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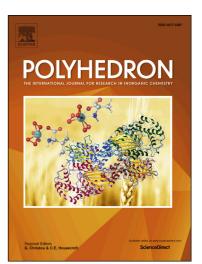
Strategies Toward Catalytic Biopolymers: Incorporation of Tungsten in Alginate Aerogels

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## ACCEPTED MANUSCRIPT

## **Strategies Toward Catalytic Biopolymers: Incorporation of Tungsten in Alginate Aerogels**

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This paper is dedicated to Professor Spyros Perlepes, a great mentor and scientist, on the occasion of his 65th birthday with our best wishes.

**Abstract:** We report the synthesis of W-doped biopolymer aerogels with potential applications to catalysis. The biopolymers were based on alginate, a cost-efficient natural material, and were prepared in environmentally friendly water/ethanol solutions. Gelation of alginate was induced by crosslinking with  $Ca^{2+}$ . The resulting wet-gels were impregnated with an ethanolic solution of the ditungsten complex  $[W_2(\mu-OEt)_2(OEt)_2(EtOH)_2Cl_4]$  ({W=W}<sup>8+</sup>) and were dried with supercritical fluid (SCF) CO<sub>2</sub> to yield W-doped alginate aerogels containing 10% w/w tungsten. Dry materials were characterized with FTIR, SEM/EDS, TGA, N<sub>2</sub> porosimetry and He pycnometry. The bulk densities were low (<0.1 g cm<sup>-3</sup>), porosities were high (96% v/v), as well as the BET surface areas (380 m<sup>2</sup>/g). Pyrolysis of those aerogels at 800 °C under Ar yielded carbon along with a mixture of calcium tungstates, i.e., CaWO<sub>4</sub> and Ca<sub>3</sub>WO<sub>6</sub> (weight ratio: 70:30), while pyrolysis under O<sub>2</sub> yielded the same tungstates, but with a different weight ratio (10:90). This can be a new process for Ca<sub>3</sub>WO<sub>6</sub>, which, due to its ordered double perovskite structure, is ideal for doping with metal ions, at relatively low temperature (800 °C vs. >1000 °C in the literature).

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