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Crystal structure, DNA binding, cleavage, antioxidant and antibacterial studies of Cu(II), Ni(II) and Co(III) complexes with 2-((furan-2-yl)methylimino)methyl)-6-ethoxyphenol Schiff base

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PII: S0022-2860(18)30143-1

DOI: 10.1016/j.molstruc.2018.02.004

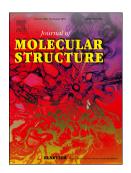
Reference: MOLSTR 24826

To appear in: Journal of Molecular Structure

Received Date: 1 November 2017
Revised Date: 30 January 2018
Accepted Date: 1 February 2018

Please cite this article as: K. Venkateswarlu, M.P. Kumar, A. Rambabu, N. Vamsikrishna, S. Daravath, K. Rangan, Shivaraj, Crystal structure, DNA binding, cleavage, antioxidant and antibacterial studies of Cu(II), Ni(II) and Co(III) complexes with 2-((furan-2-yl)methylimino)methyl)-6-ethoxyphenol Schiff base, *Journal of Molecular Structure* (2018), doi: 10.1016/j.molstruc.2018.02.004.

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ACCEPTED MANUSCRIPT

Crystal structure, DNA binding, cleavage, antioxidant and antibacterial studies of Cu(II), Ni(II) and Co(III) complexes with 2-((furan-2-yl)methylimino)methyl)-6-ethoxyphenol Schiff base

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Abstract

Three novel binary metal complexes; **1** [Cu(L)₂], **2** [Ni(L)₂] and **3** [Co(L)₃] where, **L** (2-(((furan-2-yl) methylimino)methyl)-6-ethoxyphenol, $C_{14}H_{15}NO_3$), were synthesized and characterized by various spectral techniques. Based on spectral studies square planar geometry is assigned for Cu(II) and Ni(II) complexes, whereas Co(III) owned octahedral geometry. Ligand, [Cu(L)₂] and [Ni(L)₂] are crystallized and found to be monoclinic crystal systems. CT-DNA absorption binding studies revealed that the complexes show good binding propensity ($K_b = 5.02 \times 10^3 \text{ M}^{-1}$, $2.77 \times 10^3 \text{ M}^{-1}$, $1.63 \times 10^4 \text{ M}^{-1}$ for **1**, **2** and **3** respectively). The role of these complexes in the oxidative and photolytic cleavage of supercoiled pBR322 DNA was studied and found that the complexes cleave the PBR322 DNA effectively. The catalytic ability of **1**, **2** and **3** follows the order: **3** > **1** > **2**. Antioxidant studies of the new complexes revealed that they exhibit significant antioxidant activity against DPPH radical. The Schiff base and its metal complexes have been screened for antibacterial studies by Minimum Inhibitory Concentration method. It is observed that all metal complexes showed more activity than free ligand.

Keywords: Metal complex; Crystal structure; DNA interaction; Antioxidant activity; Antibacterial activity.

1. Introduction

It is well known that deoxyribonucleic acid (DNA) is an important genetic substance in organisms. Errors in gene expression can often cause disease and play a secondary role in the outcome and severity of human disease [1]. DNA is the primary molecular target for many of the drugs that are used in cancer. The transition metal complexes of Schiff bases have tendency to interact with DNA because of their cationic character and three dimensional structural profiles

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