



Understanding the tetrazole ring cleavage reaction with hydrazines: Structural determination and mechanistic insight

Olga Ya. Shyyka^{a,*}, Nazariy T. Pokhodylo^a, Yurii I. Slyvka^b, Evgeny A. Goreshnik^c, Mykola D. Obushak^a

^a Department of Organic Chemistry, Ivan Franko National University of Lviv, Kyryla i Mefodiya St. 6, Lviv 79005, Ukraine

^b Department of Inorganic Chemistry, Ivan Franko National University of Lviv, Kyryla i Mefodiya St. 6, Lviv 79005, Ukraine

^c Department of Inorganic Chemistry and Technology, Jožef Stefan Institute, Jamova 39, SI-1000 Ljubljana, Slovenia

ARTICLE INFO

Article history:

Received 10 December 2017

Revised 1 February 2018

Accepted 6 February 2018

Available online 7 February 2018

Keywords:

1*H*-Tetrazole

Hydrazines

Thieno[2,3-*d*]pyrimidin-4(3*H*)-ones

Thieno[3,2-*d*]pyrimidin-4(3*H*)-ones

[1]Benzofuro[3,2-*d*]pyrimidin-4(3*H*)-one

Regioselectivity

ABSTRACT

A distinction in the behaviour of several hydrazines as *N*-nucleophiles in the recently developed one-pot method for pyrimidine core annulation via 1*H*-tetrazole ring cleavage was examined. The product structures, 2,3-diamino- or hydrazino derivatives of thieno[3,2-*d*]pyrimidin-4(3*H*)-ones, thieno[2,3-*d*]pyrimidin-4(3*H*)-ones and [1]benzofuro[3,2-*d*]pyrimidin-4(3*H*)-one, were elucidated based on NMR and single crystal X-ray analysis.

© 2018 Elsevier Ltd. All rights reserved.

Substituted 1*H*-tetrazoles are considered to be versatile synthetic building blocks, possessing —N=N— double bonds with a particular ability to eliminate nitrogen and thus generate reactive species.^{1–4} Previously, these compounds have been demonstrated to be suitable precursors for the construction of nitrogen-containing heterocycles, such as thienopyrimidines,^{4–6} which are well-known for their wide array of pharmacological activities especially in cancer and antiviral research.^{7,8}

Therefore, the reactivity investigation of functionalized thiophenes bearing a 1*H*-tetrazole ring is an interesting area of research for the development of convenient approaches towards novel thienopyrimidine derivatives.

Recently, we reported that alkyl 2-(1*H*-tetrazol-1-yl)-4-*R*¹-5-*R*²-thiophene-3-carboxylates, obtained from alkyl 2-amino-thiophene-3-carboxylates, undergo recyclization via cleavage of the tetrazole ring, elimination of nitrogen, and annulation of the pyrimidinone core (Fig. 1).^{5,6} This simple one-pot reaction accommodates many different basic amines and hydrazines and proceeds in high yield under mild conditions, with exclusive regioselectivity observed for a number of aliphatic amines.⁶

As a continuation of our work, the reactions of a number of hydrazines with 1*H*-tetrazoles were examined in order to better understand the nature of pyrimidine formation.

The one-pot solvent-free reaction of tetrazoles **2a–d**¹² with hydrazine hydrate **3a** for 5–15 min at reflux afforded 2,3-diamino-thieno[3,2-*d*]pyrimidin-4(3*H*)-ones **4a–c** and [1]benzofuro[3,2-*d*]pyrimidin-4(3*H*)-one **4d** in excellent yields (Table 1, entries 1–5).¹³ It should be noted that decreasing the reaction time allowed the desired fused pyrimidines to be obtained in higher yields in comparison to the previous study.⁵

Next, the reaction direction depending on the nature of the hydrazine was investigated. It is clear that the nitrogen atoms in monosubstituted hydrazines have different nucleophilicity and the formation of a mixture of products is possible.

Therefore, methylhydrazine **3b** and phenylhydrazine **3c** were also studied. Novel thieno[2,3-*d*]pyrimidin-4(3*H*)-ones **4f–h** were obtained in high yields via the reaction of 1*H*-tetrazoles **2f–h** with methylhydrazine **3b** (Table 1, entries 6–9).¹³

Conversely, treatment of 1*H*-tetrazoles **2f–h** with phenylhydrazine **3c** did not result in formation of the desired thienopyrimidine. The extended reaction time and higher temperatures led to decomposition of compounds.

In our previous work, the isomeric 2,3-diaminothieno[2,3-*d*]pyrimidin-4(3*H*)-one (Table 1, entry 5) was described,⁵ however, single crystal X-ray analysis was not performed. In this case,

* Corresponding author.

E-mail addresses: shyyka.olya@gmail.com, olga.shyyka@lnu.edu.ua (O.Ya. Shyyka).

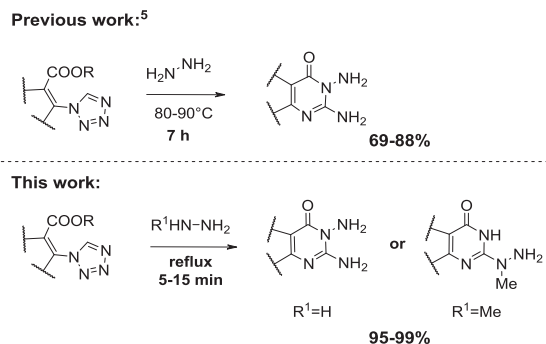


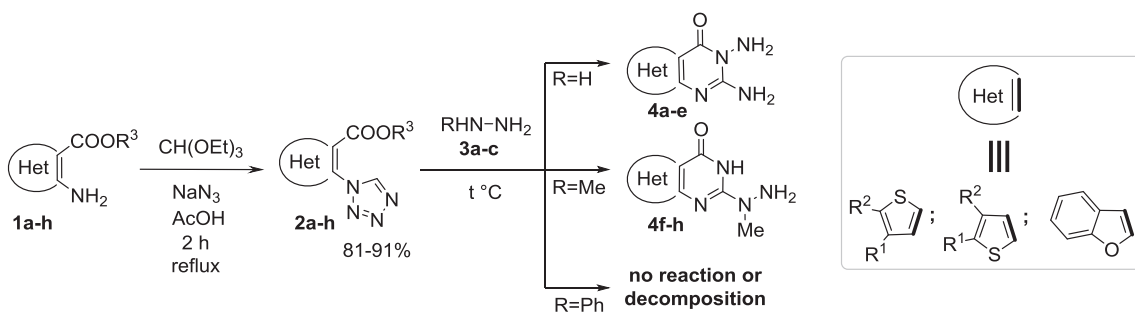
Fig. 1. Pyrimidine annulation via tetrazole ring cleavage with hydrazines.

X-ray structural determination of the previously described 2,3-diamino-5,6-dimethylthieno[2,3-d]pyrimidin-4(3H)-one **4e** (Fig. 2a) and 2-(1-methylhydrazino)-5,6,7,8-tetrahydro[1]benzothieno[2,3-d]pyrimidin-4(3H)-one **4g** (Fig. 2b) confirmed the assigned product structures (detailed information with Hirshfeld surface studies is provided in the ESI).

Previously a strong correlation between the basicity and nucleophilicity of the involved reactants was demonstrated.⁶ The target thienopyrimidines were easily prepared in excellent yield using basic nucleophilic amine reagents. In contrast, weakly basic amines ($pK_a < 6$) were unreactive in this protocol. Such a statement implicitly explains the reactivity in the described reaction of tetrazoles with hydrazines. Hydrazine hydrate and methylhydrazine are examples of good bases and good nucleophiles (pK_a 8.07 and

Table 1

Synthesis of thieno[3,2-d]pyrimidin-4(3H)-ones **4a–c**, thieno[2,3-d]pyrimidin-4(3H)-ones **4e–h** and [1]benzofuro[3,2-d]pyrimidin-4(3H)-one **4d** via the reaction of 1H-tetrazoles **2a–h** with hydrazines **3a–c**.



Entry	Tetrazole	Hydrazine	Time	Product	m.p. °C	Yield (%)
1	2a	3a H ₂ NNH ₂	15 min	4a	>300	99
2	2b	3a H ₂ NNH ₂	5 min	4b	>300	98
3	2c	3a H ₂ NNH ₂	7 min	4c	>300	95
4	2d	3a H ₂ NNH ₂	5 min	4d	287 (dec.)	98
5	2e	3a H ₂ NNH ₂	7 min	4e^a	285–286	97
6	2f	3b MeNHNH ₂	7 min	4f	238–239	98

(continued on next page)

Download English Version:

<https://daneshyari.com/en/article/7830082>

Download Persian Version:

<https://daneshyari.com/article/7830082>

[Daneshyari.com](https://daneshyari.com)