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# Piperine derivatives as green corrosion inhibitors on iron surface; DFT, Monte Carlo dynamics study and complexation modes

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## ABSTRACT

In the present paper, the interaction of three Piperine derivatives namely: 5-(1,3-benzodioxol-5-yl)-1-(piperidin-1-yl) pent-2-en-1-one ( $S_1$ ), 5-(1,3-benzodioxol-5-yl)-1-(piperidin-1-yl) penta-2,4-dien-1-one ( $S_2$ ) and 5-(1,3-benzodioxol-5-yl) penta-2,4-dienoic acid ( $S_3$ ) with iron (Fe) have been investigated theoretically by means of DFT method at the B<sub>3</sub>LYP/LANL2DZ high basis set. The interaction with iron surface therefore all possible Fe-inhibitor interaction modes at —O— and >N— active sites. The preferred complexes are those in which the Fe-atoms is in mono-dentate mode to the studied inhibitors. The quantum chemical properties most relevant to their potential action as corrosion inhibitors have also been calculated in both isolated and Fe-inhibitor ( $S_i$ ) complex form for comparison. The values of the calculated interaction energy are consistent with the experimentally observed highest corrosion inhibition efficiency trend for the three  $S_1FeN_{26}$ ,  $S_2FeN_{24}$  &  $S_3FeO_{26}$  complexes. The chemical reactivity and site selectivity of the molecule has been determined with the help of global and local reactivity descriptors of Fe-inhibitor ( $S_i$ ) complexes. The calculations show that

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