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Full Length Article

Structural and chemical characterization of CdSe-ZnS core-shell quantum dots

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ABSTRACT

The structural and compositional properties of CdSe-ZnS core-shell quantum dots (QDs) with a sub-nm shell thickness are analyzed at the atomic scale using electron microscopy. QDs with both wurtzite and zinc blende crystal structures, as well as intermixing of the two structures and stacking faults, are observed. High-angle annular dark-field scanning transmission electron microscopy suggests the presence of a lower atomic number epitaxial shell of irregular thickness around a CdSe core. The presence of a shell is confirmed using energy dispersive X-ray spectroscopy. Despite the thickness irregularities, the optical properties of the particles, such as photoluminescence and quantum yield, show clear enhancement after growth of the ZnS shell.

1. Introduction

Colloidal QDs[1] are of increasing interest for applications in photovoltaics [1,2], lighting [3] and sensing [4], as a result of their extraordinary properties, such as their tunable band gaps, as well as the reduced cost of their fabrication. For numerous applications, such QDs need to have small diameters, often between 2.5 and 5 nm [5]. The quality of the QD surfaces also influences their performance [4,5]. In particular, surface defects that include dangling bonds can create non-radiative sites that deteriorate optical properties such as QY[2] [6]. In order to overcome the detrimental effects of such defects, the encapsulation of QDs by shells has attracted attention, as it has been found to improve their performance with respect to bare colloidal QDs [2,4]. The formation of a shell has additional advantages, such as preventing the release of the non-environmentally-friendly heavy metals and providing stability against photo-oxidation. If the band gap of the shell is larger than that of the core, then the system is known as type I and electrons and holes are confined in the core. This is the case for CdSe-ZnS core-shell QDs, for which the shell has been reported to increase the QY and PL[3] intensity with respect to CdSe QDs [8]. As a result, CdSe-ZnS colloidal QDs are used extensively in the photovoltaic field [7,8]. During the synthesis of such QDs, parameters such as size, shape and high mono-dispersion must be controlled, as they have a strong influence on optical properties. For example, critical values of CdSe core size can lead to the escape of electrons from the core to the shell [9]. The core size can be controlled by the temperature and concentration of the precursors [10], which should be optimized. The structural properties of the shell, such as its thickness and crystallinity, also have a crucial impact on the optoelectronic properties of core-shell QDs [11]. It is therefore possible to obtain tunable emission from CdSe-ZnS QDs by controlling their shell thickness [8]. Vinayakan et al. reported an optimum ZnS shell thickness of 1.3–2.2 MLs[4], in order to avoid the formation of misfit dislocations [12]. Despite the importance of shell thickness, its influence is often only studied using indirect techniques such as PL and QY [11,10]. In order to achieve further progress, direct structural and compositional information about the core, the shell and the relationship between them are essential.

TEM[5] can be used to obtain direct local information about the

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[1] quantum dots
[2] quantum yield
[3] photoluminescence
[4] monolayers
[5] transmission electron microscopy

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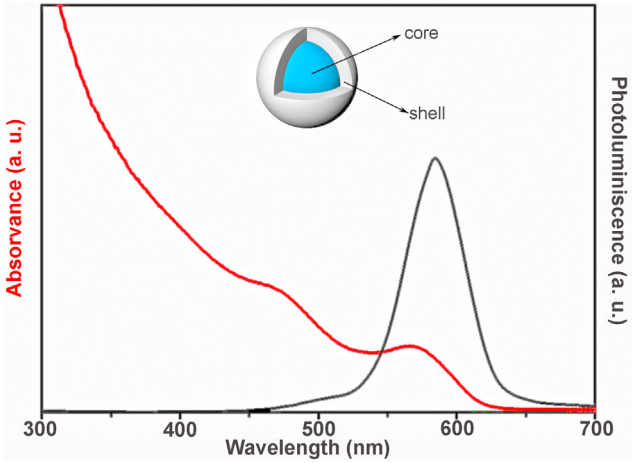

Fig. 1. Absorbance and PL spectra of CdSe-ZnS core-shell QDs in solution.

structural properties of materials. HRTEM[6] can be used to obtain atomic scale information about the crystalline quality of colloidal core-shell QDs, such as CdS-ZnS [12,13] or CdSe-ZnS-CdS [14] QDs. Interfaces in core-shell Pd-Pt nanoparticles [15] and anisotropic cation exchange evolution at interfaces in PbSe-CdS core-shell nanorods have been studied successfully using HAADF-STEM[7] [16]. The latter technique has proved to be useful for the study of Au QDs with Pd shells as thin as 2 nm [17]. Here, we use (S)TEM to study the structural properties of CdSe-ZnS colloidal core-shell QDs with sub-nm shell thicknesses. We find that the QDs adopt a mixture of zinc blende and wurtzite crystal structures and have dimensions that are consistent with optical measurements. The presence of an epitaxial shell of irregular thickness is identified using atomically resolved HAADF STEM and EDX[8].

2. Materials and methods

2.1. CdSe core synthesis

Highly luminescent CdSe QDs were prepared by a conventional synthesis route based on a hot injection method [18]. Briefly, a mixture of CdO (5 mmol) and oleic acid (30 mmol) in ODE[9] was heated in vacuum at 120 °C for 30 min to remove any residual moisture. The mixture was heated to 250 °C to ensure complete solution of the precursors under N_2. A transparent solution of Se (3.2 mmol) and TOP[10] (16 mmol) was injected swiftly at 250 °C and left to react for 5 min. The QDs were then purified by several successive precipitation and redispersion steps with a mixture of acetone and methanol and redispersed in octadecene for further processing.

2.2. CdSe-ZnS core-shell synthesis

Core-shell synthesis was carried out following the SILAR method, which is based on alternating injections of Zn and S precursors into a solution containing CdSe nanocrystals, in order to synthesize CdSe-ZnS core-shell nanocrystals. The Zn precursor solution (0.1 M) was prepared by dissolving ZnO (0.13 mmol) in oleic acid (0.9 mmol) and ODE (1 mL) at 310 °C. The S precursor (0.1 M) was prepared by dissolving sulfur (0.13 mmol) in ODE (1.34 mL) at 180 °C. Once clear solutions were obtained, the Cd and S solutions were allowed to cool to 120 °C and room temperature, respectively.

The amount of Zn and S precursor required for each layer was determined by the number of surface atoms of a given size for the core-shell nanocrystal. The average thickness of one monolayer of ZnS was taken to be 0.31 nm. One additional monolayer of growth therefore increases the shell thickness by up to 0.62 nm [5]. The typical synthesis of core-shell CdSe-ZnS nanocrystals consisted of mixing CdSe crystals (2.7 nm in diameter, 3.75×10^{-4} mmol) with 25 mL of ODE in a three-neck flask. The flask was pumped down at 100 °C for 30 min to remove any residual moisture. The reaction mixture was further heated to 240 °C under Ar for the injections. The first injection comprised 0.53 mL of the Zn precursor solution at 0.1 mL/min. After 30 min, 0.53 mL of the S precursor solution was injected slowly at 0.1 mL/min. After 30 min, the second monolayer of the shell was formed by adding 0.68 mL of the Zn precursor solution and, after 30 min, the same amount of the S solution was injected at the same rate. The reaction was cooled to room temperature and the product was purified by two successive precipitations and redispersion cycles using methanol and chloroform. Finally, the QDs were dispersed in chloroform at a concentration of 20 mg/mL.

2.3. Characterization

Optical absorbance spectra were recorded at room temperature using an ultraviolet-visible Perkin-Elmer Lambda 20 spectrophotometer. PL spectra were measured at room temperature upon excitation of the samples with a CW GaN laser (404 nm) or DSPP diode laser (533 nm). In both cases, the excitation power was fixed to be approximately 15 kW cm^2. The QY for CdSe QDs was determined using an integrated sphere (Hamamatsu model C9920-0). CdSe-ZnS specimens for the (S)TEM study were prepared by dropping the solution containing the QDs onto holey C grids. Then, a H_2/Ar plasma treatment was used for 50 s to clean the samples, in particular from residual organic compounds from the colloidal synthesis process. An FEI Titan Themis Cubed TEM and an FEI Titan G2 80-200 ChemiSTEM equipped with a Super-X EDX system were used for (S)TEM measurements. The accelerating voltages used were 200 kV for imaging and 80 kV for EDX.

3. Results and discussion

Fig. 1 shows absorbance and PL spectra recorded from CdSe-ZnS core-shell QDs in solution. The first excitonic absorption is located at 570 nm, while a second exciton is located close to 460 nm. The emission band is centered at 590 nm with a FWHM[11] of 60 nm. The QY improves from 15% to 50% after the formation of a ZnS shell of thickness 0.62 nm. The QY enhancement is attributed to the shell passivation of non-radiative surface trap sites, such as dangling bonds [7]. The measured optoelectronic properties suggest that the CdSe QDs are fully covered by a ZnS shell, which leads to an enhancement of PL and QY.

Fig. 2(a) shows an HRTEM image, in which where several CdSe-ZnS QDs are visible. One of the QDs has been marked by a circle. As expected, the QDs are crystalline. Their average diameter is measured to be 3.8 ± 0.2 nm, which is consistent with the nominal values (2.70 nm core and 0.62 nm shell). Fig. 2(b) shows a histogram of the measured QD size distribution. Most of the QDs have diameters of between 3.5 and 4.5 nm. Their sizes and optical properties are consistent with each other, according to empirical functions reported by Lu et al., which correlate size with excitonic peak position [19].

The crystal structures of the QDs were determined from the high-resolution micrographs. Fig. 2(c) and (d) show HRTEM images of two QDs and their Fourier transforms. Fig. 2(c) reveals the typical zigzag pattern of the wurtzite crystal structure viewed along [0 1 0], with a measured lattice spacing of 0.36 nm, in agreement with the expected crystalline structure. A different crystal structure is found in the QD

[6] high resolution TEM
[7] high angle annular dark field-scanning TEM
[8] energy dispersive X-ray
[9] octadecene
[10] trioctylphosphine
[11] Full Width at Half Maximum