



Rh(III)-catalyzed, 1,2,3-triazole-assisted directed C–H coupling with diazo diphosphonates

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ABSTRACT

A mild and efficient procedure was developed for the [Cp*Rh(III)]-catalyzed, 1,2,3-triazole directed C–H coupling with diazomethylene-diphosphonates. This protocol provided a step- and atom-economical protocol for C–C bond formation and led to structurally diverse 2-(1,2,3-triazol-2-yl)benzyl diphosphonates in good to excellent yields.

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Introduction

Over the last 20 years, transition metal-catalyzed C–H bond activation has emerged as an attractive strategy in organic synthesis which avoids the multi-step pre-activation of starting materials and minimizes the production of unwanted by-products [1–6]. Typically, C–H bond activation requires a metal-coordinating functional group to control the regioselectivity of transition metal insertion into a C–H bond [7]. Due to an array of structural properties and the unique pharmacophore features of 1,2,3-triazoles [8–11], our group became interested in 1,2,3-triazole directed C–H activation. For example, the ruthenium-catalyzed, 1,2,3-triazole directed intermolecular C–H amidation of arenes with sulfonylazides; the palladium-catalyzed, 1,2,3-triazole directed C–H ethoxycarbonylation of 2-aryl-1,2,3-triazoles with diethyl azodicarboxylate; the cobalt(III)-catalyzed, 1,2,3-triazole-assisted C–H amidation of arenes with dioxazolones; and the rhodium-catalyzed, 1,2,3-triazole-assisted *ortho*-cyanation of 2-aryl-1,2,3-triazoles with *N*-cyano-*N*-phenyl-*p*-toluenesulfonamide (Fig. 1) [12–15].

Diazo compounds have been widely used as carbene precursors in C–H activation. For example, in 2012 Yu and co-workers reported the Rh(III)-catalyzed, *ortho*-alkylation of directing-group-containing arene C–H bonds with diazomalones to pre-

pare aromatic malonic acids [16]. More recently, the groups of Glorius [17], Yao [18], Zhu [19], Wang [20], Li [21], and others [22–33] expanded and enriched this C–H functionalization method using different diazo compounds to construct heterocycles (Scheme 1a). In addition, we recently used diazomethylene-diphosphonates as carbene precursor reagents to insert into the O–H bond of carboxylic acids, which can be used to synthesize a bone-targeting prodrug [34], and to couple with 2-phenylpyridines generating aromatic bisphosphonates, which were identified as β -lactamase inhibitors by computational and experimental assays [35]. With the aim to enlarge the library of aromatic bisphosphonates, which may provide more possibilities to identify hit/lead compounds for clinically relevant β -lactamases as well as other protein targets, plus our interest in the C–H functionalization of 2-aryl-1,2,3-triazoles, we herein report the Rh(III)-catalyzed, 1,2,3-triazole-assisted directed C–H coupling with diazo diphosphonates (Scheme 1b).

Results and discussion

Initial experiments were carried out using 2-(*m*-tolyl)-2H-1,2,3-triazole (**1a**, 0.2 mmol) and tetraethyl diazomethylene-diphosphonate (**2a**, 0.24 mmol) in the presence of [Cp*RhCl₂]₂ (5.0 mol%) and AgSbF₆ (10.0 mol%) at 80 °C in 1,2-dichloroethane (DCE, 2 mL) for 24 h. As expected, the desired target product **3aa** was obtained in 66% yield under these initial conditions (Table 1, entry 1). Next, the effect of the metal catalyst was investigated (Entries 2–8). Unfortunately, no coupled products were observed for the tested

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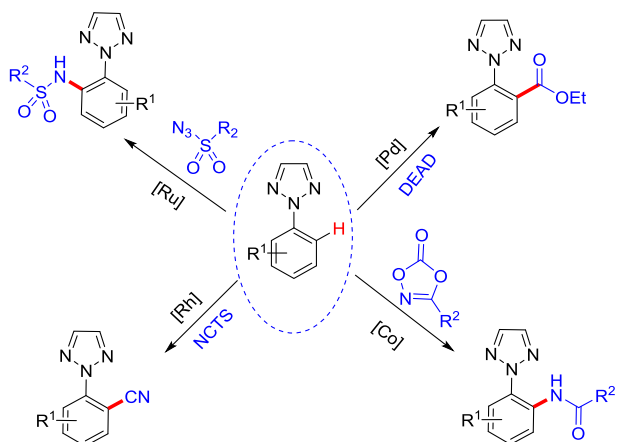
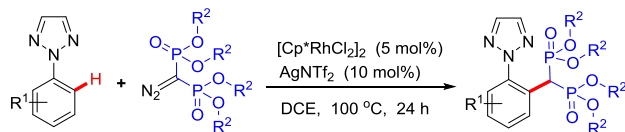


Fig. 1. Our previous work on 2-aryl-1,2,3-triazoles directed C–H functionalization [12–15].

a) Previous work regarding the direct C–H bond coupling with carbenoids



b) This work



Scheme 1. C–H coupling with diazo compounds.

catalysts, except for $[\text{Cp}^*\text{IrCl}_2]_2$ which gave a moderate yield of 61% (Entry 8). Variation of the Ag salt revealed that AgNTf_2 was the most effective with a yield of 72% (Entries 9–12). Next, we focused our attention on the solvent. 1,2,3-Trichloropropane (TCP), CH_3CN , EtOH, THF, MeOH and hexafluoroisopropyl-alcohol (HFIP), which are typically used in direct C–H bond functionalization, were less effective in this transformation than DCE (Entries 13–18). Finally, an excellent isolated yield of 91% was obtained by increasing the temperature to 100 °C (Entries 19, 20). Interestingly, the $[\text{Cp}^*\text{RhCl}_2]_2$ catalyst also produced **3aa** with an excellent isolated yield of 90% (Entry 21). Accordingly, the optimized reaction conditions are 5 mol% $[\text{Cp}^*\text{RhCl}_2]_2$, 10 mol% AgNTf_2 in DCE at 100 °C for 24 h.

With the optimized reaction conditions in hand, we set out to explore the scope and limitation of the Rh(III)-catalyzed *ortho*-selective C–H carbenoid insertion of 2-aryl-1,2,3-triazoles (Scheme 2). Initially, *ortho*-substituted arenes were investigated. Amide substituted arenes underwent the coupling reaction smoothly and the corresponding products were obtained in good to excellent yields (Scheme 23ba–da). Notably, halogen-substituted arenes (Scheme 23ea–ha) gave higher yields than those with electron-withdrawing groups (EWGs) such as CN and NO_2 (Scheme 23ea–ha). In contrast to *ortho*-substituted substrates, *meta*-substituted arenes gave the corresponding products in moderate to excellent yields, including those with EWGs such as NO_2 and COOMe (Scheme 23aa, 3ia–na). These results indicate that the coupling efficiency is to some extent influenced by the electron density of the aromatic ring and by steric hindrance.

According to our previous work regarding the sulfonamidation of 2-aryl-1,2,3-triazoles with sulfonyl azides [14], non-/para-substituted 2-aryl-1,2,3-triazoles were treated with 2.4 equivalents of **2a**. All substrates were dialkylated while no mono-alkylated products were observed (Scheme 23oa–va). Furthermore, naphthalene substituted 1,2,3-triazole (**1w**) and 1*H*-indole substituted

Table 1
Reaction optimization.^a

Entry	Catalyst	Ag salt	Solvent	T [°C]	Yield ^b
1	$[\text{Cp}^*\text{RhCl}_2]_2$	AgSbF_6	DCE	80	66%
2	$\text{Cu}(\text{OAc})_2$	AgSbF_6	DCE	80	N.R.
3	$\text{Pd}(\text{OAc})_2$	AgSbF_6	DCE	80	N.R.
4	$[\text{Ru}(p\text{-cymene})\text{Cl}_2]_2$	AgSbF_6	DCE	80	N.R.
5	$[\text{Cp}^*\text{Co}(\text{CH}_3\text{CN})_3][\text{SbF}_6]_2$	–	DCE	80	N.R.
6	$[\text{Cp}^*\text{Co}(\text{CO})]\text{I}_2$	AgSbF_6	DCE	80	N.R.
7	$[\text{Cp}^*\text{Rh}(\text{CH}_3\text{CN})_3][\text{SbF}_6]_2$	–	DCE	80	56%
8	$[\text{Cp}^*\text{IrCl}_2]_2$	AgSbF_6	DCE	80	61%
9	$[\text{Cp}^*\text{RhCl}_2]_2$	AgNTf_2	DCE	80	72%
10	$[\text{Cp}^*\text{RhCl}_2]_2$	AgOTf	DCE	80	52%
11	$[\text{Cp}^*\text{RhCl}_2]_2$	AgOAc	DCE	80	N.R.
12	$[\text{Cp}^*\text{RhCl}_2]_2$	AgTFA	DCE	80	N.R.
13	$[\text{Cp}^*\text{RhCl}_2]_2$	AgNTf_2	TCP	80	51%
14	$[\text{Cp}^*\text{RhCl}_2]_2$	AgNTf_2	CH_3CN	80	N.R.
15	$[\text{Cp}^*\text{RhCl}_2]_2$	AgNTf_2	EtOH	80	23%
16	$[\text{Cp}^*\text{RhCl}_2]_2$	AgNTf_2	THF	80	N.R.
17	$[\text{Cp}^*\text{RhCl}_2]_2$	AgNTf_2	MeOH	80	42%
18	$[\text{Cp}^*\text{RhCl}_2]_2$	AgNTf_2	HFIP	80	55%
19	$[\text{Cp}^*\text{RhCl}_2]_2$	AgNTf_2	DCE	100	91%
20	$[\text{Cp}^*\text{RhCl}_2]_2$	AgNTf_2	DCE	120	88%
21	$[\text{Cp}^*\text{IrCl}_2]_2$	AgNTf_2	DCE	100	90%

^a Reagents and conditions: **1** (0.2 mmol), **2a** (0.24 mmol), catalyst (5 mol%), Ag salt (10 mol%), solvent (2 mL), 24 h.

^b Isolated yield.

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