

Full Length Article

Humidity sensing properties of the hydrothermally synthesized WS₂-modified SnO₂ hybrid nanocomposite

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ABSTRACT

WS₂-modified SnO₂ hybrid nanocomposite was synthesized by a hydrothermal method. The morphology and structure characterization revealed that the SnO₂ nanocrystals were anchored to the WS₂ nanosheets in the homogeneous nanocomposites. The resistive humidity sensors based as-prepared nanocomposites, pure WS₂ nanosheets and pure SnO₂ microspheres were fabricated and tested in a humidity range of 11–95% RH at 25 °C. The humidity sensing properties of the nanocomposites manifested a good and stable humidity sensing. The response of the hybrid nanocomposite in air with 95% RH was found to be 8.5 and 862.8 times higher than that of pure SnO₂ microspheres and WS₂ nanosheets, respectively. The synergistic effect between WS₂ and SnO₂ played an important role in improving the humidity sensing performance of the hybrid nanocomposites.

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1. Introduction

In order to meet the demand for humidity detection in industrial productions, food storage and environment monitoring, great efforts have been made to explore humidity-sensitive materials for the fabrication of high-performance humidity sensors. Until now, various functional materials have been utilized as humidity sensing materials, such as metal oxides [1–2], carbon nanomaterials [3] and hybrids composites [4–5]. SnO₂ is an important n-type semiconductor for humidity detection [1]. Since it was reported that the single-crystalline SnO₂ nanowires exhibited a fast and high response to RH changes at 30 °C [6], SnO₂-based nanostructures have been attracting more and more interests due to their high chemical stability, wide operating temperature range, non-toxicity and low-cost [7]. However, the sensitivity and response/recovery time of the humidity sensors based on SnO₂ nanostructures have impeded their practical applications.

Recent research has proved that decorating SnO₂ nanostructures with other materials is an effective route to enhance their humidity sensing properties. The sensitivity of the humidity sensor increased by 5.5 times after the SnO₂ nanostructures were decorated with Ag nanoparticles [8]. The Li⁺-doped SnO₂ porous nanofibers obtained an ultrafast response and achieved a 15 times

higher response than that of pristine SnO₂ porous nanofibers at 85% RH [9]. The humidity sensing properties of NiO-SnO₂ nanofibers were investigated [10]. In addition to metal and metal oxide, the humidity sensing properties of SnO₂ nanostructures decorating with two-dimensional (2D) materials (such as graphene and MoS₂) were also explored. Because of the high surface-to-volume ratio of 2D materials [11–12], the decorated SnO₂ nanostructures demonstrated excellent humidity sensing properties. It was found that the reduced graphene oxide-SnO₂ (rGO-SnO₂) nanocomposites have a better sensitivity factor and faster response and recovery times compared to SnO₂ zigzag belts [13] and nanowires [6]. The humidity response of MoS₂-SnO₂ hybrid reached up to 32,850 and increased by 1.5 times compared to that of pure SnO₂ [14]. As the same as MoS₂, WS₂ is an important 2D metal dichalcogenide, and has exhibited obvious responses to NO₂, NO, NH₃ at 25 °C [15–16], indicating that WS₂ has great potential for gas and humidity detection. Herein, WS₂-modified SnO₂ hybrid nanocomposite was prepared by a hydrothermal method, and its crystallinity, morphology and structure were investigated by X-ray diffraction (XRD), field emission scanning electron microscopy (FESEM), transmission electron microscopy (TEM) and energy dispersive X-ray spectrometry (EDS), respectively. The resistant humidity sensors based as-prepared nanocomposites, pure WS₂ nanosheets, and pure SnO₂ were fabricated and tested in a humidity range of 11–95% RH at 25 °C. The humidity sensing properties of the nanocomposites manifested a good and stability humidity sensing. The response of the nanocomposite was found to be

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8.5 and 862.8 times higher than that of SnO₂ microspheres and WS₂ nanosheets at 95% RH, respectively.

2. Experimental

2.1. Materials synthesis

The synthesis of the WS₂-modified SnO₂ hybrid nanocomposite consists of the prior preparation of WS₂ nanosheets and the following synthesis of WS₂-modified SnO₂. In the preparation process of WS₂ nanosheets, high purity sodium tungstate (Na₂WO₄·2H₂O, 99.9%, 1.5 mmol) and Cetyltrimethyl ammonium bromide (CTAB, 99.9%, 2.7 mmol) were dissolved in deionized water, and the pH of the solution was adjusted to the range of 2–3 by the addition of diluted hydrochloric acid (2 mol/L). The suspension was placed in a Teflon-lined stainless steel autoclave and the temperature was kept at 180 °C in a drying oven for 48 h. The resulting suspension was centrifuged and washed with ethanol for several times and then dried in a vacuum drying oven at 80 °C for 8 h. WS₂ nanosheets were obtained after the sulfurization of the primary product at 800 °C at the down stream of sulfur powder in a tube furnace. The obtained WS₂ nanosheets (40 mmol/L), SnCl₄·5H₂O (70 mmol/L) and NaOH (0.5 mol/L) were dissolved in deionized water and stirred for 0.5 h. Hydrothermal treatment of the above-mentioned solutions was performed at 180 °C for 48 h. Finally, WS₂-modified SnO₂ hybrid nanocomposites were obtained after the ordinary washing and drying process as mentioned above. What's more, in order to compare the humidity sensing properties between pure SnO₂ and WS₂-modified SnO₂ hybrid nanocomposites, SnO₂ microspheres were obtained with SnCl₄·5H₂O (70 mmol/L) and NaOH (0.5 mol/L), and the preparation conditions were same with that of WS₂-modified SnO₂ hybrid nanocomposites.

The as-prepared WS₂ nanosheets, SnO₂ microspheres and WS₂-modified SnO₂ hybrid nanocomposites were characterized by XRD with Cu K α radiation (PA National X' Pert Pro) and FESEM (JEOL, JSM-6700F). The SnO₂ microspheres and WS₂-modified SnO₂ hybrid nanocomposites were further characterized by TEM (JEOL, JEM-2100) and EDS.

2.2. Fabrication and testing of the resistive humidity sensors

Resistive humidity sensors were fabricated based on the commercial alumina ceramic substrates fixed with interdigital silver electrodes on the top surface. The as-prepared WS₂-modified SnO₂ hybrid nanocomposites were dispersed on the interdigital silver electrodes by brush coating. Each microelectrode in the interdigital silver electrodes was 0.2 mm in width and 5.6 mm in length, and the distance between adjacent microelectrodes were 1 mm. The working area of the humidity sensor was about 7 mm \times 7.4 mm in size. The humidity sensing properties of as-fabricated resistive humidity sensors were investigated by exposing the sensors to different RH levels at 25 °C. The saturated salt solutions of LiCl, MgCl₂, Mg(NO₃)₂, NaCl, KCl and KNO₃ were used to yield 11, 33, 54, 75, 85, 95% RH levels, respectively.

3. Results and discussion

3.1. Materials characterizations

The crystallographic structure and phase purity of the as-fabricated WS₂ nanosheets, WS₂-modified SnO₂ hybrid nanocomposites and SnO₂ microspheres were revealed by the XRD pattern in Fig. 1. All the diffraction peaks of the WS₂ nanosheets and SnO₂ microspheres were well ascribed to the hexagonal WS₂

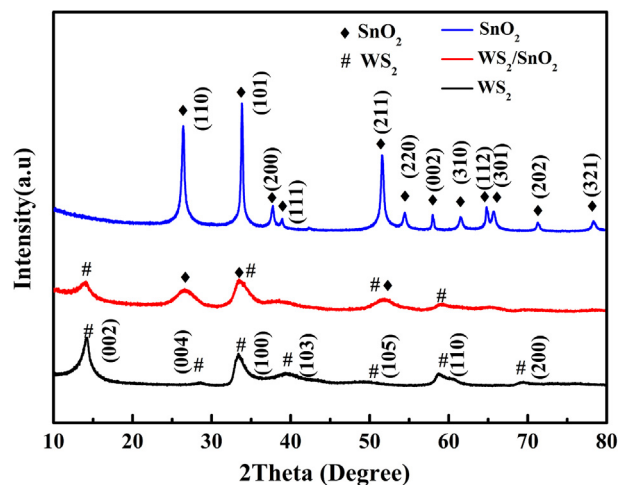


Fig. 1. XRD pattern of the as-prepared WS₂ nanosheets, WS₂-modified SnO₂ hybrid nanocomposites and SnO₂ microspheres.

(JCPDS-ICDD 84-1398) and tetragonal SnO₂ (JCPDS-ICDD 77-0451), respectively. The diffraction peaks of WS₂-modified SnO₂ hybrid nanocomposites at 27°, 14° and 59° were accurately indexed to the (1 1 0) plane of tetragonal SnO₂, (0 0 2) and (1 1 0) planes of hexagonal MoS₂, respectively. Since several diffraction peak positions of tetragonal SnO₂ and hexagonal MoS₂ are very close with each other, tetragonal SnO₂ and hexagonal MoS₂ both contributed to the diffraction peaks around 33° and 52°, resulting in the broadening FWHM (full width at half maximum) of the two diffraction peaks. All the diffraction peaks of the WS₂-modified SnO₂ hybrid nanocomposites corresponded to the presence of MoS₂ and SnO₂ crystals, illustrating the successful preparation of the nanocomposite.

The typical morphologies of as-prepared WS₂ nanosheets and WS₂-modified SnO₂ hybrid nanocomposites were demonstrated by SEM, as shown in Fig. 2a–b. As reported in the previous work [17], the WS₂ nanosheets were ultra-thin and possessed a three dimensional wall-like structures. In the WS₂-modified SnO₂ hybrid nanocomposites, there were nanocrystals uniformly attached to the surface of WS₂ nanosheets. The High resolution TEM image in Fig. 2c revealed that the SnO₂ nanocrystals were anchored to the WS₂ nanosheets. The lattice spacings of 0.26 nm and 0.34 nm corresponded to the (1 0 1) and (1 1 0) planes of tetragonal SnO₂, respectively. A layered structure with a lattice spacing of 0.62 nm was recognized as hexagonal WS₂. The selected area electron diffraction (SAED) pattern of the hybrid nanocomposites were shown in Fig. 2d, and the disperse rings in the SAED pattern were well indexed to the crystal planes of tetragonal SnO₂ or hexagonal WS₂. Because several lattice spacings of tetragonal SnO₂ and hexagonal WS₂ are approximately equal to each other, the corresponding disperse rings of them appeared to overlap, which is consistent with the XRD pattern in Fig. 1. The corresponding crystal planes of tetragonal SnO₂ and hexagonal WS₂ were both marked in Fig. 2d. The disperse rings in the SAED pattern were well indexed to the crystal planes of SnO₂ (1 1 0), WS₂ (1 0 0)/SnO₂ (1 0 1), WS₂ (1 0 3)/SnO₂ (1 1 1), WS₂ (1 0 5)/SnO₂ (2 1 1), WS₂ (1 0 7)/SnO₂ (3 1 0), SnO₂ (3 3 0), respectively. What's more, the HRTEM image in Figs. 2c and S1c revealed that the SnO₂ grain size in the hybrid nanocomposites was smaller than that in pure SnO₂. As a result, the broadening diffraction peak corresponding to (1 1 0) plane of tetragonal SnO₂ in the XRD pattern of the hybrid nanocomposites can be attributed to the grain refinement.

In order to explore the distribution of SnO₂ and WS₂ in the WS₂-modified SnO₂ hybrid nanocomposites, the scanning

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