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Understanding the oxidation behavior of a ZrB2-MoSi2 composite at ultra-high temperatures

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ABSTRACT

This basic research investigates the microstructure evolution of a composite based on ZrB₂-MoSi₂ from the

as-sintered features to the changes occurring upon oxidation at ultra-high temperatures, 1650 and 1800°C,

in a bottom-up loading furnace for 15 minutes. Scanning and transmission electron microscopy evidenced

the formation of a matrix typified by ZrB₂-cores surrounded by (Zr,Mo)B₂-rims with dispersed MoSi₂

particles and SiO₂ glass trapped at the triple junctions. The oxidation at 1650°C induced the migration of

silica to the surface, which formed a continuous and protective scale. Below this scale, the matrix evolved

into ZrO₂ grains encasing MoB nano-inclusions, as a result of the oxygen and boron oxide partial pressures

established in the subscales. Underneath, a MoSi₂-depleted boride region, but substituted by SiO₂ and MoB

was found. The same phases were observed upon oxidation at 1800°C, but a thicker and more turbulent

oxidized layer formed as a consequence of the rapid evolution of MoO₃, SiO and B₂O₃ gases from the scales

beneath the outermost silica-layer.

According to the observed phases and the calculated phase stability diagrams, the partial pressures

gradient within the oxide layer were defined and the effect of Mo-doping in boride matrices on the

oxidation behavior was compared to that of other transition metals to establish a criterion design for the

realization of ceramics with improved oxidation resistance.

Keywords: UHTC; Oxidation; TEM; Microstructure; Inclusion.

1. Introduction

The great interest in borides and carbides of transition metals, a class of ceramics known as Ultra-

High Temperature Ceramics (UHTCs), is motivated by the search for materials that can withstand extreme

environments in terms of temperature, chemical reactivity, mechanical stress, radiation and ablation,

especially in hypersonic and space aviation [1]. ZrB2 can be considered a leading material in this field of

research, due to its unique combination of properties in terms of high melting point above 3000°C,

relatively low density, high thermal conductivity and good strength and refractoriness at elevated

temperatures [1-5]. However, pure ZrB₂ is extremely difficult to sinter, due to its strong covalent bonds and

low self-diffusion rates [6] and therefore it requires pressure-assisted sintering techniques at very high

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