

Trans- and cis-2-phenylindole platinum(II) complexes as cytotoxic agents against human breast adenocarcinoma cell lines



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- Trans- and cis-isomers of [Pt(L)Cl₂(DMSO)] complexes with a new 3-methoxyimino-2-phenyl-3H-indole ligand have been prepared and characterized by X-ray diffraction.
- Conclusive proofs of the key role of the DMSO ligand and its location in the coordination sphere of the Pt(II) and on the crystal architecture are presented.
- The study of the antitumoral activity of cis- and trans- [Pt(L)Cl₂(DMSO)] shows that most of them are potent cytotoxic agents.
- The influence of the relative disposition of the Cl[−] ligands in the cis- and trans-isomers on their cytotoxic activities is discussed.
- Compounds 4b and 5a are more cytotoxic than cisplatin in MDA-MB231 and MCF-7 breast cancer lines.

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The synthesis and characterization of the new 2-phenylindole derivative: C₈H₃N-2-C₆H₅-3NOMe-5OMe (**3c**) and the trans- and cis-isomers of [Pt(**3c**)Cl₂(DMSO)] complexes (**4c** and **5c**, respectively) are described. The crystal structures of **4c**·CH₂Cl₂ and **5c** confirm: (a) the existence of a Pt–N_{indole} bond, (b) the relative arrangement of the Cl[−] ligands [*trans*- (in **4c**) or *cis*- (in **5c**)] and (c) the *anti*-(*E*) configuration of the oxime. The cytotoxic assessment of C₈H₃N-2-(C₆H₄-4'R¹)-3NOMe-5R² [with R¹ = R² = H (**3a**); R¹ = Cl, R² = H (**3b**) and R¹ = H, R² = OMe (**3c**)] and the geometrical isomers of [Pt(L)Cl₂(DMSO)] with L = **3a–3c** [*trans*- (**4a–4c**) and *cis*- (**5a–5c**), respectively] against human breast adenocarcinoma cell lines (MDA-MB231 and MCF-7) is also reported and reveals that all the platinum(II) complexes (except **4a**) are more cytotoxic than cisplatin in front of the MCF7 cell line. Electrophoretic DNA migration studies of the synthesized compounds in the absence and in the presence of topoisomerase-I have been performed, in order to get further insights into their mechanism of action.

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1. Introduction

Heterocycles have important applications as fine and bulk chemicals, agrochemicals and participate in the structure of many pharmaceutical agents and biomolecules [1]. They are also valuable ligands in coordination and organometallic chemistry [2,3], in which the nature of the transition metal, its oxidation state,

the electronic and/or steric features of heterocyclic ligands and their relative disposition modify their physical and chemical properties and have direct implications on their applications in chemistry, medicine, biology and materials science [1–3].

Moreover, the interest in novel *cis*- and *trans*-isomers of [Pt(L¹)(L²)Cl₂] with L¹ or L² = heterocycles with N donor atoms has increased considerably in recent years mainly due to their potential use in cancer therapy and biomedicine [4–11]. Platinum(II) complexes [Pt(L¹)(L²)Cl₂] or [Pt(L¹)₂Cl₂] with L¹ derived from pyrazole, imidazole, benzimidazole or purine and pyrimidine bases of DNA have been reported [8–10].

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On the other hand, 2-phenylindole derivatives have attracted great interest since their introduction by von Angerer in 1983 [12]. The 2-phenylindole-3-carboxaldehydes (**1**) (Fig. 1) proved to exert an antimitotic activity in human breast cancer cells by inhibition of tubulin polymerization [13]. Related derivatives containing oximes, methylamines, propanedinitriles and hydrazones [14] or the ferrocenylmethyl unit (**2** in Fig. 1) on position 3 have been also investigated [15].

Despite the interest of exploring the potential synergic effect induced by the presence of phenylindole ligands in *cis*- or *trans*-isomers of $[\text{Pt}(\text{L}^1)(\text{L}^2)\text{Cl}_2]$, platinum(II) complexes with indolyl frameworks are extremely scarce [16]. Quite recently we have developed synthetic procedures to achieve the two 3-methoxyimino-2-phenyl-3*H*-indoles: $\text{C}_8\text{H}_4\text{N}-2-(\text{C}_6\text{H}_4-4'\text{R}^1)-3\text{NOMe}$ with $\text{R}^1 = \text{H}$ (**3a**) or Cl (**3b**) (Fig. 1) [17] and the *trans*- (**4a** or **4b**) and *cis*- (**5a** and **5c**) isomers of compounds $[\text{Pt}(\text{L}^3)\text{Cl}_2(\text{DMSO})]$ (with $\text{L}^3 = \text{3a}$ or **3b**, Fig. 1) [18]. In this paper we report the new 2-phenylindole derivative, $\text{C}_8\text{H}_3\text{N}-2-(\text{C}_6\text{H}_4-4'\text{R}^1)-3\text{NOMe}-5\text{R}^2$ with $\text{R}^1 = \text{H}$, $\text{R}^2 = \text{OMe}$ (**3c**) (Fig. 1 and Scheme 1) and the two geometrical isomers of $[\text{Pt}(\text{3c})\text{Cl}_2(\text{DMSO})]$. A comparative study of the cytotoxic activity of the free ligands (**3a–3c**) and the *trans*- (**4a–4c**) and *cis*- (**5a–5c**) isomers of the Pt(II) complexes in two human breast adenocarcinoma cell lines (MDA-MB231 and MCF-7) is also reported, together with a study of their effect on the electrophoretic mobility of pBluescript SK⁺ DNA and their behavior in solution.

2. Experimental

2.1. Materials and methods

Cis- $[\text{PtCl}_2(\text{DMSO})_2]$, the 3-hydroxyimino-5-methoxy-2-phenyl-3*H*-indole, ligands **3a**, **3b** and complexes **4a**, **4b**, **5a** and **5b**, were prepared as reported previously [17–19]; 2-phenylindole and cetyltrimethylammonium bromide (CTAB) were obtained from commercial sources and used as received. Solvents were dried

and distilled using the standard procedures [20]; except methanol which was of HPLC grade [21]. Elemental analyses were carried out at the *Serveis Científic-Tècnics* (Universitat de Barcelona). Mass spectra (ESI⁺) were performed at the *Servei d'Espectrometria de Masses* (Universitat de Barcelona), using a VG-Quattro instrument. Infrared spectra were obtained with a Nicolet 400FTIR, Nicolet 520 or 5700 instrument using KBr pellets; while far IR-spectra of **4c** and **5c** (in the range 200–400 cm^{−1}) were registered with a Bomem-DA3 instrument using polyethylene discs. The ultraviolet–visible (UV–vis) spectra of **3c–5c** in CH_2Cl_2 were recorded at 298 K with a Varian Cary 100 spectrophotometer. Routine ¹H and ¹³C{¹H} NMR spectra were registered with a Mercury-400 MHz instrument. High resolution ¹H NMR spectra and the two dimensional [¹H–¹H]-NOESY, COSY and [¹H–¹³C] HSQC and HMBC experiments were registered with a Varian Inova 500 and Bruker DMX-500 MHz instruments at 298 K. The ¹⁹⁵Pt{¹H} NMR spectra of **4c** and **5c** were obtained with a Bruker DRX-250 MHz instrument. The chemical shifts (δ) are given in ppm, the coupling constants (*J*) in Hz and the labeling of the atoms corresponds to the pattern shown in Scheme 1. The splitting of proton resonances in the reported ¹H NMR spectra is defined as s = singlet, d = doublet, dd = doublet of doublet, m = multiplet. Except where quoted, the solvent used for the NMR experiments was CDCl_3 (99.9%) and the references were SiMe_4 [for ¹H and ¹³C] and $\text{H}_2[\text{PtCl}_6]$ for ¹⁹⁵Pt{¹H} NMR.

2.2. Preparation of the compounds

2.2.1. Synthesis of $\text{C}_8\text{H}_3\text{N}-2-\text{C}_6\text{H}_5-3-\text{NOMe}-5-\text{OMe}$ (**3c**)

To a solution containing 3-hydroxyimino-5-methoxy-2-phenyl-3*H*-indole (666 mg, 3.0×10^{-3} mol), CTAB (690 mg, 1.9×10^{-3} mol) and NaOH (12 mL of a 40% solution) in CH_2Cl_2 (125 mL), methyl iodide (2.28 g, 1 mL , 16×10^{-3} mol) was added. The reaction mixture was stirred at room temperature for 24 h. After this period water (75 mL) was added and the resulting solution was transferred to a separating funnel. The organic layer was separated

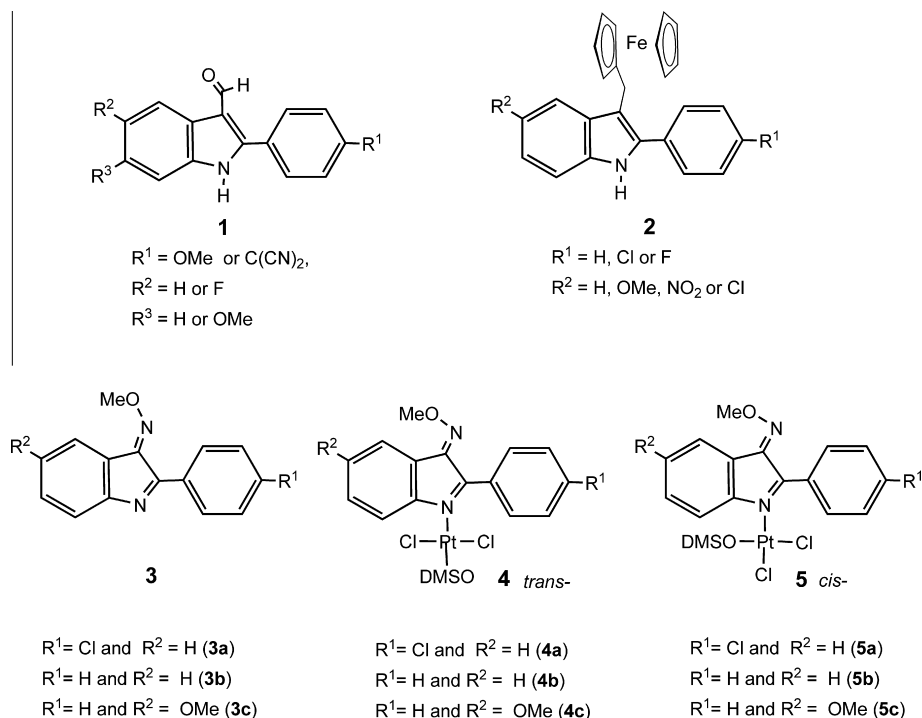


Fig. 1. General formulae of different types of 2-phenylindole derivatives with antitumoral activity (**1** and **2**) [12–15], the functionalized compounds **3a** and **3b** and their platinum(II) complexes (**4a**, **4b**, **5a**, **5b**) reported recently [18].

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