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One-pot synthesis of quinazolin-4(3*H*)-ones and fused quinazolinones by a palladium-catalyzed domino process



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ABSTRACT

An efficient one-pot synthesis of quinazolin-4(3*H*)-ones, benzoimidazo[2,1-*b*]quinazolin-12(6*H*)-ones and imidazo[2,1-*b*]quinazolin-5(1*H*)-ones via a palladium-catalyzed domino process has been developed. The Pd-catalyzed reactions of 2-azidobenzamides **1** with isocyanides **2** produced quinazolin-4(3*H*)-ones **4** at room temperature by a domino Pd-catalyzed cross-coupling/carbodiimide-mediated cyclization. However, as 2-azido-*N*-(2-bromophenyl)benzamides **1** were used under heating condition in the presence of Cs₂CO₃, the benzoimidazo[2,1-*b*]quinazolin-12(6*H*)-ones **5** were directly obtained by twice Pd-catalyzed domino cyclization. A domino reogioselective 5-*exo-dig* intramolecular cyclization reaction of alkynyl-containing azides **6** with isocyanides **2** generated imidazo[2,1-*b*]quinazolin-5(1*H*)-ones **9** in 74 –93% yields in the presence of catalyst Pd(PPh₃)₄ and K₂CO₃.

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1. Introduction

Quinazolines and ring-fused derivatives have emerged as wellknown class of heterocyclic motifs showing a variety of potential biological and pharmaceutical activities such as antimicrobial, ¹ antifungal,² antibacterial,³ nitric oxide synthase (NOS) inhibitive,⁴ transient receptor potential A1 (TRPA1) antagonistical⁵ and antitumor activities.⁶ The quinazolinone skeleton have also been found in some natural products and drugs (Fig. 1).7 Great attention has been paid to them by the organic chemists owing to their widespread biological activities and remarkable chemical structures. Over the past decades, a number of efficient methods for the preparation of quinazolinone skeleton from a range of starting materials have been developed. Classical synthetic routes to the quinazolinone include the utilization of anthranilic acids⁸ or its derivatives, such as anthranilamides, isatoic anhydrides and 2aminobenzonitriles, ¹¹ the microwave-assisted dehydrative cyclization of diamides, ¹² the transition-metal catalyzed reactions, ¹³ and the aza-Wittig reactions. 14 However, despite of their many advantages, these procedures suffer from some drawbacks including high temperature, acid- or base-sensitive substrates, multi-step process and low atom efficiency. Therefore, the development of new and

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step-economic process is still highly desirable.

Carbodiimides are important chemical reagents in organic chemistry and they have been utilized as intermediates or precursors for some heterocycles. 15 Recently an elegant work for the formation of unsymmetric carbodiimides utilizing palladiumcatalyzed cross-coupling reaction of azides with isocyanides was reported by Zhang and co-workers. 16 Inspired by their work, we speculated that if 2-azidobenzamides were used to react with isocyanides in palladium catalyst, quinazolinones would be generated through further cyclization of the carbodiimide intermediate. As a continuation of our interest in developing new synthetic protocols for various heterocycles via isocyanide chemistry, ¹⁷ herein, we wish to report a new strategy for one-pot preparation of quinazolin-4(3H)-ones, benzoimidazo[2,1-b]quinazolin-12(6H)ones and imidazo[2,1-b]quinazolin-5(1H)-ones via a Pd-catalyzed domino process.

2. Results and discussion

The 2-azido-5-chloro-N-p-tolylbenzamide $\mathbf{1a}$ ($R^1 = 5$ -Cl, $R^2 = 4$ -MeC₆H₄, 1 equiv) was initially selected to react with n-butylisocyanide $\mathbf{2a}$ ($R^3 = n$ -Bu, 1 equiv) in DMF in the presence of catalyst Pd(PPh₃)₄ (0.05 equiv). The palladium-catalyzed cross-coupling reaction was carried out smoothly at room temperature to produce directly the quinazolin-4(3H)-one product $\mathbf{4a}$ in good yield (81%). The above one-pot operation was then applied for various 2-

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Fig. 1. Some quinazolinone natural products and drugs.

azidobenzamides **1** and isocyanides **2** (Table 1). All of the reactions were carried out smoothly to give the corresponding quinazolin-4(3*H*)-ones **4**, and good yields (72–86%) were obtained with different substituents on the reactants. As indicated in Table 1, good

yields were reached no matter the R^2 substituent of the azide $\mathbf{1}$ was an alkyl group (compounds $\mathbf{4q}$ and $\mathbf{4r}$) or an aryl group with substituents (4-Cl, 4-Br, 2-Br, 2-I, 2-CH₃, 4-CH₃ and 4-OCH₃) on the benzene ring (compounds $\mathbf{4a-4p}$). Satisfactory yields were also obtained as various alkyl ($R^3 = t$ -Bu, cyclohexyl, n-Bu) or aryl ($R^3 = 4$ -BrC₆H₄, 4-ClC₆H₄, 4-MeC₆H₄) isocyanides $\mathbf{2}$ were utilized. It's noteworthy that good yields (74–80%) of the quinazolin-4(3H)-ones (compounds $\mathbf{4d}$, $\mathbf{4d}$, $\mathbf{4q}$ and $\mathbf{4r}$) were directly obtained at room temperature even when R^3 is the steric t-Bu group. The formation of quinazolin-4(3H)-ones $\mathbf{4d}$ can be viewed as an initial palladium-catalyzed cross-coupling reaction between 2-azidobenzamide $\mathbf{1d}$ and isocyanide $\mathbf{2d}$ to create the carbodiimide $\mathbf{3d}$ as highly reactive intermediate. Further intramolecular nucleophilic attack of the amide group on the carbodiimide $\mathbf{3d}$ produces quinazolin-4(3H)-

Table 1 Preparation of quinazolin-4(3*H*)-ones **4**.

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