



Line intensities of the 30011e – 00001e band of $^{12}\text{C}^{16}\text{O}_2$ by laser-locked cavity ring-down spectroscopy

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ABSTRACT

Thirty well isolated ro-vibrational transitions of the 30011e – 00001e band of $^{12}\text{C}^{16}\text{O}_2$ at $1.54\ \mu\text{m}$ have been recorded with a laser-locked cavity ring-down spectrometer. The line intensities were obtained with accuracies better than 0.85%. Comparisons of the line intensities determined in this work with literature experimental values and those from HITRAN2016, AMES, UCL-IAO and CDSD-296 line lists are given.

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1. Introduction

As the second most abundant greenhouse gas in Earth's atmosphere, carbon dioxide currently has a global average concentration of 404 parts per million by volume [1]. The terrestrial atmospheric concentration of carbon dioxide has increased almost 43 percent since pre-industrial times due to human activities. Its significant role in climate change impels several agencies to launch space-based observations of carbon dioxide, for example OCO-2 (USA) [2], GOSAT (Japan) [3], ASCENDS (USA) [4], Tan-Sat (China) [5], to monitor CO_2 global levels and their variations. These projects aim to retrieve the atmospheric column-averaged CO_2 dry air mole fraction (X_{CO_2}) with precisions in the range of 0.25–1%. These remote sensing activities heavily depend on the molecular absorption model used within the retrievals, therefore the accuracy of the line parameters and line profiles. To fulfill the goals of the current missions, the accuracy of CO_2 line intensities are required to reach 0.3–0.5% [6].

Line intensities are much more difficult to be determined accurately than line positions. Along with the development of high-precision laser spectroscopy techniques and the significant progress of line profile models, the accuracy of CO_2 line intensities determined at the laboratory has fulfilled the requirements for

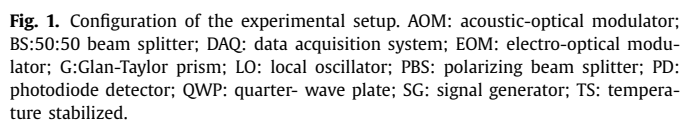
modern atmospheric remote sensing experiments [7–13]. However, only several studies [10] accessed to obtain a few lines with a precision of 0.3–1%. In the meantime, attempts [12,14–20] have been made to provide high-accuracy theoretical line intensities for CO_2 . For instance, Huang et al. [14–17] has performed a series of *ab initio* studies, obtained the most accurate potential energy surface (PES) of CO_2 , and generated the infrared line lists for 13 CO_2 isotopologues denoted as AMES [16,17]. Tennyson's group from University College London have shown that it is possible to derive computed line intensities using *ab initio* calculated dipole moment surface (DMS) with an accuracy comparable to available measurements for tri-atomic molecules such as the water molecule, hydrogen ion and carbon dioxide [12,21–24]. The agreement at the 0.3% level between *ab initio* computations and high-accuracy experiments [12] for $^{12}\text{C}^{16}\text{O}_2$ indicates the possibility of the atmospheric remote sensing research using *ab initio* calculations. Another set of calculated CO_2 line lists, named as UCL-IAO list, for 13 isotopologues of carbon dioxide has been produced by Zak et al. [19,25,26] with the combination of the high accuracy *ab initio* DMS [12] and the methodology used in Ref. [21]. Meanwhile, an empirical carbon dioxide spectroscopic databank (CDSD-296) [18,19] in the spectral range of 6–14,075 cm^{-1} has been created by Tashkun et al. within the framework of effective operators and based on the global weighted fit of spectroscopic parameters to the observed data from more than 200 measured spectra with uncertainties varying between ~0.1% [8] and over 100% [27]. It leads to a high percentage of line intensities in CDSD databank with the stated un-

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In the present work, the line intensities of thirty transitions with no interference from other absorption lines have been recorded by a laser-locked cavity ring-down spectrometer with high precision as well as high sensitivity. This spectrometer is similar to the experimental set-ups developed in Refs. [31,32]. High-precision line intensities are retrieved from the recorded spectra with a relative uncertainty of 0.7% on average, which was considerably improved over our previous studies on CO₂ and N₂O around 780 nm [33,34]. Precise intensities of the lines in the 30011e – 00001e “sensitive” band will be useful for the evaluation of the HITRAN2016, CDSD-296, AMES and UCL-IAO database.

The diagram of the experimental setup is presented in Fig. 1, which consists of frequency locking and spectral probing. A tunable external-cavity diode laser (ECDL, Toptica DL Pro-1550) is split into two beams by a polarizing beam splitting cube. The s-polarization beam is locked to a temperature-stabilized ring-down (RD) cavity using the Pound–Drever–Hall (PDH) method. The RD cavity is composed of two high-reflectivity mirrors with a distance of 110.8 cm. Each mirror has a reflectivity of 99.996% at 1.5–1.7 μm (Layertec GmbH). Therefore, the RD cavity has a finesse of about 7.8×10^4 and a mode width of about 1.6 kHz. The RD cavity is made of Invar, located in a stainless-steel vacuum chamber. A feedback circuit controls the heating-current in a wire surrounding the stainless-steel chamber to stabilize the Invar cavity temperature at about 298.8 K which was measured with two calibrated platinum thermal sensors attached at two sides of the RD cavity. We used a frequency counter (Agilent 53181A) to monitor the beat frequency between the laser and a frequency comb, and the results are shown in Fig. 2(a). The comb is synthesized by an Er: fiber oscillator operated at 1.56 μm . The repetition frequency and carrier offset frequency of the comb are locked to precise radio-frequency sources referenced to a GPS-disciplined rubidium clock (SRS FS725). The beat signal has a long-term drift of about 1 MHz per h, which is consistent with the thermal expansion of the cavity under a temperature drift of 50 mK. The temperature uncertainty of the RD cavity was estimated to be 0.2 K at maximum taking into ac-



Series	1 (Pa)	2 (Pa)	3 (Pa)	4 (Pa)	Relative uncertainty
a	5.00	6.22	8.47	9.97	0.7%
b	22.88	31.51	37.49	59.01	0.3%
c	100.18	120.64	676.20	800.60	0.25%

Natural carbon dioxide sample gas with a stated purity of 99.995% was bought from the Nanjing Special Gas Co. and further purified by a “freeze-pump-thaw” process before use. Three pressure series, listed in [Table 1](#), were adopted to reduce the statisti-

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