



# Resonant energy transfer and near-infrared emission enhanced by tri-doped $\text{Sr}_2\text{SiO}_4$ : $\text{Ce}^{3+}$ , $\text{Tb}^{3+}$ , $\text{Yb}^{3+}$ phosphors for silicon solar cells

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## ABSTRACT

A novel tri-doped  $\text{Sr}_2\text{SiO}_4$ :  $\text{Ce}^{3+}$ ,  $\text{Tb}^{3+}$ ,  $\text{Yb}^{3+}$  phosphor was prepared using a conventional solid reaction. The resonant energy transfer (ET) with  $\text{Ce}^{3+} \rightarrow \text{Tb}^{3+}$  through electric dipole-dipole interactions was demonstrated with luminescent spectra and decay lifetime curves. Upon UV photon excitation at 372 nm of  $\text{Tb}^{3+}$  ions and 350 nm of  $\text{Ce}^{3+}$  ions,  $\text{Yb}^{3+}$  emissions were detected in the near-infrared (NIR) region with a main peak at 977 nm, which belonged to the  $^2\text{F}_{5/2} \rightarrow ^2\text{F}_{7/2}$  transition and corresponding perfectly with silicon solar cell maximum spectral responses. The NIR emission intensity of  $\text{Sr}_2\text{SiO}_4$ :  $\text{Ce}^{3+}$ ,  $\text{Tb}^{3+}$ ,  $\text{Yb}^{3+}$  was almost 125 times greater than that of  $\text{Sr}_2\text{SiO}_4$ :  $\text{Yb}^{3+}$ , which demonstrated that the 4f–5d luminescence of  $\text{Ce}^{3+}$  could be used to sensitize the  $\text{Tb}^{3+}$ – $\text{Yb}^{3+}$  down-conversion process. This tri-doped  $\text{Ce}^{3+}$ – $\text{Tb}^{3+}$ – $\text{Yb}^{3+}$  system offered potential applications for NIR down-conversion phosphors for silicon solar cells.

## 1. Introduction

Ultraviolet (UV) and visible (VIS) to near-infrared (NIR) down-conversion (DC) phosphors doped with rare earth (RE) ion crystals have been extensively studied for their potential applications in silicon solar cells (SSC) [1–3]. Among RE ions,  $\text{Yb}^{3+}$  has a simple energy level with a single excited level ( $\sim 1.2$  eV), roughly in line with the silicon band gap ( $\sim 1.1$  eV) [4,5]. Therefore, much research has focused on  $\text{RE}^{3+}$ – $\text{Yb}^{3+}$  co-doped materials (RE = Tb [6,7], Tm [8,9], Pr [10,11], and Er [12,13]) to adjust the solar spectrum to enhance the efficiency of SSC. However,  $\text{RE}^{3+}$  (RE = Tb, Tm, Pr, Er) have weak and line-like absorption in the VIS region by reason of the proscribed nature of f–f transitions, preventing the absorption of a large part of the broad band of distributed sunlight [14]. Because of the accepted 4f–5d transition,  $\text{Ce}^{3+}$  could be a perfect donor ion, having a wide and strong absorptive capacity in the desired spectral region [15].  $\text{Ce}^{3+}$  is an efficient sensitizer for absorbing the UV and VIS portions of the solar spectrum and its excitation energy efficiently transfers to the  $^5\text{D}_4$  state of  $\text{Tb}^{3+}$  for the  $\text{Tb}^{3+}$ – $\text{Yb}^{3+}$  NIR process in other hosts [16], such as  $\text{Ca}_2\text{BO}_3\text{Cl}$  [17], YAG [18],  $\text{KSrPO}_4$  [19], and borate glass [20]. Nevertheless, Duan et al. have discovered that the energy transfer (ET) direction between  $\text{Ce}^{3+}$  and  $\text{Tb}^{3+}$  is closely linked to the matrix in  $\text{Ce}^{3+}$ / $\text{Tb}^{3+}$  co-doped  $\text{Re}_2\text{Si}_4\text{N}_6\text{C}$  (Re = Lu, Y, and Gd) luminescent materials [21]. In  $\text{Lu}_2\text{Si}_4\text{N}_6\text{C}$  and  $\text{Y}_2\text{Si}_4\text{N}_6\text{C}$  matrix co-doped  $\text{Ce}^{3+}$ / $\text{Tb}^{3+}$ , the dominant direction of ET is  $\text{Ce}^{3+} \rightarrow \text{Tb}^{3+}$ . In contrast, in the  $\text{Gd}_2\text{Si}_4\text{N}_6\text{C}$  matrix, the ET dominant direction is exactly the opposite. Therefore, the ET mechanism in such a

system requires further analysis and evidence to explore for a more efficient wide absorption band NIR conversion system.

The presence of ET among  $\text{Ce}^{3+}$ – $\text{Tb}^{3+}$ – $\text{Yb}^{3+}$  prompted us to employ this process in our tri-doped silicate phosphors in view of the silicate's good mechanical and chemical stability. In this paper, the fabrication and performance of  $\text{Sr}_2\text{SiO}_4$ :  $\text{Ce}^{3+}$ ,  $\text{Tb}^{3+}$ ,  $\text{Yb}^{3+}$  phosphor, based on the resonant ET, were reported. This material cropped UV photons and displayed an extremely increased NIR emission of  $\text{Yb}^{3+}$  ion, located at  $\sim 1000$  nm and at  $\sim 125$ -fold that of  $\text{Sr}_2\text{SiO}_4$ :  $\text{Yb}^{3+}$ , which demonstrated that the 4f–5d luminescence of  $\text{Ce}^{3+}$  could be used to sensitize the  $\text{Tb}^{3+}$ – $\text{Yb}^{3+}$  down-conversion process. This tri-doped  $\text{Ce}^{3+}$ – $\text{Tb}^{3+}$ – $\text{Yb}^{3+}$  system will be utilized as a platform for NIR down-conversion phosphor for silicon solar cells.

## 2. Experimental

Reagent grade  $\text{SrCO}_3$ ,  $\text{SiO}_2$ ,  $\text{CeO}_2$ ,  $\text{Tb}_4\text{O}_7$ , and  $\text{Yb}_2\text{O}_3$  were employed as raw materials and were weighed stoichiometrically and mixed thoroughly with an agate mortar and pestle, with 5 wt%  $\text{B}_2\text{O}_3$  (A.R.) was added as flux. The mixture was subsequently settled into an alumina crucible and calcined at 1100 °C for 3 h in a reducing atmosphere. Then, the samples were cooled passively to room temperature and then reheated again.

The resulting products' compositions were analyzed using an X-ray diffractometer (XRD; Neo X D/Max-2500VB2, Rigaku Corp., Tokyo, Japan) with Cu-K $\alpha$  radiation. The emission or photoluminescent (PL)

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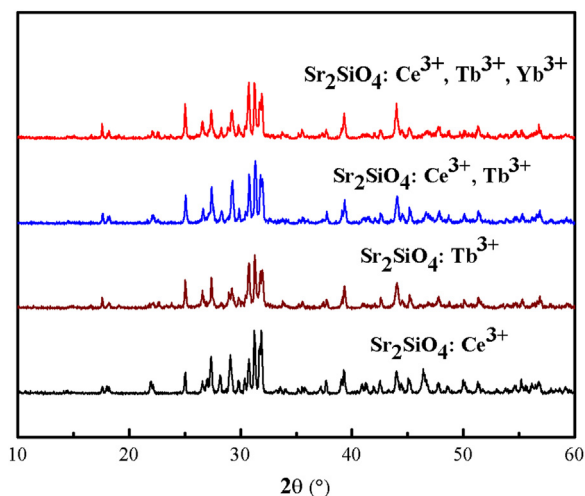


Fig. 1. XRD patterns of specimens doped with different ions.

spectra and photoluminescence excitation (PLE) were obtained using an F97Pro fluorescence spectrophotometer (Shanghai Lengguang Technology Co., Ltd., Shanghai, China) with a 150 W Xe lamp. The NIR excitation and emission spectra were registered using a FLS 980 steady/transient fluorescence spectrometer (Edinburgh Instruments Ltd., EH54 7DQ, UK) with a 150 W Xe lamp. All tests were at room temperature.

### 3. Results and discussion

#### 3.1. Luminescence and ET in $\text{Ce}^{3+}$ – $\text{Tb}^{3+}$ co-doped $\text{Sr}_2\text{SiO}_4$

The synthesized samples were consistent with standard  $\text{Sr}_2\text{SiO}_4$ , as confirmed by XRD (Fig. 1). The emission and excitation spectra of  $\text{Tb}^{3+}$  or  $\text{Ce}^{3+}$  solely doped and  $\text{Ce}^{3+}$ – $\text{Tb}^{3+}$  co-doped  $\text{Sr}_2\text{SiO}_4$  phosphors were measured (Fig. 2). The emission and excitation spectra of  $\text{Sr}_2\text{SiO}_4: 0.04\text{Ce}^{3+}$  was at 414 nm and the excitation spectrum involved a wider band, ranging from 230 to 400 nm and originating from the  $4f \rightarrow 5d$  transition for  $\text{Ce}^{3+}$  (Fig. 2a). In addition, the emission spectrum was composed of an asymmetric broadband. Under near UV excitation of 357 nm, the peak located at 321 nm was attributed to  $4f^0 5d^1 \rightarrow 4f^1 5d^0$  transitions of  $\text{Ce}^{3+}$ . The influence of  $\text{Ce}^{3+}$  concentration in the  $\text{Sr}_2\text{SiO}_4$  phosphor on emission intensity was investigated and it was clear that, as the  $\text{Ce}^{3+}$  concentration increased, the emission intensity was found to increase and then decrease (Fig. 2d). The maximum emission intensity obtained when  $\text{Ce}^{3+}$  concentration was 0.04.

The emission and excitation spectra of  $\text{Sr}_2\text{SiO}_4: \text{Tb}^{3+}$  showed that the  $\text{Sr}_2\text{SiO}_4$  excitation spectrum displayed a number of bands ranging from 300 and 400 nm in monitoring at 545 nm for  $\text{Tb}^{3+}$ , which was attributed to  $f \rightarrow f$  transitions of  $\text{Tb}^{3+}$  within the  $4f^8$  configuration (Fig. 2b). At 372 nm excitation, the emission peaks included 489, 545, 587, and 622 nm, which contributed to the  $^5\text{D}_4 \rightarrow ^7\text{F}_J$  ( $J = 6, 5, 4, 3$ ) transitions of  $\text{Tb}^{3+}$ . The optimal  $\text{Tb}^{3+}$  concentration was also found to be 0.16 (Fig. 2e).

The PL and PLE of  $\text{Sr}_2\text{SiO}_4$  phosphor co-doped with  $\text{Ce}^{3+}$  and  $\text{Tb}^{3+}$  were investigated and the experimental results shown in Fig. 2c. Emissions by  $\text{Ce}^{3+}$  (414 nm,  $5d \rightarrow 4f$ ) and  $\text{Tb}^{3+}$  (480–650 nm,  $^5\text{D}_4 \rightarrow ^7\text{F}_J$ ,  $J = 6, 5, 4, 3$ ) were both produced with excitation at 350 nm. The peak intensity of  $\text{Sr}_2\text{SiO}_4: 0.04\text{Ce}^{3+}, 0.06\text{Tb}^{3+}$  at 414 nm was significantly lower than that of  $\text{Sr}_{1.96}\text{SiO}_4: 0.04\text{Ce}^{3+}$  (Fig. 2a). Meanwhile, the emission peak intensity located at 545 nm was clearly higher than that of  $\text{Sr}_{1.94}\text{SiO}_4: 0.06\text{Tb}^{3+}$ , resulting from the existence of ET between  $\text{Ce}^{3+}$  and  $\text{Tb}^{3+}$  (Fig. 2c). In accordance with Dexter sensitization theory [22],  $\text{Ce}^{3+}$  was considered as an ideal sensitizer for  $\text{Tb}^{3+}$  when the emission waveband of  $\text{Ce}^{3+}$  was superimposed with the excitation waveband of  $\text{Tb}^{3+}$ . Comparison of Fig. 2a and b, with a dotted line,

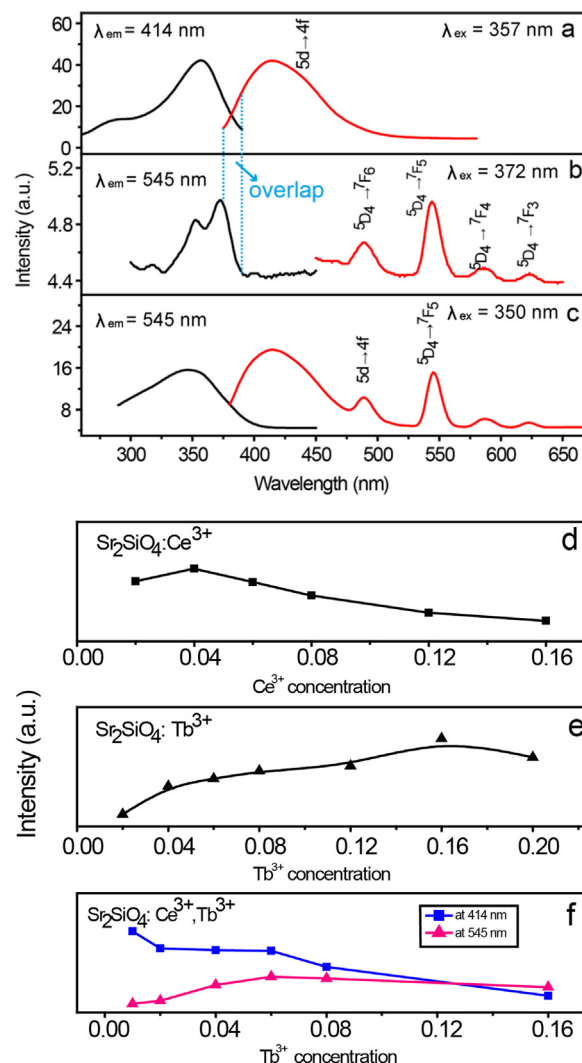


Fig. 2. Excitation and emission spectra of phosphors in left: (a)  $\text{Sr}_2\text{SiO}_4: 0.04\text{Ce}^{3+}$ , (b)  $\text{Sr}_2\text{SiO}_4: 0.06\text{Tb}^{3+}$ , and (c)  $\text{Sr}_2\text{SiO}_4: 0.04\text{Ce}^{3+}, 0.06\text{Tb}^{3+}$ . The relative PL intensities in right: (d)  $\text{Sr}_2\text{SiO}_4: x\text{Ce}^{3+}$ , (e)  $\text{Sr}_2\text{SiO}_4: y\text{Tb}^{3+}$ , and (f)  $\text{Sr}_2\text{SiO}_4: 0.04\text{Ce}^{3+}, y\text{Tb}^{3+}$ .

made clear the significant superimposition over 350 and 450 nm, which was concluded to be the effective ET for  $\text{Ce}^{3+} \rightarrow \text{Tb}^{3+}$  that occurred on the  $\text{Sr}_2\text{SiO}_4$  host, as expected.

Based on the results of Fig. 2d and e, the  $\text{Ce}^{3+}$  content was fixed at 0.04 and the emission intensity observed, with changes in Tb concentration in co-doping, so as to obtain the quenching concentration. Conversely, a fixed  $\text{Tb}^{3+}$  content of 0.16 was used to investigate the effects of Ce concentration changes. The emission intensity of  $\text{Ce}^{3+}$  was found to decrease monotonically with increasing of  $\text{Tb}^{3+}$  emission, while the  $\text{Tb}^{3+}$  quenching concentration was determined with 0.06 (Fig. 2f). The optimal component ratio of co-doped phosphor was thus  $\text{Sr}_2\text{SiO}_4: 0.04\text{Ce}^{3+}, 0.06\text{Tb}^{3+}$ , which was lower than that of solely-doped  $\text{Sr}_2\text{SiO}_4: \text{Tb}^{3+}$ .

To analyze the ET process of  $\text{Ce}^{3+} \rightarrow \text{Tb}^{3+}$ , a resonant ET mechanism with two main aspects was proposed that involved multipolar and exchange interactions [23]. The amount of space  $R_c$  between the sensitizer and activator was used as the judgment criteria and  $R_c$  must be  $< 5 \text{ \AA}$  in the ET mechanism for the exchange interaction, which was calculated as [24]

$$R_c \approx 2 \left( \frac{3V}{4\pi x_c Z} \right)^{1/3} \quad (1)$$

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