Accepted Manuscript

Nicotinium methane sulfonate (NMS): A bio-renewable protic ionic liquid and bi-functional catalyst for synthesis of 2-amino-3-cyano pyridines



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PII: S0167-7322(17)33325-1

DOI: doi:10.1016/j.molliq.2017.10.153

Reference: MOLLIQ 8146

To appear in: Journal of Molecular Liquids

Received date: 24 July 2017
Revised date: 2 October 2017
Accepted date: 28 October 2017

Please cite this article as: Fatemeh Tamaddon, Davood Azadi , Nicotinium methane sulfonate (NMS): A bio-renewable protic ionic liquid and bi-functional catalyst for synthesis of 2-amino-3-cyano pyridines. The address for the corresponding author was captured as affiliation for all authors. Please check if appropriate. Molliq(2017), doi:10.1016/j.molliq.2017.10.153

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Nicotinium methane sulfonate (NMS): a bio-renewable protic ionic liquid

and bi-functional catalyst for synthesis of 2-amino-3-cyano pyridines

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Abstract

In this work, a simple approach was used to quantitative preparation of nicotinum methane

(1-methyl-2-(pyridin-3-yl)pyrrolidin-1-ium methanesulfonate sulfonate

nicotine and methane sulfonic acid. NMS, as a novel, mild, efficient, economic, and task-

specific ionic liquid (TSIL) with dual acid and base functional groups, has been characterized

by NMR, FT-IR, acidity measurements, and elemental analysis. This protic ionic liquid

shows excellent catalytic activity in one-pot synthesis of 2-amino-3-cyanopyridines in 78-

98% from malononitrile, aromatic aldehydes, methyl ketones, and ammonium acetate under

solvent-free conditions. NMS is a nature-based recyclable and reusable catalyst.

Keywords: Nicotine, Protic ionic liquid, Methansulfonate, Aminocyanopyridine.

1. Introduction

Multicomponent reactions (MCRs) are key tools to access to complex molecules that used for

multi-disciplinary scientific purposes [1-5]. Amino-cyano pyridines (AmCyPs), with a

variety of biological activities in addition to their reaction intermediate potential for synthesis

of other important materials, are among the functionalized molecules prepared by MCRs [6-

8]. Four-component reaction of acetophenone or 1,3-dicarbonyls with aldehydes,

malononitrile, and ammonium acetate using various catalysts is the most developed route to

synthesis of AmCyPs [9].

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