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# Application of indoline dyes attached with strongly electronwithdrawing carboxylated indan-1,3-dione analogues linked with a hexylthiophene ring to dye-sensitized solar cells



Masaki Matsui <sup>a, \*</sup>, Nagisa Tanaka <sup>a</sup>, Kazumasa Funabiki <sup>a</sup>, Yuki Haishima <sup>a</sup>, Kazuhiro Manseki <sup>a</sup>, Jiye Jin <sup>b</sup>, Yukiko Inoue <sup>c</sup>, Shinji Higashijima <sup>c</sup>, Yasuhiro Kubota <sup>a</sup>

<sup>a</sup> Department of Chemistry and Biomolecular Science, Faculty of Engineering, Gifu University, 1-1 Yanagido, Gifu, 501-1193, Japan
<sup>b</sup> Department of Chemistry, Faculty of Science, Shinshu University, 3-1-1-Asahi, Matsumoto, Nagano, 390-8621, Japan

<sup>c</sup> Chemicrea Inc., Ohtsurugi Shimogawa, Izumi-machi, Iwaki, Fukushima, 971-8183, Japan

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# 1. Introduction

Indoline dyes are one of the most efficient organic sensitizers in dye-sensitized solar cells (DSSCs).<sup>1</sup> Therefore, much effort has been devoted to obtain bathochromic indoline dyes by changing the pull moiety. Indoline dyes attached with cyanoacrylic acid,<sup>2</sup> single,<sup>3</sup> double,<sup>4</sup> and triple rhodanine acetic acid<sup>5</sup> pull moiety have been prepared and evaluated as the sensitizers. The first absorption maximum ( $\lambda_{max}$ ) of triple rhodanine indoline dye was observed at 572 nm in chloroform.<sup>5</sup> Those of double rhodanine indoline dyes linked with mono-, di-, and trithienyl spacers were observed at 568, 564, and 550 nm, respectively.<sup>6</sup> Thus, these indoline dyes are not bathochromic as expected. In 2012, carboxylated indan-1,3dione was used as an electron acceptor anchor moiety in DSSCs.<sup>7</sup> Recently, an indoline dye attached with a 5-carboxyindan-1,3dione pull anchor moiety linked with a thienyl ring has been reported to exhibit the  $\lambda_{max}$  at around 600 nm in chloroform and to show higher conversion efficiency than **D205**.<sup>8</sup> We consider that

#### ABSTRACT

An indoline dye attached with a carboxylated indan-1,3-dione moiety linked with a hexylthiophene ring exhibited the highest conversion efficiency among six analogues and **D205**. This result comes from the bathochromic UV–vis absorption band, suitable energy levels, good stability for redox process, and sufficiently long excited-state lifetime.

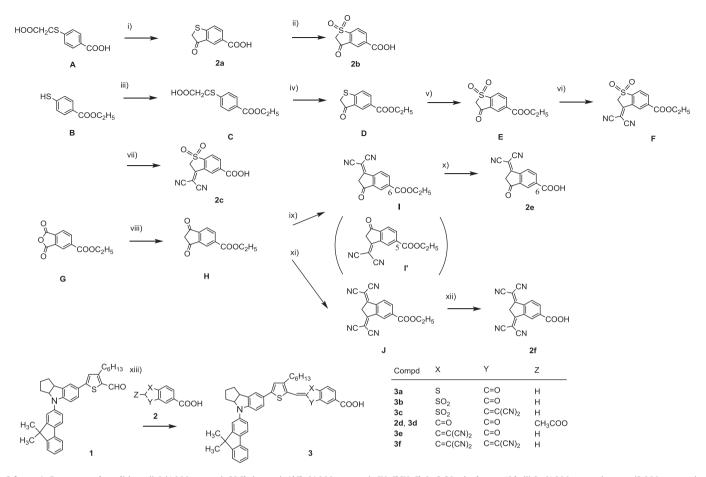
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the indoline dyes having more electron-withdrawing pull moiety can show more bathochromic absorption bands to exhibit higher conversion efficiency. We report here the synthesis and photoelectrochemical properties of novel indoline dyes with a carboxylated benzo[*b*]thiophene-3(2*H*)one, benzo[*b*]thiophene-3(2*H*)one 1,1-dioxide, benzo[*b*]thiophene-3(2*H*)dicyanovinilidene 1,1dioxide, indan-1,3-dione, indan-1-dicyanovinilidene-3-one, and indan-1,3-bis(dicyanovinilidene) anchor pull moiety linked with a hexylthiophene ring in DSSCs.

# 2. Results and discussion

### 2.1. Synthesis

Novel indoline dyes **3a**, **3b**, **3c**, **3d**, **3e** and **3f** were synthesized as shown in Scheme 1. Compound **2a** was obtained by the intramolecular Friedel–Crafts alkanoyl reaction of **A**.<sup>9</sup> Compound **2b** was provided by the oxidation of **2a**. Compound **B**, prepared by the reaction of 4-mercaptobenzoic acid with ethanol in the presence of concentrated sulfuric acid, was allowed to react with chloroacetic acid to afford a carboxymethylthio derivative C, followed by the



Scheme 1. Reagents and conditions: i) A (1.0 M amount), SOCl<sub>2</sub> (excess), AlCl<sub>3</sub> (4.0 M amounts), CH<sub>2</sub>ClCH<sub>2</sub>Cl, 0 °C, 20 min then rt, 18 h, ii) **2a** (1.0 M amount), oxone (2.0 M amounts), 1,4-dioxane-water (5/1), rt, 3 h, iii) CH<sub>2</sub>ClCOOH (1.5 M amounts), K<sub>2</sub>CO<sub>3</sub> (2.0 M amounts), MeCN, rt, 3 h, then B (1.0 M amount), 60 °C, 3 h, iv) C (1.0 M amount), SOCl<sub>2</sub> (5 M amounts), reflux, 30 min, then, AlCl<sub>3</sub> (4.0 M amounts), CH<sub>2</sub>ClCH<sub>2</sub>Cl, 0 °C, 30 min, then rt, 21 h, v) D (1.0 M amount), oxone (2.0 M amounts), 1,4-dioxane-water (5/1), rt, 23 h, vi) E (1.0 M amount), CH<sub>2</sub>(CN)<sub>2</sub> (2.5 M amounts), AcONa (1.0 M amount), EtOH, reflux, 17 h, vii) F (1.0 M amount), Me<sub>3</sub>SiOK (1.0 M amount), THF, rt, 21 h, then 10% HCl, viii) G (1.0 M amount), AcCH<sub>2</sub>COOEt (1.0 M amount), EtOH, reflux, 17 h, vii) F (1.0 M amount), CH<sub>2</sub>(CN)<sub>2</sub> (3.0 M amounts), AcONa (1.0 M amount), EtOH, reflux, 1 h, x) I (1.0 M amount), Me<sub>3</sub>SiOK (3.0 M amount), EtOH, reflux, 1 h, x) I (1.0 M amount), Me<sub>3</sub>SiOK (3.0 M amounts), rt, 24 h, xi) H (1.0 M amount), CH<sub>2</sub>(CN)<sub>2</sub> (4.0 M amount), AcONa (1.0 M amount), EtOH, reflux, 17 h, xiii) J (1.0 M amount), H<sub>2</sub>O (excess), LiOH (7.0 M amounts), rt, 24 h, xiii) I (1.0 M amount), CH<sub>2</sub>(CN)<sub>2</sub> (4.0 M amounts), AcON<sub>4</sub> (1.0 M amount), 2**d**, 2**e**, 2**f** (1.0 M amount), AcOH, 55–100 °C, 3–4 h, rt

intramolecular Friedel–Crafts alkanoyl reaction to yield a sulfide **D**. Compound **D** was oxidized to provide a sulfone **E**, which was allowed to react with malononitrile followed by the hydrolysis of the ester moiety to give a dicyanovinilidene derivative **2c**. Compound **G** was allowed to react with ethyl acetoacetate to give a compound **H**, which was reacted with 3.0 M amounts of malononitrile to provide both the mono(dicyanovinilidene) **I** and **I'** (51%) and bis(dicyanovinilidene) derivatives **J** (28%). The mono and bis derivatives were separated by column chromatography. For the mono derivative, both the 6- and 5-ethoxycarbonyl *regio* isomers **I** and **I'** were formed. These isomers were not separated by chromatography due to the same R<sub>f</sub> value (SiO<sub>2</sub> TLC, R<sub>f</sub> = 0.35, CHCl<sub>3</sub>: MeOH = 3: 1). However, when the mixture was recrystallized from ethanol, only the 6-ethoxycarbonyl isomer **I** was obtained as a crystalline form in a 28% yield.

The heteronuclear multiple bond coherence (HMBC) spectrum of **I** is shown in Fig. 1. The spectrum indicates that the H<sup>5</sup> peak correlates with the C<sup>7</sup>, C<sup>5</sup>, and COO peaks. The H<sup>7</sup> peak correlates with the C<sup>5</sup>, C<sup>3a</sup>, and COO peaks. The H<sup>4</sup> peak correlates with the C<sup>6</sup>, C<sup>7a</sup>, and C<sup>3</sup> peaks. Thus, compound **I** was identified as ethyl indan-3-dicyanovinilidene-1-one-6-carboxylate. The hydrolysis of ester moiety of **I** afforded the carboxyl derivative **2e**. Compound **H** was allowed to react with 4.0 M amounts of malononitrile to give **J** in a

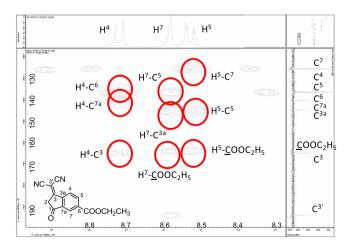


Fig. 1. HMBC spectrum of I.

78% yield. The hydrolysis of the ester moiety of **J** provided a carboxylic acid **2f**. Finally, an indoline carboaldehyde **1** was allowed to react with **2a**, **2b**, **2c**, **2d**, **2e**, and **2f** to afford **3a**, **3b**, **3c**, **3d**, **3e**, and

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