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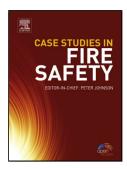
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## ACCEPTED MANUSCRIPT

Mechanisms of hot corrosion of pure nickel at 700°C: influence of testing conditions

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Highlights

• Mechanisms of hot corrosion governed by both *pO*2 and *pSO*3.

• Inward transport of SO2/SO3 demonstrated to rule the oxidation-sulphidation of pure nickel.

• The Na2SO4 deposit decreases the *pO*2 and the *pSO*3 at the scale surface.

• A sufficient *pSO*3 is required to stabilize Na2SO4-NiSO4 liquid solutions and simulate Type-II

hot corrosion conditions.

• The role of capillarity effects is elucidated.

**ABSTRACT** 

The hot corrosion behaviour of pure nickel was investigated at 700°C to highlight the influence of testing conditions (composition of the atmosphere and presence or not of a Na<sub>2</sub>SO<sub>4</sub> deposit) on the mechanisms of hot corrosion. The influence of SO<sub>2</sub>/SO<sub>3</sub> inward transport through the porous NiO growing scales was studied by exposing nickel in synthetic air and in synthetic air + 0.5%  $SO_2/SO_3$  without Na<sub>2</sub>SO<sub>4</sub>. The addition of Na<sub>2</sub>SO<sub>4</sub> decreased the  $p_{O_2}$  and the  $p_{SO_3}$  at the sample surface, which promoted the formation of Ni-S liquid solution at the metal/scale interface. A sufficient  $p_{SO_3}$  was required to stabilize liquid Na<sub>2</sub>SO<sub>4</sub>-NiSO<sub>4</sub> solutions and simulate Type-II hot

corrosion conditions.

**Keywords:** Nickel, Oxidation, Sulphidation, Hot corrosion, Sodium sulphate

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