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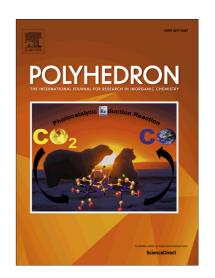
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Effect of solvents and metal ions on the structural diversity of coordination polymers based on a dipyridylamide ligand: Construction, fluorescent and photocatalytic properties

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Abstract

Five new Ni^{II}/Cd^{II}/Co^{II} coordination polymers (CPs), namely [Ni(3-pna)(mip)]·H₂O (1), $[Ni(3-pna)(mip)(H_2O)]\cdot 4H_2O$ (2), $[Cd_2(3-pna)(mip)_2(H_2O)] \cdot 2H_2O$ $[Cd(3-pna)(mip)]_2 \cdot 3H_2O$ **(4)** and $[Co(3-pna)(mip)] \cdot H_2O$ **(5)** [3-pna 3-pyridylnicotinamide, H_2 mip = 5-methylisophthalic acid, have been hydrothermally or solvothermally synthesized. These complexes were characterized by IR spectroscopy, thermal analysis and single-crystal X-ray diffraction. Complexes 1-5, based on the same dipyridylamide (3-pna) and dicarboxylate (mip) ligands but using different solvent systems and metal ions, exhibit various structures. Complexes 1 and 5 feature isostructural 2D networks, which are very similar to that of 4. Their structures contain $[M_2(COO)_2-(3-pna)_2]_n$ (M = Ni, Cd or Co) ladder-like double chains. Complex 2 displays a 2D 4-connected grid structure consisting of 1D [Ni-3-pna]_n and [Ni-mip]_n chains. Complex 3 presents a new 3D framework constructed from $Cd_4(COO)_4$ subunits and μ_2 -bridging 3-pna ligands. The versatile structures reveal the remarkable impact of the solvent systems on the architectures. Furthermore, the fluorescent behaviors of 3-4 and the photocatalytic properties of 1-5 under UV irradiation have been studied.

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