

Effects of $\text{Sr}^{2+}/\text{Zn}^{2+}$ co-substitution on crystal structure and properties of nano- β -tricalcium phosphate

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Abstract:

The incorporation of therapeutic ions like Sr^{2+} , Si^{4+} , Zn^{2+} and Li^{+} into biomaterials has become a promising approach to promote bone regeneration. However, the effects of Sr^{2+} and Zn^{2+} co-substitution on the crystal structure and properties of β -tricalcium phosphate (β -TCP) have not been elucidated well. In this study, $\text{Sr}^{2+}/\text{Zn}^{2+}$ co-substituted β -tricalcium phosphate (SrZnTCP) nano-powders with different extents of substitution (0-4.8 mol%) were synthesized by poly(ethylene glycol)-assisted co-precipitation and subsequent heat treatment. The as-synthesized SrZnTCP nano-powders were characterized by x-ray diffraction, Fourier transform infrared spectroscopy, elemental analysis, Rietveld refinement and differential scanning calorimetry. The results showed that the conversion of calcium-deficient apatite to β -TCP was achieved after heat-treatment above 800 °C. The a -axis and c -axis lattice parameters gradually decreased with increasing level of $\text{Sr}^{2+}/\text{Zn}^{2+}$ co-substitution in β -TCP lattice. Sr^{2+} and Zn^{2+} preferentially occupied the ninefold coordinated Ca (4) sites and the sixfold coordinated Ca (5) sites, respectively. The co-substitution of $\text{Sr}^{2+}/\text{Zn}^{2+}$ for Ca^{2+} significantly improved the thermal stability of β -TCP. The release rate of Zn^{2+} from SrZnTCP depended on Ca^{2+} concentration over 63-day immersion in PBS solution while that of Sr^{2+} was not affected by Ca^{2+} concentration. The amount of Sr^{2+} released increased with increasing Sr^{2+} content in SrZnTCP. Collectively, SrZnTCP showed great promise as a $\text{Sr}^{2+}/\text{Zn}^{2+}$ -releasing biomaterial for bone repair, although no obvious mineralization was observed on β -TCP and SrZnTCP disc samples during 56 days of immersion in simulated body fluid.

Keywords: β -tricalcium phosphate; Therapeutic metallic ions; Rietveld refinement; Thermal stability; Release behavior

1. Introduction

Bone tissue engineering aims to repair bone defects and involves a combination of osteoconductive scaffold, osteoinductive growth factors and osteogenic cells [1]. The seeding of osteoconductive scaffold with stem cells often offers little or no benefit in increasing both regenerated bone volume within the scaffold and rate of callus formation [2]. To overcome this problem, various growth factors like bone morphogenetic protein-2 have been used to endow scaffolds with osteoinductivity to enhance healing of bone defects. However, these growth factors may rapidly lose bioactivity due to their short biological half-life and instability, which leads to limited therapeutic efficacy in bone regeneration [3]. Genetic modification of scaffolds is considered a promising strategy to improve bone regeneration over the past decade, but its extensive application is restrained by either potential

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