and  $\text{Zn}^{2+}$  co-doped  $\text{CeO}_2$ -based solid solutions on the hydrogen storage electrochemical and kinetic properties of  $\text{Mg}_2\text{Ni}$  alloys were detected. The electrochemistry properties of the  $\text{Mg}_2\text{Ni}$ -Ni-5wt%  $\text{Ce}_{1-x}\text{Cu}_{0.5x}\text{Zn}_{0.5x}\text{O}_2$  composites indicated that the doped catalysts could provide better optimizations to improve the maximum discharge capacities and the discharge potentials. On the other hand, the charge transfer abilities on the surface and diffusion rate of H atoms in the bulk of alloys also got improved. The DSC measurements showed that the hydrogen desorption activation of the hydrogenated composites with  $\text{Ce}_{0.88}\text{Cu}_{0.06}\text{Zn}_{0.06}\text{O}_2$  solid solutions decreased to  $77.03 \text{ kJ}\cdot\text{mol}^{-1}$ , while that of the composites with pure  $\text{CeO}_2$  was  $97.62 \text{ kJ}\cdot\text{mol}^{-1}$ . The catalysis effect was enhanced by the doped content increase that means that the catalysis mechanism had close links to the oxygen vacancy concentration and the lattice defects in the solid solutions. On the other hand, the doped  $\text{Cu}^{2+}$  and  $\text{Zn}^{2+}$  ions could also play an important role in the catalytic process.

## Graphical abstract

The corresponding HRTEM image as is shown in the figure reveals the well defined fringes with interplanar spacing of 0.31 nm corresponding to the (111) planes of  $\text{CeO}_2$ . The SEAD pattern presents several concentric diffraction circles, which suggests the existence of polycrystalline species. These diffraction rings can be indexed to the (111), (200), (220), (311), (400), (422) and (311) of  $\text{CeO}_2$  nanoparticles.

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