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Behavior of water and methane bound to hydrophilic and hydrophobic nanosilicas and their mixture

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ABSTRACT

Methane adsorption onto hydrated (0.1 g/g) initial and compacted nanosilica A-50 and compacted A-50/hydrophobic AM1 (based on nanosilica A-300) and water behavior *vs*. temperature were analyzed using ¹H NMR and theoretical methods. Methane adsorption increases with temperature because of changes in structure of mobile water. Stronger compaction of A-50 than that of A-50/AM1 (due to negative effect of AM1 preventing formation of tight contacts between A-50 nanoparticles) leads to decrease in adsorption of methane onto dense A-50 alone. These effects cause the difference in temperature behavior of bound water.

Keywords: Hydrophilic nanosilica; Hydrophobic nanosilica; Structural organization; Bound water; Methane adsorption

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